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Modeling of the initial deposition of individual particles during the cross-flow membrane filtration

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HIGHLIGHTS

- Comprehensive force analysis was conducted on initial particle deposition on membrane.
- Semi-Brownian particles are mostly likely the membrane foulants.
- The effect of hydraulic factors on the deposition of individual particles can be simulated.
- Critical flux exists for semi-Brownian and Brownian particles during water filtration.

G R A P H I C A L A B S T R A C T

The particle within the viscous sub-layer may be exerted by three groups of forces, which include hydrodynamic forces (drag forces and lift force, f_{TD} , f_{D} and f_{L}), inherent force (the random Brownian force, f_{B}) and interfacial forces (hydrophobic force, f_{HP} , and electrostatic force, f_{EL}). The particle transport trajectories can be simulated. Under otherwise identical conditions, there exists a critical flux for a non-Brownian particle, above which the particle will be transported to and attached on the membrane surface, causing particle deposition.



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ABSTRACT

This study is devoted to the modeling of the deposition of individual particles onto a clean membrane surface in cross-flow filtration systems. Comprehensive force analysis approach is applied, which accounts for the random Brownian force and the polar component of the particle–membrane interactive forces. The inclusion of the polar interactive force is important in that when a hydrophilic membrane is involved, it can easily predominate over the rest of lateral forces in the near-field. The repulsive polar particle–membrane interaction can greatly decrease the stability of the particle on the membrane surface. In the far-field that is about 0.1 µm or farther away from the membrane, the particle transport is primarily dictated by the hydrodynamic lift and drag forces and the Brownian force. In sharp contrast to semi- or non-Brownian particles, the transport trajectory of Brownian particle is hardly definitive. The filtration flux and the cross-flow velocity can influence the particle transport trajectory of all sizes. Nevertheless, the existence of critical flux or critical cross-flow velocity is more evident for non-Brownian particles. Above the critical cross-flow rate or below the critical flux, particle deposition is minimized. Under appropriate operational conditions, a force-balanced level exists in the viscous sub-layer for a particular particle size, which is independent of the initial position of the particle. The model can be expanded further for more complicated water filtration conditions.

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1. Introduction

In the past decades, membrane filtration, especially microfiltration and ultrafiltration, has been increasingly attractive to water and wastewater treatment, primarily as the substitution for the

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	Nomen	clature		AB+	polar component a
	List of sy	List of symbols		$\gamma_{\rm m}$	tension of the men
	а	particle transport accelerating rate in the <i>z</i> direction (m/s ²)		$\gamma^{AB-}_{ m m}$	polar component a
	C _f	membrane friction factor		$\gamma_{\rm m}^{\rm LW}$	apolar component
	$C_{L,w}$	C_L when the particle is in contact with the mem-		$\gamma_{ m P}^{AB+}$	polar component a
	D	inner diameter of the membrane tubular flow chan-		$\gamma_{ m p}^{ m AB-}$	polar component a
	$d_{\rm p}$	particle diameter (m)		$\gamma_{ m p}^{ m LW}$	tension of the part apolar component
	e F F	Faraday constant (96485.3 C/mol)		$\gamma^{AB+}_{ m w}$	ticles (J/m ²) polar component a
	r _{at}	(N)		$\gamma^{AB-}_{ m w}$	tension of water (J, polar component a
	F _{dt}	(N)		r LW	tension of water (J, apolar component
	$f_{\rm AB}$	force of Lewis acid-base particle-membrane inter- action (N)		δ	(J/m ²) thickness of the v
	f _B f _D	Brownian force on a particle (N) hydrodynamic drag force on a particle (N)		0	boundary layer) (n
	$f_{\rm EL}$	force of electrostatic particle-membrane interac- tion (N)		ε ε ₀	dielectric permitti
	$f_{ m HP}$	force resulted from hydrophobic effect, i.e. the sum of f_{IW} and f_{AB} (N)		κ	Debye parameter (
	$f_{\rm L}$	hydrodynamic lift force on a particle (N) force of Lifebitz-yan der Waals particle-membrane		λ	water (0.6×10^{-9} n
	JLW	interaction (N)		μ	Water viscosity (Pa Water density (kg/
	ΔG_{AB}	a distance of h_0 (J/m ²)		$ ho_{s} au_{w}$	particle density (ka Membrane wall sh
	ΔG_{LW}	<i>LW</i> free energy per unit area of two parallel plates at a distance of h_0 (J/m ²)		Ψ_{d1}	potential of the particular fuse double layer b
	h	separation distance between the membrane and a particle (m)		Ψ_{d2}	potential of the me
	h ₀	minimum separation distance between plates due to Born repulsion $(0.157 \times 10^{-9} \text{ m})$			
	l i	IONIC STRENGTH (MOI/L) electrolyte charge number		conventio	nal deep-bed filtratio
	I	apparent filtration flux $(m^3/m^2/s)$	1	espective	ly. Membrane filtrati
	k _B	Boltzmann constant $(1.381 \times 10^{-23} \text{ J/K})$		echnolog	y to polish the treat
	R	gas constant (8.314 J/K/mol)		inique se	naration principle th
	Re	Reynolds number		water (an	d part or all of disso
	ке _р	particle Reynolds number surface roughpose of the membrane tube (m)	1	tion of pa	rticulate matters in t
	S T	absolute temperature (K)	1	retention	often leads to memb
	t	time (s)	1	nembran	e permeability [6,7].
	U	average cross-flow velocity in the filtration tube (m/s)		orane filtr which feat	ation processes are o tures that the water fi
	и	local water velocity (m/s)		eed flow.	a formation i a foulin
	u_x	local water velocity in the <i>x</i> direction (m/s)	i	f the appl	ied filtration flux is si
	u_z	local water velocity in the z direction (m/s)		super-crit	ical flux is applied in
	u*	membrane wall friction velocity (m/s)	i	nevitable	[11]. It has been rea
	v 11.	particle velocity (III/S) (m/s)	1	esult of t	he lateral transport
	v_{χ} v_{τ}	particle velocity in the z direction (m/s)	1	nembran	e (i.e. excessive forth
	x	cross-flow direction (Fig. 1)	i	n the far-	field region, followed
	Ζ	normalized distance from the membrane in the fil-		ittachmer	nt of the particles on
		tration direction	1	Soth the	lateral transport and
	Ζ	distance from the membrane in the filtration direction (Fig. 1)		notated b nodynam [15–17]. T	y a number of forces ic, electrodynamic, e Therefore, by using th
1			1.	an a circe!	la nantiala in the fla

Greek symbols

αβγ parameters in Eq. (11)

- s electron acceptor of the surface nbrane (J/m²)
- as electron donor of the surface nbrane (J/m²)
- of the surface tension of the
- s electron acceptor of the surface icles (J/m²)
- as electron donor of the surface icles (J/m²)
- of the surface tension of the par-
- s electron acceptor of the surface (m^2)
- as electron donor of the surface (m^2)
- of the surface tension of water
- iscous sub-layer (hydrodynamic n)
- constant of water (78.5)
- ivity of vacuum (8.854×10^{-12})
- 1/m
- Lewis acid-base interactions in n)
- as)
- m³)
- g/m^3)
- ear stress (Pa)
- rticle at the plane where the difpegins (V)
- embrane at the plane where the er begins (V)

on [1] and the secondary clarifier [2], ion is becoming the most promising ed water or wastewater for better ltration processes make use of the at selectively allows permeation of olved species) and therefore retenthe feed side [5]. However, particle brane fouling, which decreases the To minimize fouling, most memperated using the cross-flow mode, iltration is perpendicular to the bulk

g by particle deposition, is expected ufficiently low [8-10]. However, if a itially, cake formation could still be lized that particle deposition is the of the particles approaching to the -transport over the back-transport) by the near-field phenomena of the to the membrane surface [12–14]. d the near-field phenomenon are that have the hydrodynamic, therelectrostatic or mass-related origins he force analysis approach, usually on a single particle in the flow regime in the filtration apparatus, membrane fouling can be simulated by deriving the particle transport "trajectories" and evaluating the particle "stability" on the membrane surface if attached [12,14,18–22].

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