



## Towards an understanding of induced-charge electrokinetics at large applied voltages in concentrated solutions

Martin Z. Bazant <sup>a,b,d,\*</sup>, Mustafa Sabri Kilic <sup>b</sup>, Brian D. Storey <sup>c</sup>, Armand Ajdari <sup>d</sup>

<sup>a</sup> Department of Chemical Engineering, Massachusetts Institute of Technology, Cambridge, MA 02139, United States

<sup>b</sup> Department of Mathematics, Massachusetts Institute of Technology, Cambridge, MA 02139, United States

<sup>c</sup> Franklin W. Olin College of Engineering, Needham, MA 02492, United States

<sup>d</sup> CNRS UMR Gulliver 7083, ESPCI, 10 rue Vauquelin, 75005 Paris, France

### ARTICLE INFO

Available online 8 October 2009

#### Keywords:

Nonlinear electrokinetics  
Microfluidics  
Induced-charge electro-osmosis  
Electrophoresis  
AC electro-osmosis  
Concentrated solution  
Modified Poisson–Boltzmann theory  
Steric effects  
Hard-sphere liquid  
Lattice-gas  
Viscoelectric effect  
Solvation  
Ionic liquids  
Non-equilibrium thermodynamics

### ABSTRACT

The venerable theory of electrokinetic phenomena rests on the hypothesis of a dilute solution of point-like ions in quasi-equilibrium with a weakly charged surface, whose potential relative to the bulk is of order the thermal voltage ( $kT/e \approx 25$  mV at room temperature). In nonlinear electrokinetic phenomena, such as AC or induced-charge electro-osmosis (ACEO, ICEO) and induced-charge electrophoresis (ICEP), several  $V \approx 100 kT/e$  are applied to polarizable surfaces in microscopic geometries, and the resulting electric fields and induced surface charges are large enough to violate the assumptions of the classical theory. In this article, we review the experimental and theoretical literatures, highlight discrepancies between theory and experiment, introduce possible modifications of the theory, and analyze their consequences. We argue that, in response to a large applied voltage, the “compact layer” and “shear plane” effectively advance into the liquid, due to the crowding of counterions. Using simple continuum models, we predict two general trends at large voltages: (i) ionic crowding against a blocking surface expands the diffuse double layer and thus decreases its differential capacitance, and (ii) a charge-induced viscosity increase near the surface reduces the electro-osmotic mobility; each trend is enhanced by dielectric saturation. The first effect is able to predict high-frequency flow reversal in ACEO pumps, while the second may explain the decay of ICEO flow with increasing salt concentration. Through several colloidal examples, such as ICEP of an uncharged metal sphere in an asymmetric electrolyte, we show that nonlinear electrokinetic phenomena are generally ion-specific. Similar theoretical issues arise in nanofluidics (due to confinement) and ionic liquids (due to the lack of solvent), so the paper concludes with a general framework of modified electrokinetic equations for finite-sized ions.

© 2009 Elsevier B.V. All rights reserved.

### Contents

|        |   |    |
|--------|---|----|
| 1.     | Introduction . . . . .  | 49 |
| 1.1.   | Nonlinear “induced-charge” electrokinetic phenomena . . . . . | 49 |
| 1.2.   | Scope and context of the article . . . . .                    | 50 |
| 2.     | Background: theory versus experiment . . . . .                | 51 |
| 2.1.   | The standard model . . . . .                                  | 51 |
| 2.1.1. | Electrochemical relaxation . . . . .                          | 51 |
| 2.1.2. | Viscous flow . . . . .  | 51 |
| 2.2.   | Open questions . . . . .                                      | 52 |
| 2.2.1. | The “correction factor” . . . . .                             | 52 |
| 2.2.2. | Electrolyte dependence . . . . .                              | 53 |
| 2.2.3. | Flow reversal . . . . .                                       | 54 |
| 2.3.   | Nonlinear dynamics in a dilute solution . . . . .             | 55 |
| 3.     | Crowding effects in a concentrated solution . . . . .         | 56 |
| 3.1.   | Mean-field local-density approximations . . . . .             | 56 |
| 3.1.1. | Modified Poisson–Boltzmann theories . . . . .                 | 56 |

\* Corresponding author. Department of Chemical Engineering, Massachusetts Institute of Technology, Cambridge, MA 02139, United States.  
E-mail address: [bazant@mit.edu](mailto:bazant@mit.edu) (M.Z. Bazant).

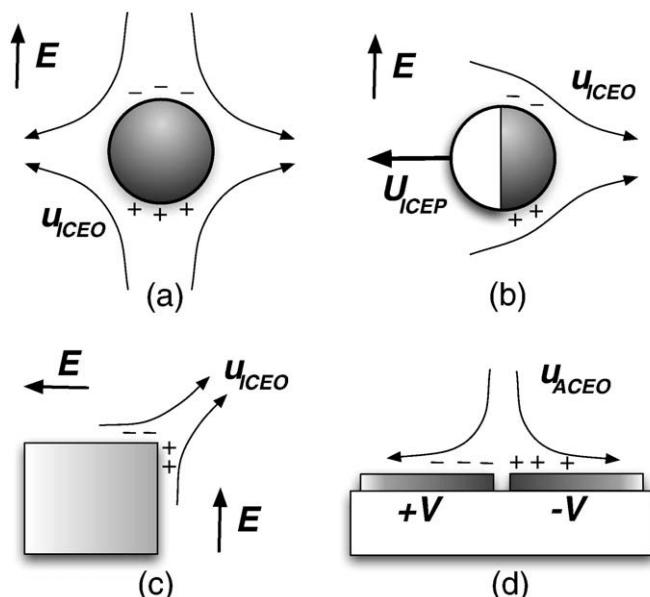
|                            |   |    |
|----------------------------|---|----|
| 3.1.2.                     | The Bikerman–Freise formula . . . . .   | 57 |
| 3.1.3.                     | Hard-sphere liquid models . . . . .   | 59 |
| 3.1.4.                     | Interpretation of the effective ion size . . . . .                                | 60 |
| 3.1.5.                     | Comparison with experiments on blocking surfaces without ion adsorption . . . . . | 61 |
| 3.1.6.                     | Dielectric response in a concentrated solution . . . . .                          | 62 |
| 3.2.                       | Implications for nonlinear electrokinetics . . . . .                              | 64 |
| 3.2.1.                     | High-frequency flow reversal of AC electro-osmosis . . . . .                      | 64 |
| 3.2.2.                     | Field-dependent electrophoretic mobility . . . . .                                | 66 |
| 4.                         | Viscoelectric effect in a concentrated solution . . . . .                         | 68 |
| 4.1.                       | Mean-field local-density theories . . . . .                                       | 68 |
| 4.1.1.                     | Modified Helmholtz–Smoluchowski slip formulae . . . . .                           | 68 |
| 4.1.2.                     | The viscoelectric effect . . . . .  | 69 |
| 4.1.3.                     | Charge-induced thickening . . . . .   | 70 |
| 4.2.                       | Implications for nonlinear electrokinetics . . . . .                              | 71 |
| 4.2.1.                     | Electro-osmotic slip at large voltages in concentrated solutions . . . . .        | 71 |
| 4.2.2.                     | Ion-specific electrophoretic mobility . . . . .                                   | 73 |
| 4.2.3.                     | Concentration dependence of AC electro-osmosis . . . . .                          | 75 |
| 5.                         | Mathematical modeling of electrokinetics in a concentrated solution . . . . .     | 75 |
| 5.1.                       | Nanoscale physics . . . . .   | 75 |
| 5.1.1.                     | Electrostatic correlations . . . . .  | 76 |
| 5.1.2.                     | Specific ion-ion interactions . . . . .   | 76 |
| 5.1.3.                     | Ion-surface correlations . . . . .  | 76 |
| 5.1.4.                     | Surface heterogeneity . . . . .   | 77 |
| 5.1.5.                     | Specific adsorption of ions . . . . .   | 77 |
| 5.1.6.                     | Normal current and Faradaic reactions . . . . .                                   | 77 |
| 5.2.                       | Modified electrokinetic equations . . . . .                                       | 78 |
| 5.2.1.                     | Continuum modeling approaches . . . . .   | 78 |
| 5.2.2.                     | Electrochemical transport . . . . .   | 78 |
| 5.2.3.                     | Electrostatics . . . . .  | 79 |
| 5.2.4.                     | Electrochemical hydrodynamics . . . . .   | 79 |
| 5.3.                       | Modified boundary conditions . . . . .  | 80 |
| 5.3.1.                     | Electrostatic boundary conditions . . . . .                                       | 80 |
| 5.3.2.                     | Electrochemical boundary conditions . . . . .                                     | 81 |
| 5.3.3.                     | Hydrodynamic boundary conditions . . . . .  | 81 |
| 5.4.                       | Thin double layers and diffusion layers . . . . .                                 | 82 |
| 6.                         | Conclusion . . . . .  | 82 |
| Acknowledgements . . . . . | 83  |    |
| References . . . . .       | 83  |    |

## 1. Introduction

### 1.1. Nonlinear “induced-charge” electrokinetic phenomena

Due to favorable scaling with miniaturization, electrokinetic phenomena are finding many new applications in microfluidics [1–3], but often in new situations that raise fundamental theoretical questions. The classical theory of electrokinetics, dating back to Helmholtz and Smoluchowski a century ago [4], was developed for the effective *linear* hydrodynamic slip driven by an electric field past a surface in chemical equilibrium with the solution, whose double-layer voltage is of order the thermal voltage,  $kT/e = 25$  mV, and approximately constant [5–10]. The discovery of AC electro-osmotic flow (ACEO) over micro-electrodes [11–13] has shifted attention to a new *nonlinear* regime [14], where the *induced* double-layer voltage is typically several  $V \approx 100 kT/e$ , oscillating at frequencies up to 100 kHz, and nonuniform at the micron scale. Related phenomena of induced-charge electro-osmosis (ICEO) [15–17] also occur around polarizable particles [18,19] and microstructures [20,21] (in AC or DC fields), as well as driven biological membranes [22]. Due to broken symmetries in ICEO flow, asymmetric colloidal particles undergo nonlinear, induced-charge electrophoresis (ICEP) [15,23–25]. Some of these fundamental nonlinear electrokinetic phenomena are illustrated in Fig. 1.

A “standard model” (outlined below) has emerged to describe a wide variety of induced-charge electrokinetic phenomena, but some crucial aspects remain unexplained. In their pioneering work 25 years ago in the USSR, which went unnoticed in the West until recently [15,16], V. A. Murtsovkin, A. S. Dukhin and collaborators first predicted



**Fig. 1.** Examples of nonlinear electrokinetic phenomena, driven by induced charge (+, −) in the diffuse part of the electrochemical double layer at polarizable, blocking surfaces, subject to an applied electric field  $E$  or voltage  $V$ . (a) Induced-charge electro-osmosis (ICEO) around a metal post [15,16,18,20], (b) induced-charge electrophoresis (ICEP) of a metal/insulator Janus particle [23,25], (c) a nonlinear electrokinetic jet of ICEO flow at a sharp corner in a dielectric microchannel [26,27], and (d) AC electro-osmosis (ACEO) over a pair of microelectrodes [11,12].

Download English Version:

<https://daneshyari.com/en/article/591095>

Download Persian Version:

<https://daneshyari.com/article/591095>

[Daneshyari.com](https://Daneshyari.com)