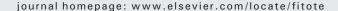
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### **Fitoterapia**





## Isolation, stability and bioactivity of Jatropha curcas phorbol esters

Joy S. Roach, Rakshit K. Devappa, Harinder P.S. Makkar\*, Klaus Becker

Institute for Animal Production in the Tropics and Subtropics (480b), University of Hohenheim, Stuttgart-70599, Germany

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#### ABSTRACT

Jatropha curcas seed oil, which can be utilized for biodiesel production upon transesterification, is also rich in phorbol esters (PEs). In this study, PEs from J. curcas oil (Jatropha factors  $C_1$  and  $C_2$  (purified to homogeneity), Jatropha factors  $C_3$  and  $(C_4 + C_5)$  (obtained as mixtures) and PE-rich extract (containing all the above stated Jatropha factors) were investigated. The concentrations of Jatropha PEs were expressed equivalent to Jatropha factor  $C_1$ . In the snail (Physa fontinalis) bioassay, the order of potency (EC50, µg/L) was: PE-rich extract < factor  $C_3$  mixture < factor  $C_2$  < factor  $C_4$  + C5). In the Artemia salina bioassay, the order of potency (EC50, µg/L) was: factor  $C_2$  < factor  $C_3$  mixture < factor  $C_4$  < factor  $C_4$  + C5) mixture. In addition, Jatropha PEs exhibited platelet aggregation (ED50, µM, factor  $C_2$  < factor  $C_3$  mixture < factor  $C_4$  + C5) mixture. The stability of a PE-rich extract was evaluated and found to be low at room temperature but favourable in ethanol over a range of temperatures. By integrating the isolation methodology developed in this study in the Jatropha biodiesel industry, PEs could be obtained as value-added co-products.

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#### 1. Introduction

Jatropha curcas L. is a promising feedstock of oil for biodiesel production. J. curcas is one of approximately 175 species in its genus and a member of the Euphorbiaceae family [1]. It is a perennial shrub or small tree native to South and Central America but cultivated in many tropical regions, including Africa and Asia. It thrives in poor, stony soils and under adverse climatic conditions. J. curcas has a variety of uses. In many parts of the world it is used as a live fence and for erosion control. The seed kernel of the plant contains up to 60% oil [2], which can be transesterified to biodiesel. The oil is also incorporated in cosmetics and soap production. The seed kernel meal remaining after oil extraction is rich in nutrients and is used as an organic fertilizer [3,4]. Although high in protein the use of the defatted kernel meal as livestock feed is precluded by the presence of phorbol esters (PEs), which are the major toxic principles in *I. curcas* seeds [3].

E-mail address: Harinder.Makkar@fao.org (H.P.S. Makkar).

Jatropha oil cannot be used for edible purposes without detoxification, making it attractive for biodiesel production. During mechanical extraction, the majority of PEs (~70%) present in the seed is extracted with the oil fraction [5]. During biodiesel production, the oil is subjected to many treatments (stripping, degumming and esterification), which leads to partial or complete destruction of the PEs [5]. Instead of losing the PEs, if a suitable method can be adopted to extract and isolate these esters before biodiesel production, the PEs could be valued co-products, which would contribute to enhancing economic viability and sustainability of the Jatropha oil-based biodiesel production chain. This is subject to the condition that the extraction of PEs from the oil does not adversely affect the quality of biodiesel produced.

Haas et al. [6] identified six J. curcas PEs, namely Jatropha factors  $C_1$ – $C_6$  (Fig. 1). The reason for choosing this group of compounds (PEs) for this study was that they are highly bioactive both  $in\ vitro$  and  $in\ vivo$ , but they are currently considered to be merely toxic, unwanted biomaterial in the Jatropha biodiesel production chain [7]. The recent increase in the cultivation of Jatropha means that there are potentially huge quantities of PEs that could be used for various applications. In this paper, we provide an optimized method for obtaining a J. curcas PE-rich extract and purified Jatropha PEs. In

<sup>\*</sup> Corresponding author at: Institute for Animal Production in the Tropics and Subtropics (480b), University of Hohenheim, Stuttgart-70599, Germany. Tel.: +49 711 459 23640; fax: +49 711 459 23702.

Fig. 1. Phorbol esters present in Jatropha curcas oil (Haas et al. [6]).

addition, the storage stability of the PE-rich extract and the biological activity of purified PEs are also presented.

#### 2. Materials and methods

#### 2.1. Processing of J. curcas oil

*J. curcas* seeds (Jaipur, India; Jaipur; geographical coordinates:  $26^{\circ}55'0''$  N,  $75^{\circ}49'0''$  E) were mechanically pressed and the resulting oil was centrifuged at  $3150 \times g$  for 20 min to remove solid material. Clear oil was collected and stored at room temperature (23 °C) for further studies.

## 2.2. Phorbol ester-rich extract preparation and Jatropha factors $C_1$ – $C_5$ isolation

Approximately 1 L of *J. curcas* seed oil (904.1 g) was extracted with methanol (MeOH, 750 ml) by stirring (300 rpm) in a 60 °C water bath for 15 min. The layers were allowed to separate then the MeOH layer was collected. The extraction process was repeated three times on the remaining oil. Combined MeOH extracts were concentrated *in vacuo* (60 °C) to yield a yellow oily residue (41.8 g). The crude MeOH extract was subjected to flash chromatography on 50 g of silica gel (40–63 µm, Merck), which had been preconditioned with dichloromethane (DCM). The column was eluted successively with DCM (500 ml), 1% DCM in MeOH (500 ml) then 5% DCM in MeOH (250 ml, collected in 50 ml fractions). Fractions containing PEs eluted with the 5% DCM in MeOH mixture and were concentrated *in vacuo* to yield a yellow PE-rich residue (0.26 g). The PEs were identified by a gradient elution high

performance liquid chromatography (HPLC) method described by Makkar et al. [3,8]. Briefly, samples were injected on a reverse phase  $C_{18}$  Licrospher 100, 5  $\mu$ m (250×4 mm) from Merck (Darmstadt, Germany), protected by a guard column of the same material. Starting with 60% solvent A (1.75 ml of 85% o-phosphoric acid in 1 L water and 40% solvent B (acetonitrile), B was increased to 50% over 10 min before increasing again to 75% over the next 30 min. From 30 to 45 min, B was increased to 100% before washing the column with solvent C (2% tetrahydrofuran in MeOH). Monitoring at 280 nm, four PEs peaks appeared between 25.5 and 30.5 min.

Additional oil was extracted, 1 L at a time, to yield a combined 2.0 g of the PE-rich product. This was subjected to semi-preparative HPLC on a reverse-phase  $C_{18}$ , 5 µm (250×10 mm) column from Phenomenex (California, USA). The column was eluted with 75% acetonitrile–water (0.01% trifluoroacetic acid) at a flow rate of 4 ml/min, monitored at 280 nm. Four prominent peaks were observed, which corresponded to *Jatropha* factors  $C_1$  (18.3 min),  $C_2$  (20.5 min),  $C_3$  (23.4 min) and a  $C_4 + C_5$  mixture (25.2 min). Each of the four samples was subjected to open column chromatography on Sephadex LH-20 for purification. *Jatropha* factors  $C_1$  and  $C_2$  obtained were confirmed by 1D ( $^1$ H,  $^{13}$ C) and 2D (HSQC, COSY, TOCSY, HMBC) NMR and comparison with data published by Haas et al. [6]. *Jatropha* factors  $C_3$  and the factor  $(C_4 + C_5)$  mixture needed further purification.

#### 2.3. Bioassay for toxicity in Artemia salina

The assay was carried out by a slight modification of the method described by Kinghorn et al. [9]. The *A. salina* cysts

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