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Low-temperature deposition of VO_2 films with high crystalline degree by embedding multilayered structure



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ABSTRACT

It is a long-standing challenge in the fabrication of single-phase VO₂ films at low temperature on Si substrate or glass substrate due to their thermodynamic phase-instability and the polymorphic structures. Recently, the demand for high-quality thin films deposited at low temperature is strongly pushed by their applications on devices and smart windows. Thermochromic films of VO₂, as well as multilayer films of VO₂/ZnO and VO₂/TiO₂/ZnO, were made by magnetron sputtering method and characterized by X-ray diffraction (XRD), atomic force microscopy (AFM), UV–visible-NIR spectrophotometer, (scanning) transmission electron microscopy ((S) TEM) and Electron energy loss spectroscopy (EELS). Films with high crystalline degree VO₂ single crystal phase were obtained at a fairly low temperature of 300 °C without subsequent annealing by embedding ZnO and TiO₂ layers between VO₂ and substrates. It is demonstrated that VO₂/TiO₂/ZnO films could display higher luminous transmittance and better semiconductor to metal transition (SMT) performance. Our data can serve a promising point for industrial production with high degree of crystallinity at a low temperature.

1. Introduction

Vanadium dioxide (VO₂) has attracted the interest because of its thermochromic semiconductor-to-metal transition (SMT) that exhibits tremendous contrast in optical, electrical and magnetic properties upon heating and cooling. VO₂ thin films are excellent materials for technological applications such as thermochromic coatings [1], ultrafast switching devices [2], sensors and micromechanical systems [3,4], etc. In particular, transparent VO₂-based thin films provide an attracting application in smart windows, that is, windows that capable of regulating solar/heat transmission for energy efficiency and comfort [5–7].

Physical deposition method plays a dominant role in VO₂-based thin films manufacture and there are numbers of specific methods for VO₂ film deposition [8–10]. However, most of them have the problem of co-deposition of impurity vanadium oxides (VO, V₂O₃, V₂O₅) [10– 12], causing degradation of thermochromic films properties. Generally, VO₂ film is deposited with substrate temperature of 350–500 °C for crystallization [1,8,10,13,14], which makes it difficult to produce with the present industrial production lines. Therefore, it is very meaningful to lower the temperature deposition of VO₂ films. Wang et al. reported a fabrication method of vanadium oxide films at a relatively low growth temperature of 200 °C, but it required post-growth annealing at 450 °C to get the VO2 (M) films [11]. Melnik et al. investigated lowtemperature method for thermochromic high ordered VO2 phase formation, which is also need two steps including post annealing process [12]. In our consideration, the post annealing treatment is not recommended for the easier nucleation of impurity vanadium oxides (enthalpy of formation $\Delta H_{V2O5} = -1557 \text{ cal/mole}, \Delta H_{V2O3} = -$ 1219 cal/mole, ΔH_{VO2} =- 713 cal/mole) and increasing mechanical stress due to the difference in thermal expansion coefficient α between film and substrate ($\alpha_{VO2} = 2.1 \times 10^{-5}$ /°C, $\alpha_{glass} = 3.3 - 8.5 \times 10^{-6}$ /°C). Zhang et al. proposed a 200 °C magnetron sputtering method for VO₂ thin film, where post annealing was inessential but a substrate bias voltage of -160 V was required [15]. Specific substrate bias voltage control can only be realized in small laboratorial sputtering devices but difficult in industrial production lines. Another low temperature (300 °C) method was proposed by Aijaz et al. [16], where the special equipment of high power impulse magnetron sputtering (HiPIMS) was applied. The inherent defects of high energy consumption and low

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deposition rate may keep HiPIMS away from practical application, thus new method for lowing sputtering temperature is still needed.

Differ from previous work, our paper proposed a novel low temperature (300 °C) preparation method without any post treatment process for high crystalline degree VO₂ films deposition by embedding multilayered structure. Since crystalline ZnO thin film can be relatively easily deposited among wide temperature range [17–19] and TiO₂ thin film always acts as buffer layer and growth template of VO₂ (M) films due to their similar lattice parameters [5,20,21], ZnO and TiO₂ thin film come to be the first choice of embedding layer. In our research, we systematically explored the possibilities of benefiting from SMT performance in multilayer stacks of the types VO₂/ZnO and VO₂/TiO₂/ZnO. Scanning transmission electron microscopy (STEM) and Electron Energy Loss Spectroscopy (EELS) results exhibited the good growth of VO₂ films and the interfaces. Some interesting results were found and this work can provide some good suggestions for low-temperature depositing VO₂ films with high performance.

2. Experimental

Films of VO₂ with single layer, bilayer (VO₂/ZnO), and trilayer (VO₂/TiO₂/ZnO) structure were fabricated on Si (100) or 50×50 mm² flat glasses substrates by magnetron sputtering deposition. Single stoichiometric phase VO2 thin films were synthesized by means of dc sputtering from a VO2 ceramic target. Compared to traditional reactive V metallic target sputtering, sputtering from a VO₂ ceramic target has an advantage of the stability of the deposited VO₂ stoichiometry being determined mainly by V-O stoichiometry in sputtering target. ZnO and TiO₂ layers were formed at bottom of VO₂ by rf magnetron sputtering from ZnO and TiO₂ ceramic target, respectively. All substrates were kept at elevated temperature 300 °C during sputtering, the sputtering temperature of 300 °C was chosen based on our early explorative experiments, in which the lower temperature (250 °C) sample showed poor crystallinity in STEM images and poor optical properties. The base pressure in sputtering chamber was 3×10^{-4} Pa, the dc gun power was 65 W, rf gun power was 80 W for both ZnO and TiO₂, sputtering time was 10 min for ZnO, 30 min for TiO₂, 60 min for VO₂ and the sputtering gas was supplied at 44 sccm for Ar and 1 sccm for O₂ to maintain 1.0 Pa. Schematic structure of the proposed multilayer structure is presented in Fig. 1.

The optical properties of the films were characterized by UVvisible-NIR spectrophotometer (HITACHI U-3100). The crystallinity and top morphology of the films were characterized by X-ray diffraction (XRD, Rigaku Ultima IV) and atomic force microscopy (AFM, SII Nano Technology Ltd, Nanonavi II). The crystal structure and chemical composition concerning the thin film growth and interface engineering were studied by STEM (JEOL-2100F). The bright field imagining (BF) and high angle annular dark field imaging (HAADF) were obtained along with STEM tests. EELS in the Cs-corrected scanning transmission electron microscope (NION USTEM200) were also taken.

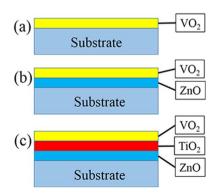


Fig. 1. Schematic structure of the proposed multilayer thin films with (a) single layer, (b) bilayer (VO_2/ZnO), and (c) triple layer ($VO_2/TiO_2/ZnO$) structure.

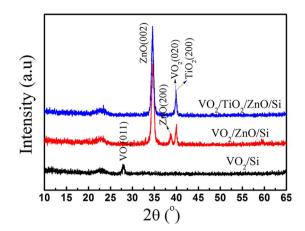


Fig. 2. XRD patterns VO₂ films with multilayer structure.

3. Results and discussion

3.1. Structure and surface analysis of VO₂-based multilayered films

Fig. 2 gives the XRD patterns of deposited films. The observed diffraction peaks in each pattern are successfully indexed to the ZnO, TiO2 or VO2(M) without impurity phase, confirming the purity of all the samples. VO₂ mainly shows (011) plane in single layer sample but (020) plane in bilayer and triple layer samples, which proves the influence on VO₂(M) orientation of embedding layer. XRD results also indicate the poor crystalized VO₂(M) in single layer sample and enhanced crystalline degree in triple concluding from the related peak intensity. ZnO displays main orientation of (002) and secondary orientation of (200) in bilayer sample but only (002) orientation remain in triple layer sample. It is probably attributed to the longer sputtering time in triple layer sample, ZnO keep the process of arrangement during TiO₂ sputtering and get a more uniform orientation. TiO₂ should appear as rutile TiO₂ (TiO₂ (R)) and show (200) plane, which can later be proved by TEM results, but the diffraction angle of 39.2° (PDF: 21-1276) is hidden by (020) of VO2(M) (PDF: 43-1051). AFM test was taken on the surface of the samples for calculating root mean square roughness (RMS). The RMS results are 5.47 nm for single layer, 3.98 nm for bilayer and 1.86 nm for triple layer sample. It reveals that VO₂ film in triple layer sample becomes relatively flat and the size of crystal grains becomes more uniform.

The microstructure analyses for these three sets of samples are shown in Figs. 3–6, respectively. For all three samples, there exhibit an amorphous layer of SiO_x between deposited layers and Si substrate, so we consider that the deposited layers show no epitaxial growth relationship with substrate. For single layer VO₂ sample (Fig. 3), it can be noticed that the VO₂ layer is nearly amorphous, only few crystal grain with the size of 10–20 nm can be found near the substrate. The crystalline is confirmed by high-resolution TEM images in Fig. 3(b). Therefore, the microscope results demonstrate the poor crystallinity of single layer VO₂ sample deposited with 300 °C, as coincident with XRD results (Fig. 2).

Fig. 4 is the STEM-BF and HAADF images for bilayer VO₂/ZnO sample. Compared with Fig. 3, by embedding 20 nm ZnO layer, both ZnO and VO₂ appear to be polycrystalline and better crystalized. It can be noticed that there is a clear transition layer with the thickness of ~ 2 nm between ZnO and VO₂ film (yellow lines in Fig. 4(b)), but the transition layer is not smooth and there are regions of poor crystallization or even amorphous near it. Poor crystallization or amorphous regions exhibit darker contrast in HAADF image of Fig. 4(a). The size of domain in ZnO is counted as ~15 nm in wide while ~10 nm in VO₂ layer, which means the grains of VO₂ and the grains of ZnO are not well corresponded. The out-of-plane orientation is considered mainly to be the (001) plane in ZnO and (010) plane in VO₂ based on the XRD

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