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# The influence of working gas characteristics on energy separation of vortex tube

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# HIGHLIGHTS

• Test device was designed, and R728, R744, R32, R22, R161, R134a were selected as working fluids.

• Performance of the vortex tube for energy separation with different working fluids was studied.

• Thermal diffusivity, kinematic viscosity and thermal conductivity were considered as main factors.

• Isenthalpic throttling effect of working fluids affected the cooling effect of the vortex tube.

• The conditions for the liquefying phenomena using HFCs were evaluated.

### ARTICLE INFO

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# ABSTRACT

Currently, little research has been done regarding the influence of working gases, especially HFCs, on energy separation in vortex tubes. Based on this, a vortex tube performance test device was designed, R728, R744, R32, R22, R161, R134a were selected as working fluids. The inlet pressure of the vortex tube was changed from 0.2 MPa to 1.3 MPa (abs), and the inlet temperature was adjusted to about 12 °C. The performances of the vortex tube with different working fluids were tested. It was found that the temperature separation and outlet pressure increased with the rise of inlet pressure for R728 and R744, but there was an extreme value for R744 when the inlet pressure reached about 1.1 MPa (i.e. when the temperature reached a peak point of about 15 °C), and three fluid characteristics (specific heat ratio, kinematic viscosity, thermal conductivity) were considered as main influencing factors of energy separation. The isenthalpic throttling effect of the working fluids also influenced the cooling effect of the vortex tube hot end were measured; the phenomena of the cold end of the vortex tube were observed; and the conditions for the liquefying phenomena using HFCs were evaluated.

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# 1. Introduction

The vortex tube is an effective energy separation device which is structurally simple and has no moving parts [1]. It can separate a high pressure gas into two fluids with distinctly different temperatures, hot and cold [2].

Although Ranque patented the first vortex tube in 1934, Hilsch [3] improved the design of vortex tube and did many experiments to study its principle. He thought that the friction of the vortex gas was the main reason which made the temperature gradient in the vortex tube. Fulton [4] also did research with vortex tubes and

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1359-4311/\$ – see front matter  $\odot$  2013 Elsevier Ltd. All rights reserved. http://dx.doi.org/10.1016/j.applthermaleng.2013.07.027 agreed with Hilsch's conclusions. In order to understand the inner counter flow of a vortex tube, Bruun investigated the characteristics of gases with different initial velocities, temperatures and pressures, etc. [5]. The experimental results showed that the initial velocity was predominantly tangential. On the other hand, Kurosaka [6] demonstrated through theoretical analysis and experiment that the acoustic streaming induced by orderly disturbances within the swirling flow was caused by the Rangue-Hilsch effect. Ahlborn and Gordon [7] quantified the performance of vortex tube as a thermodynamic machine and developed simple analytic formulas for the temperature and pressure profiles within the tube. In experiments carried out by Wang et al. [8], a vortex tube with six channels was chosen. The results showed that the cooling effect and the heating effect of the vortex tube increased with the increase of the inlet pressure, and the cold mass flow fraction decreased with the biggest cooling and heating effects. Xue [9]







studied the effect of the angle of rotating flow on the performance and efficiency of the Ranque—Hilsch vortex tube. Dincer et al. [10] conducted the exergy analysis and experimental investigation of a vortex tube with various nozzle cross-section areas. It was found that the exergy efficiency strongly depended on the level of inlet pressure, fraction of cold flow, and the velocity of cold stream.

In the above-mentioned work, many workers had conducted some basic research regarding the performance of the vortex tube for energy separation, and the conclusions were mainly obtained by using air as the working gas. However, the performance of the vortex tube is also influenced by the characteristics of working gases [11]. Based on these considerations, performance experiments of vortex tubes with different fluids (except air which is studied so much as working gas in vortex tube) have been conducted, and some of the outcomes are summarized in Table 1.

As shown in Table 1, it can be seen that these researchers mostly focused on the energy separation effect of different gases (i.e. air,  $O_2$ ,  $N_2$ , He,  $CO_2$ ). However, some problems still need to be addressed, for example, working gases such as HFCs widely used in refrigeration and air conditioning, have rarely been studied in previous research, and the influence of working gas characteristics on the energy separation in vortex tubes has not been fully discussed.

#### Table 1

Researches of vortex tube with different fluids.

| Year | Researcher           | Fluid  | Outcome  |
|------|----------------------|--|--|
| 1957 | Martynovskii<br>[12] | Ammonia,<br>CH <sub>4</sub> , CO <sub>2</sub>                | When $Pr_{turbulent} < 0.5$ ,<br>a temperature reversal occurs:<br>the hot fluid flows from the cold<br>end and the cold fluid flows from<br>the hot end. This phenomenon<br>was not explained.  |
| 1971 | Williams<br>[13]     | Methane-rich<br>mixture                                      | Under a given outlet pressure,<br>low temperature of inlet gas led<br>to the gas being liquefied in<br>the vortex tube, thus decreasing<br>the effect of energy separation.  |
| 1979 | Collins [14]         | Two-phase<br>propane   | When inlet pressure $p_{in} = 0.791$<br>MPa and quality $x < 80\%$ , then<br>the temperature separation<br>effect decreased.   |
| 1988 | Balmer [15]          | Water  | When inlet pressure $p_{in} = 20-30$<br>MPa, temperature separation ( $10-20$ °C) could be obtained in the exit, and all outlet temperatures were higher than the corresponding inlet temperatures.                                    |
| 1995 | Jin [16]             | Two-phase<br>fluid which<br>consisted<br>of water<br>and air | The vortex tube with the liquid also had the effect on energy separation.  |
| 1997 | Keller [17]          | R22, R134a,<br>R744  | Simulating the system with R22, R134a<br>and R744, and the COP of the system<br>using R22, R134a and R744 increased<br>by 5%, 10%, 2.5% respectively, while<br>the COP of the system using R22 by<br>the experiment increased by 1.5%. |
| 2003 | Poshernev<br>[18]    | Natural gas  | The outcome was similar to that of Williams [13].  |
| 2005 | Cao [19]             | N <sub>2</sub> , Ne, He                                      | The fluid with small coefficient of thermal conductivity enhanced the effect of energy separation in the vortex tube. $N_2$ had the lowest coefficient of thermal conductivity.  |
| 2006 | Dincer [20]          | O <sub>2</sub> , N <sub>2</sub> , CO <sub>2</sub>            | Compared with $O_2$ and $N_2$ , $CO_2$<br>could attain lower temperature<br>of hot fluid and cold fluid.   |
| 2006 | Aydin [21]           | Air, O <sub>2</sub> , N <sub>2</sub>                         | The molar mass of $N_2$ was smaller<br>than that of $O_2$ , thus its energy<br>separation effect was better.   |

Building on these previous works, the current project will expand the scope regarding the influence of working gas characteristics on the energy separation effect in vortex tubes. In the experiment, the fluids to be evaluated are R728 (nitrogen), R744 (carbon dioxide), R32, R22, R134a and R161. Also, the performance of energy separation in the vortex tube will be obtained when the inlet pressure is changed.

# 2. Experimental study

#### 2.1. Experimental setup

In this work, a vortex tube test device with different fluids was developed and built, as shown in Fig. 1. The gas from a pressurized gas source, GS, flows into the vortex tube through shut-off valve V0, diaphragm valve V1, heat exchanger EX, and mass flow meter M1, and then it is separated into two temperature fluids in the vortex tube. The cold fluid flows through mass flow meter M2, diaphragm valve V2, and the hot fluid flows through diaphragm valve V3. The inlet pressure of the vortex tube is adjusted by the diaphragm valve V1. The outlet pressures of the cold end and the hot end are adjusted by the diaphragm valves V2 and V3, respectively.

The EX is a concentric tube heat exchanger (outer tube is  $\varphi$ 19 mm  $\times$  0.9 mm, inner tube is  $\varphi$  6 mm  $\times$  0.6 mm, total length of heat exchanger is 3.4 m), and water for the heat transfer in the EX comes from tap water which has the same temperature with environment. The inlet temperature of the vortex tube is adjusted by this heat exchanger. The diaphragm valves V1, V2 and V3 are used to adjust the gas pressure. Because of the temperature's influence on the gas pressure, the inlet pressure of the vortex tube is adjusted by V1 and EX together. In the experiment, it is found that the inlet temperature fluctuation  $(\pm 3 \degree C)$  has little influence on the temperature separation effect. The entire apparatus is used for the experiment, and the working gases are discharged into the environment in a controlled manner, and then they are recycled, which makes the experiment environmental friendly. The vortex tube (Model: SCFM3202, EXAIR) is used in this experiment and its main dimensions are showed in Fig. 2 [22]. The measurement devices are placed as showed in Fig. 3. The measurement points of  $T_{in}$  and  $p_{in}$ are placed between M1 and the vortex tube (outside but near the inlet of vortex tube). The measurement points of  $T_c$  and  $P_c$  are placed between M2 and the vortex tube (outside and near the cold outlet of vortex tube). The measurement points of  $T_{\rm h}$  and  $p_{\rm h}$  are placed between V3 and the vortex tube (outside and near the hot outlet of vortex tube) and the details of the experimental instruments are showed in Table 2.

The platinum resistance thermometers are obliquely inserted into the brass tube to measure the fluid temperature. Because



Fig. 1. The experimental apparatus for the vortex tube performance testing.

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