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Chlorination of bromide-containing waters: Enhanced bromate formation in the presence of synthetic metal oxides and deposits formed in drinking water distribution systems

Chao Liu, Urs von Gunten, Jean-Philippe Crou°

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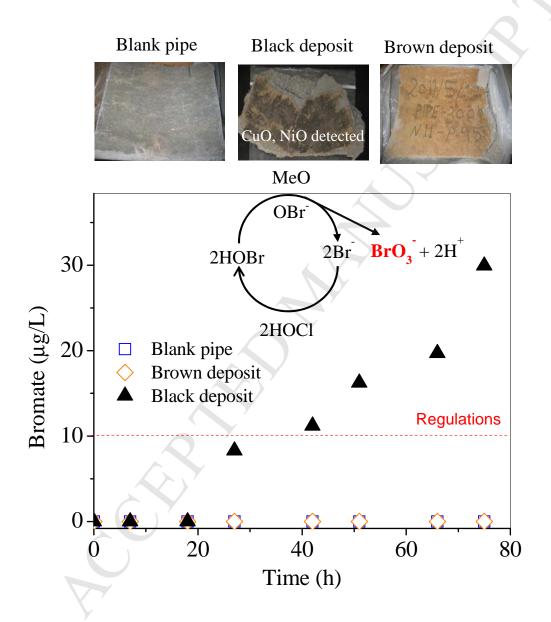
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## **Graphical Abstract**



 $[Cl_2]_0 = 2.5 \text{ mg/L}, [Br^-]_0 = 0.8 \text{ mg/L}, pH = 8.6, T = 21^{\circ}C$ 

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