

# Natural convection heat transfer utilizing ionic nanofluids with temperature-dependent thermophysical properties



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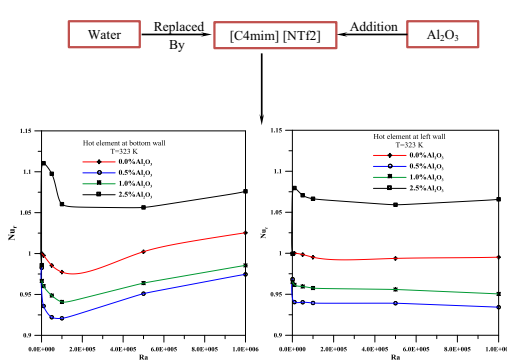
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## HIGHLIGHTS

- Numerical evaluation of an ionic liquids nanofluid is proposed.
- A comparison with an alumina nanofluid in terms of Nu is inserted.
- New Nu correlation for IoNanofluids were developed.

## GRAPHICAL ABSTRACT



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## ABSTRACT

Ionic liquid based nanofluids are a very new and novel class of fluids used for heat transfer. The thermophysical properties of these new fluids are extremely encouraging in comparison to base ionic liquids and recommend these new fluids for solar applications. This paper deals with a numerical implementation of an ionic liquid nanofluid in a square enclosure considering two heating situations; bottom heating and lateral heating of the enclosure. Comparison with a regular alumina nanofluid in terms of Nusselt number is inserted. Thermophysical properties of these ionic liquid based nanofluids were considered variable with temperature and the numerical results are correlated as a function of  $Ra$  and volume concentration ( $\phi$ ) for the two studied cases in the range of  $10^4 \leq Ra \leq 10^6$  and  $0\% \leq \phi \leq 2.5\%$ . As an overall conclusion, this innovative class of heat transfer fluids reveals great potential in advanced heat transfer applications.

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## 1. Introduction

Novel heat transfer fluids with a high thermal stability are desirable for both high temperature direct solar collectors and concentrated solar collector as well as for other important heat transfer applications. Recently, in the open literature can be found a lot of studies that considered nanoparticles suspended in various liquids

in order to increase the heat transfer capability of common fluids (Wilhelm and Letcher, 2014; Tan, 2011; Paul, 2014; Paul et al., 2015; Patankar, 1980; El-Maghlany et al., in press, 2014; El-Maghlany and Elazm, 2016; Teamah and El-Maghlany, 2010; Teamah et al., 2013; Aminossadati and Ghasemi, 2011; Liu et al., 2015). Most studied base fluids are water and ethylene glycol and the resulting new heat transfer fluids are usually called “nanofluids”.

On the other hand, ionic liquids, composed of organic cations and organic or inorganic anions (Rogers and Seddon, 2003), have been demonstrated to have a wide range of new applications, especially where common heat transfer fluids cannot be considered

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**Nomenclature**

$c_p$	specific heat, J/kg K
$g$	acceleration of gravity, m/s <sup>2</sup>
$k$	fluid thermal conductivity, W/m K
$L$	cavity width, m
$Nu$	average Nusselt number
$Nu_L$	local Nusselt number
$p$	pressure, N/m <sup>2</sup>
$P$	dimensionless pressure
$Pr$	Prandtl number
$Ra$	Rayleigh number
$T$	local temperature, K
$T_{av}$	average temperature, K
$T_c$	cold wall temperature, K
$T_h$	hot wall temperature, K
$\Delta T$	temperature difference, K
$u$	velocity components in x direction
$v$	velocity components in y direction
$U$	dimensionless velocity component in X direction
$V$	dimensionless velocity component in Y direction
$w$	the hot element length, m

$x, y$	dimensional coordinates
$X, Y$	dimensionless coordinates

*Greek symbols*

$\alpha$	thermal diffusivity, m <sup>2</sup> /s
$\beta$	coefficient of thermal expansion, K <sup>-1</sup>
$\theta$	dimensionless temperature
$\mu$	dynamic viscosity, kg/m s
$\nu$	kinematics viscosity, m <sup>2</sup> /s
$\rho$	local density, kg/m <sup>3</sup>
$\phi$	volume concentration

*Subscript*

correlated	refers to the points given by the correlation data
data	refers to the points obtained from numerical analysis
f	base fluid
ionf	IoNanofluids
nf	nanofluids based on water and alumina
r	ratio

(Welton, 1999). Their remarkable properties indicate that ionic liquids are the best candidates of heat transfer fluids in concentrated solar thermal systems (Liu et al., 2015; Rogers and Seddon, 2003;

Welton, 1999; Fredlake et al., 2004; Aparicio et al., 2010). Plus, the thermophysical properties of ionic liquids have been broadly reported in the open literature (Fredlake et al., 2004; Aparicio

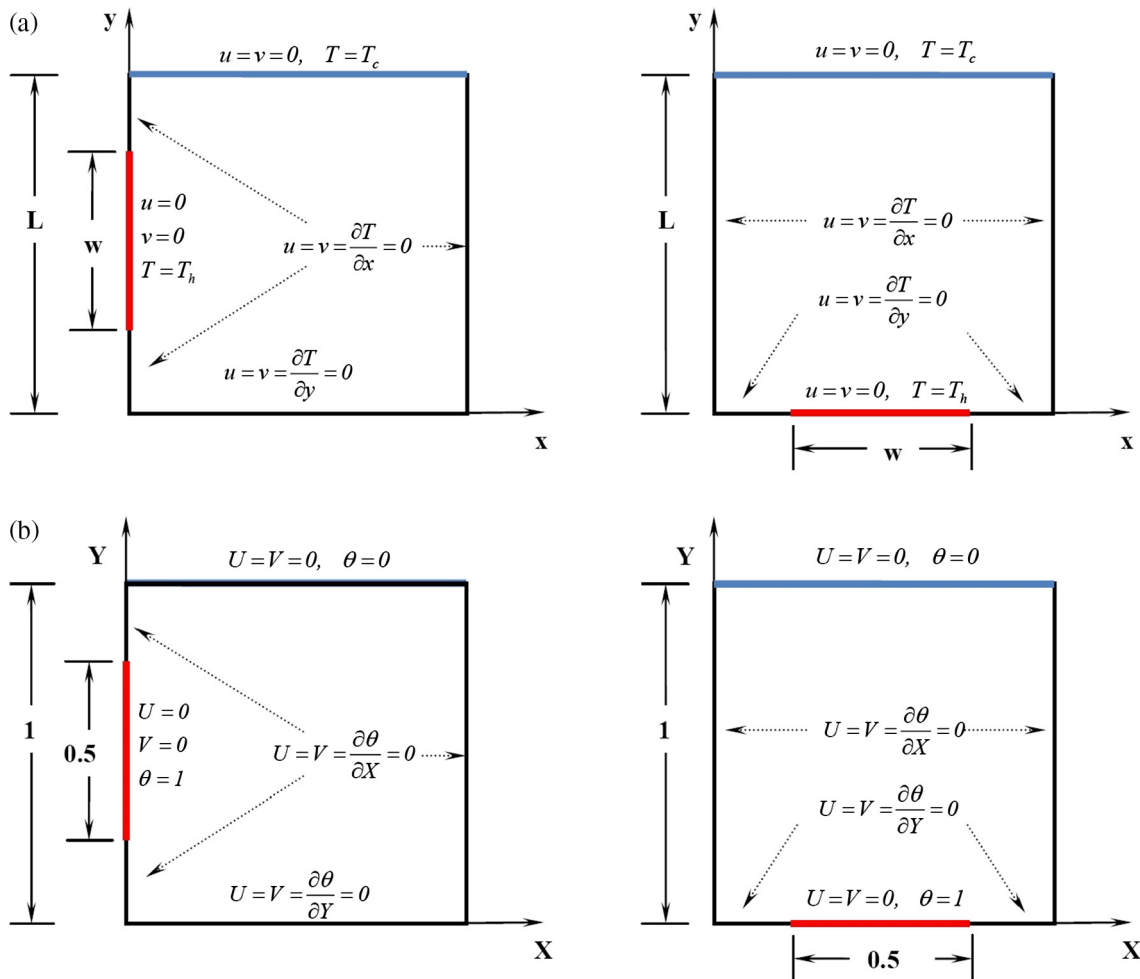


Fig. 1. A schematic diagram of the problem showing coordinates orientation and boundary conditions in dimensional (a) and dimensionless forms (b).

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