



## CO<sub>2</sub>-dependent carbon isotope fractionation in dinoflagellates relates to their inorganic carbon fluxes



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### ABSTRACT

Carbon isotope fractionation ( $\epsilon_p$ ) between the inorganic carbon source and organic matter has been proposed to be a function of  $p\text{CO}_2$ . To understand the CO<sub>2</sub>-dependency of  $\epsilon_p$  and species-specific differences therein, inorganic carbon fluxes in the four dinoflagellate species *Alexandrium fundyense*, *Scrippsiella trochoidea*, *Gonyaulax spinifera* and *Protoceratium reticulatum* have been measured by means of membrane-inlet mass spectrometry. In-vivo assays were carried out at different CO<sub>2</sub> concentrations, representing a range of  $p\text{CO}_2$  from 180 to 1200  $\mu\text{atm}$ . The relative bicarbonate contribution (i.e. the ratio of bicarbonate uptake to total inorganic carbon uptake) and leakage (i.e. the ratio of CO<sub>2</sub> efflux to total inorganic carbon uptake) varied from 0.2 to 0.5 and 0.4 to 0.7, respectively, and differed significantly between species. These ratios were fed into a single-compartment model, and  $\epsilon_p$  values were calculated and compared to carbon isotope fractionation measured under the same conditions. For all investigated species, modeled and measured  $\epsilon_p$  values were comparable (*A. fundyense*, *S. trochoidea*, *P. reticulatum*) and/or showed similar trends with  $p\text{CO}_2$  (*A. fundyense*, *G. spinifera*, *P. reticulatum*). Offsets are attributed to biases in inorganic flux measurements, an overestimated fractionation factor for the CO<sub>2</sub>-fixing enzyme RubisCO, or the fact that intracellular inorganic carbon fluxes were not taken into account in the model. This study demonstrates that CO<sub>2</sub>-dependency in  $\epsilon_p$  can largely be explained by the inorganic carbon fluxes of the individual dinoflagellates.

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### 1. Introduction

During photosynthetic carbon fixation, the lighter carbon isotope <sup>12</sup>C is preferred over the heavier carbon isotope <sup>13</sup>C, thereby causing carbon isotope fractionation ( $\epsilon_p$ ) between the inorganic carbon (C<sub>i</sub>) source and the organic carbon. Values for  $\epsilon_p$  of marine phytoplankton have been shown to be CO<sub>2</sub>-sensitive (e.g. Degens et al., 1968), and thus were discussed to serve as a proxy for past CO<sub>2</sub> concentrations (Jasper and Hayes, 1990; Pagani, 2014; Van de Waal et al., 2013; Hoins et al., 2015). Large species-specific differences in  $\epsilon_p$  have been

described, which are yet poorly understood (e.g. Hinga et al., 1994; Burkhardt et al., 1999). Moreover, irrespective of the phytoplankton species investigated, most of these studies have solely described the relationship between  $\epsilon_p$  and CO<sub>2</sub>, and only few have investigated the underlying physiological processes. Such mechanistic understanding is, however, needed to identify the reasons of the CO<sub>2</sub>-dependency of  $\epsilon_p$ .

Carbon isotope fractionation of phytoplankton is primarily driven by the enzyme ribulose-1,5-bisphosphate Carboxylase/Oxygenase (RubisCO), which is responsible for the fixation of CO<sub>2</sub> into organic compounds. The intrinsic fractionation associated with RubisCO ( $\epsilon_f$ ) has been estimated to range between ~22 and 30‰ (e.g. Roeske and O'Leary, 1984; Guy et al., 1993; Scott et al., 2007), even though a recent study obtained values as low as 11‰ for the RubisCO of the coccolithophore *Emiliania huxleyi* (Boller et al., 2011). While RubisCO principally sets the upper limit of fractionation, other processes strongly determine the degree to which RubisCO can express its fractionation (Sharkey and Berry, 1985; Burkhardt et al., 1999; Rost et al., 2002). First, there is leakage, i.e. the amount of CO<sub>2</sub> diffusing out of the cell in relation to C<sub>i</sub> uptake. With higher leakage, the intracellular C<sub>i</sub> pool is 'refreshed', thereby preventing accumulation of <sup>13</sup>C and allowing RubisCO to approach its upper fractionation values. Second, the relative

**Abbreviations:** C<sub>i</sub>, inorganic carbon; CCM, CO<sub>2</sub>-concentrating mechanism; Chl-*a*, Chlorophyll-*a*;  $\epsilon_p$ , carbon isotope fractionation;  $\epsilon_{p\text{-meas}}$ , measured carbon isotope fractionation;  $\epsilon_{p\text{-mod}}$ , modeled carbon isotope fractionation;  $\epsilon_s$ , equilibrium fractionation between CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup>;  $\epsilon_f$ , kinetic fractionation associated with the CO<sub>2</sub> fixation of RubisCO; L<sub>CO2</sub>, ratio of CO<sub>2</sub> efflux relative to total C<sub>i</sub> uptake; DIC, dissolved inorganic carbon; HCO<sub>3</sub><sup>-</sup>, bicarbonate; R<sub>HCO3</sub>, ratio of HCO<sub>3</sub><sup>-</sup> to total C<sub>i</sub> uptake; RubisCO, ribulose-1,5-bisphosphate Carboxylase/Oxygenase; CA, carbonic anhydrase; TA, total alkalinity.

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contribution of bicarbonate ( $\text{HCO}_3^-$ ) to total  $\text{C}_i$  uptake plays a role, as  $\text{HCO}_3^-$  is enriched in  $^{13}\text{C}$  by ~10‰ relative to  $\text{CO}_2$  (Mook et al., 1974). An increasing  $\text{HCO}_3^-$  contribution thus lowers  $\epsilon_p$ . The enzyme carbonic anhydrase (CA), which accelerates the otherwise slow interconversion between  $\text{CO}_2$  and  $\text{HCO}_3^-$ , can also influence  $\epsilon_p$  under certain conditions, e.g. by influencing leakage as well as the relative  $\text{HCO}_3^-$  contribution. All these processes play a role in the  $\text{CO}_2$ -concentrating mechanisms (CCMs) of phytoplankton. Assessing the mode of CCMs may therefore help to understand the reasons for  $\text{CO}_2$ -dependent changes in  $\epsilon_p$  and species-specific differences therein.

Dinoflagellates are cosmopolitan unicellular algae that occur in many different environments, including eutrophic coastal regions and oligotrophic open oceans. In this study, we investigated whether the  $\text{CO}_2$ -dependency of  $\epsilon_p$ , which was found in the dinoflagellate species *Alexandrium fundyense*, *Gonyaulax spinifera*, *Protoceratium reticulatum* and *Scrippsiella trochoidea* (Burkhardt et al., 1999; Hoins et al., 2015), can be explained by changes in their  $\text{C}_i$  fluxes. Characteristics of CCMs in the tested species, including their CA activities and  $\text{C}_i$  fluxes, were measured by means of membrane-inlet mass spectrometry (MIMS). Results were fed into a single-compartment model that considers cellular leakage, the relative  $\text{HCO}_3^-$  contribution as well as the carbon isotope fractionation of RubisCO (Sharkey and Berry, 1985; Burkhardt et al., 1999). The calculated carbon fractionation ( $\epsilon_{p\text{-mod}}$ ) was then compared to the measured carbon fractionation ( $\epsilon_{p\text{-meas}}$ ).

## 2. Material and methods

### 2.1. Incubations

Cultures of the dinoflagellate species *A. fundyense* (formerly *Alexandrium tamarense* strain Alex5; John et al., 2014), *S. trochoidea* (strain Geob267; culture collection of the University of Bremen), *G. spinifera* (strain CCMP 409) and *P. reticulatum* (strain CCMP 1889) were grown in 0.2  $\mu\text{m}$  filtered North Sea water (salinity 34), which was enriched with 100  $\mu\text{mol L}^{-1}$  nitrate and 6.25  $\mu\text{mol L}^{-1}$  phosphate. Metals and vitamins were added according to f/2 medium (Guillard and Ryther, 1962), except for  $\text{FeCl}_3$  (1.9  $\mu\text{mol L}^{-1}$ ),  $\text{H}_2\text{SeO}_3$  (10  $\text{nmol L}^{-1}$ ) and  $\text{NiCl}_2$  (6.3  $\text{nmol L}^{-1}$ ) that were added according to K medium (Keller et al., 1987). Each of the strains was grown in 2.4 L air-tight borosilicate bottles at 15 °C and  $250 \pm 25 \mu\text{mol photons m}^{-2} \text{s}^{-1}$  at a 16:8 h light:dark cycle. Bottles were placed on roller tables in order to avoid sedimentation.

Dissolved  $\text{CO}_2$  concentrations ranged from ~5–50  $\mu\text{mol L}^{-1}$  and were reached by pre-aerating culture medium with air containing 180, 380, 800 and 1200  $\mu\text{atm pCO}_2$ . The carbonate chemistry was calculated based on pH and total alkalinity (TA), using the program CO2sys (Pierrot et al., 2006). pH values were measured using a WTW 3110 pH meter equipped with a SenTix 41 Plus pH electrode (WTW, Weilheim, Germany), which was calibrated prior to measurements to the National Bureau of Standards (NBS) scale. An automated TitroLine burette system (SI Analytics, Mainz, Germany) was used to determine TA. Dissolved inorganic carbon (DIC) was determined colorimetrically using a QuAatro autoanalyser (Seal Analytical, Mequon, USA). For more details on the carbonate chemistry in the acclimations, please refer to Eberlein et al. (2014) for *A. fundyense* and *S. trochoidea* and to Hoins et al. (2015) for *G. spinifera* and *P. reticulatum*.

To determine  $\epsilon_p$  values, the isotopic composition of the organic material was measured using an Automated Nitrogen Carbon Analyser mass spectrometer (ANCA-SL 20–20, SerCon Ltd., Crewe, UK), and the isotopic composition of the DIC in growth medium was measured using a GasBench-II coupled to a Thermo Delta-V advantage isotope ratio mass spectrometer (see Hoins et al., 2015 for details on isotope analysis). Prior to assays, cells were acclimated to the different  $\text{CO}_2$  concentrations for at least 7 generations (i.e. >21 days). To prevent changes in the carbonate chemistry, i.e.

keeping drawdown of DIC <3%, incubations were terminated at low cell densities (<400 cells  $\text{mL}^{-1}$ ).

### 2.2. MIMS assays

A custom-made membrane-inlet mass spectrometer (MIMS; Isoprime, GV Instruments, Manchester, UK; see Rost et al., 2007 for details) was used to determine CA activities and  $\text{C}_i$  fluxes of *A. fundyense* and *S. trochoidea* acclimated to four different  $\text{pCO}_2$  (i.e. 180, 380, 800 and 1200  $\mu\text{atm}$ ; Eberlein et al., 2014), and of *G. spinifera* and *P. reticulatum* acclimated to a low and high  $\text{pCO}_2$  (i.e. 180 and 800  $\mu\text{atm}$ ). Assays were performed in an 8 mL temperature-controlled cuvette, equipped with a stirrer. Assay tests over ~1 h confirmed that conditions during the assay do not cause physiological stress (i.e. no decline in  $\text{O}_2$  production rates), and subsequent microscopic inspection did not reveal any visual effects on cell morphologies. Prior to the measurements, acclimated cells were concentrated using a 10  $\mu\text{m}$  membrane filter (Millipore, Billerica, MA) by gentle vacuum filtration (<200 mbar) and stepwise transferred into  $\text{C}_i$ -free medium buffered with a 4-(2-hydroxyethyl)-1-piperazine-ethanesulfonic acid (50  $\text{mmol L}^{-1}$  HEPES) solution at  $15 \pm 0.3$  °C and a pH of  $8.0 \pm 0.1$ . Chlorophyll *a* (Chl-*a*) concentrations were determined fluorometrically by using a TD-700 Fluorometer (Turner Designs, Sunnyvale, CA, USA) and ranged between 0.15 and 1.70  $\mu\text{g mL}^{-1}$  during the assays.

To quantify activities of extracellular CA (eCA), the  $^{18}\text{O}$  depletion rate of doubly labeled  $^{13}\text{C}^{18}\text{O}_2$  in seawater was determined by measuring the transient changes in  $^{13}\text{C}^{18}\text{O}^{18}\text{O}$  ( $m/z = 49$ ),  $^{13}\text{C}^{18}\text{O}^{16}\text{O}$  ( $m/z = 47$ ) and  $^{13}\text{C}^{16}\text{O}^{16}\text{O}$  ( $m/z = 45$ ) in the dark, following the approach of Silvermann (1982). If cells possess eCA, exchange rates of  $^{18}\text{O}$  are accelerated relative to the spontaneous rate. To monitor the spontaneous rate,  $\text{NaH}^{13}\text{C}^{18}\text{O}_3$  label was injected to the cuvette, waiting until the  $m/z = 49$  signal reached a steady-state decline. This rate was then compared to the steady-state decline after cells were added. Following Badger and Price (1989), eCA activity is expressed as percentage decrease in  $^{18}\text{O}$ -atom fraction upon the addition of cells, normalized to Chl-*a*. Consequently, 100 units (U) correspond to a doubling in the rate of interconversion between  $\text{CO}_2$  and  $\text{HCO}_3^-$  per  $\mu\text{g Chl-}a$ .

Photosynthetic  $\text{O}_2$  and  $\text{C}_i$  fluxes were determined following Badger et al. (1994). Making use of the chemical disequilibrium, this approach estimates  $\text{CO}_2$  and  $\text{HCO}_3^-$  fluxes during steady-state photosynthesis. It is based on the simultaneous measurements of  $\text{O}_2$  and  $\text{CO}_2$  concentrations during consecutive light and dark intervals with increasing amounts of DIC. Oxygen fluxes in the dark and light are converted into  $\text{C}_i$  fluxes by applying a respiratory quotient of 1.0 and a photosynthetic quotient of 1.1 (Burkhardt et al., 2001; Rost et al., 2003). The light intensity in the cuvette was adjusted to the acclimation conditions (i.e.  $250 \pm 25 \mu\text{mol photons m}^{-2} \text{s}^{-1}$ ). Net  $\text{CO}_2$  uptake was calculated from the steady-state decline in  $\text{CO}_2$  concentration at the end of the light period, corrected for the interconversion between  $\text{CO}_2$  and  $\text{HCO}_3^-$ . The uptake of  $\text{HCO}_3^-$  was calculated by subtracting net  $\text{CO}_2$  uptake from net  $\text{C}_i$  uptake, and the  $\text{CO}_2$  efflux from the cells was estimated from the initial slope after turning off the light. Rate constants  $k_1$  and  $k_2$  were determined based on temperature, salinity and pH (Zeebe and Wolf-Gladrow, 2001; Schulz et al., 2006), yielding mean values of  $0.9241 (\pm 0.0506) \text{ min}^{-1}$  and  $0.0085 (\pm 0.0008) \text{ min}^{-1}$ , respectively. To eliminate any eCA activity, a prerequisite to apply the rate constants, we added dextran-bound sulfonamide (DBS; 50  $\mu\text{mol L}^{-1}$ ) to the cuvette. For more details on the calculations, please refer to Badger et al. (1994) and Schulz et al. (2007).

### 2.3. Single-compartment model

To calculate  $\epsilon_{p\text{-mod}}$ , results for the relative  $\text{HCO}_3^-$  contribution and leakage were fed into a single-compartment model after Sharkey and

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