



Multifunctional dendritic mesoporous silica nanospheres loaded with silver nanoparticles as a highly active and recyclable heterogeneous catalyst



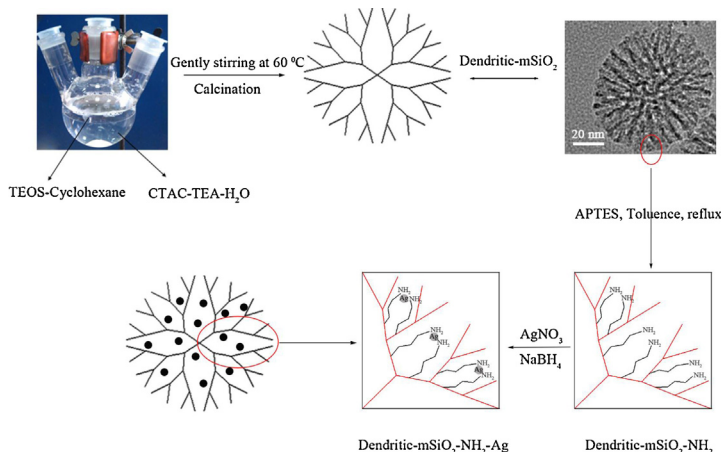
Zebin Sun, Guijia Cui, Haizhen Li, Yaxi Tian, Shiqiang Yan*

College of Chemistry and Chemical Engineering, Lanzhou University, Lanzhou 730000, PR China

HIGHLIGHTS

- A multifunctional dendritic silica nanocatalyst was successfully fabricated by an oil–water biphasic stratification coating strategy.
- The dendritic silica nanomaterials Dendritic- $mSiO_2$ were chosen as the catalyst's support for the first time.
- The as-synthesized nanocatalyst exhibited excellent catalytic activity and reusability due to easy accessibility of active sites.

GRAPHICAL ABSTRACT



ARTICLE INFO

Article history:

Received 8 September 2015
Received in revised form 26 October 2015
Accepted 26 October 2015
Available online 30 October 2015

Keywords:

Dendritic mesoporous silica
Dendritic- $mSiO_2$ - NH_2 -Ag
Nanocatalyst
Catalytic reduction
Nitro-compounds

ABSTRACT

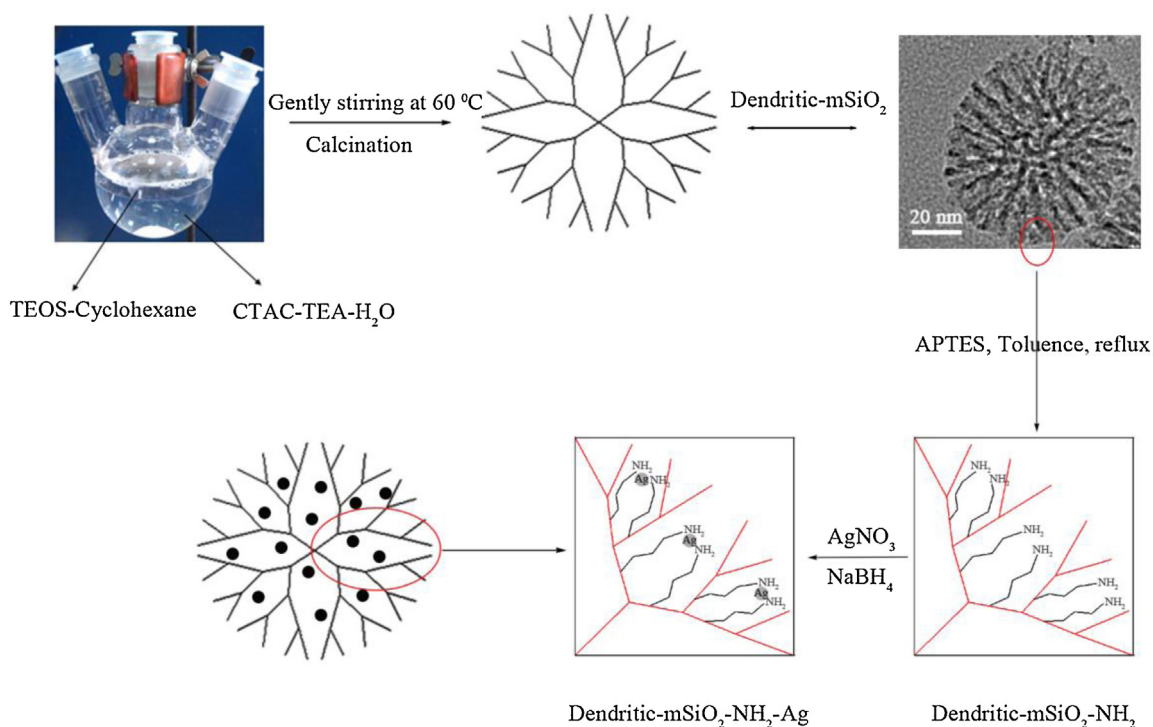
In this work, a multifunctional dendritic silica nanocatalyst Dendritic- $mSiO_2$ - NH_2 -Ag with easy accessibility of active sites and high surface area was successfully developed by an oil–water biphasic stratification coating strategy, and characterized by transmission electron microscopy, scanning electron microscopy, high-resolution transmission electron microscopy, X-Ray photoelectron spectroscopy, X-ray diffraction, N_2 adsorption–desorption, and Fourier transform infrared spectroscopy. The as-synthesized nanocatalyst Dendritic- $mSiO_2$ - NH_2 -Ag showed excellent catalytic activity for the catalytic reduction of 4-nitrophenol and 2-nitroaniline using sodium borohydride in aqueous solution at room temperature owing to easy accessibility of active sites. Additionally, the novel catalyst could be conveniently recovered from the reaction system and recycled for at least five times without obvious loss in activity. These results indicate that the aforementioned approach based on dendritic silica Dendritic- $mSiO_2$ provided a promising platform for the fabrication of noble metal nanocatalysts with easy accessibility and excellent catalytic activity, which could be highly efficient in various catalytic reduction reactions.

© 2015 Elsevier B.V. All rights reserved.

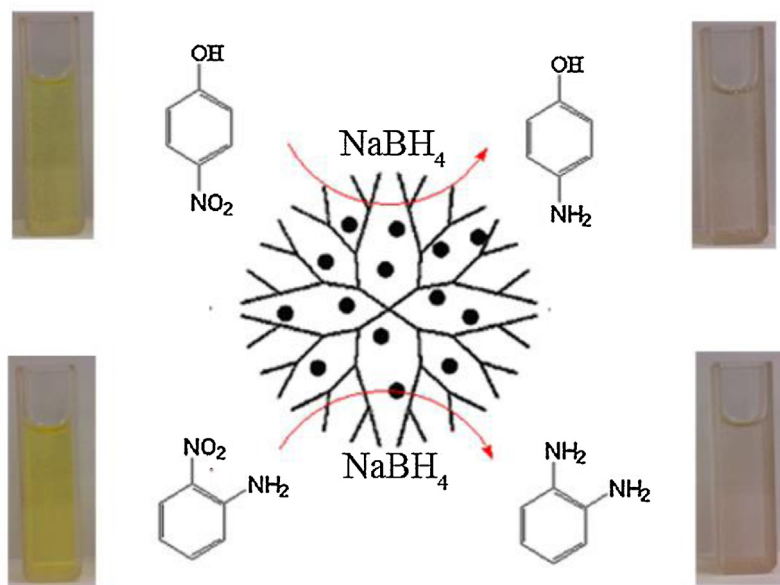
1. Introduction

Nitrophenol and its derivative are among the most common refractory water pollutants with high toxicity and carcinogenic

* Corresponding author. Fax: +86 931 8912582.
E-mail address: yansq@lzu.edu.cn (S. Yan).



Scheme 1. Synthesis procedure for the multifunctional dendritic mesoporous silica nanocatalysts (Dendritic-mSiO₂-NH₂-Ag) by combining the oil-water bi-phase stratification with a post-synthesis strategy.



Scheme 2. The reduction of 4-NP and 2-NA on the surface of Dendritic-mSiO₂-NH₂-Ag nanocatalyst.

character. In particular, 4-nitrophenol (4-NP) is a toxic industrial organic pollutant and environmentally hazardous material exhibiting high stability and solubility in wastewater, which can cause the harmful effects to animals, agricultural plants and human beings [1]. Moreover, the low concentration of 2-nitroaniline (2-NA) in water is harmful to human health and aquatic life in terms of its mutagenic effects, toxicity, and potential carcinogenic [2,3]. Therefore, the United States Environmental Protection Agency (EPA) has listed 4-NP and 2-NA as toxic wastes and priority hazardous pollutants. Hence, their removal from the environment is a crucial task. Till today, various methods such as microbial degradation,

electrocoagulation, electro-Fenton method, photocatalytic degradation, microwave-assisted catalytic oxidation, and electrochemical treatment have been utilized for the disposal of these nitro-compounds [4–6]. However, the catalytic reduction of nitro group to amino group is believed to be the most efficient, eco-friendly, and economical approach to dispose these nitro-compounds [7–10], which can be ascribed to that the reduction products such as 4-aminophenol (4-AP) and *o*-phenylenediamine (*O*-PDA) can be recycled because they are vital category of intermediates for the synthesis of drugs and dyes [2,11,12]. Therefore, it remains a challenge to develop a low-cost and highly effective

Download English Version:

<https://daneshyari.com/en/article/591878>

Download Persian Version:

<https://daneshyari.com/article/591878>

[Daneshyari.com](https://daneshyari.com)