



Full Length Article

Enhanced oil recovery mechanism of CO₂ water-alternating-gas injection in silica nanochannelYouguo Yan^a, Chuanyong Li^a, Zihan Dong^a, Timing Fang^a, Baojiang Sun^b, Jun Zhang^{a,*}^a College of Science, China University of Petroleum, 266580 Qingdao, Shandong, People's Republic of China^b School of Petroleum Engineering, China University of Petroleum, 266580 Qingdao, Shandong, People's Republic of China

HIGHLIGHTS

- Molecular dynamics simulation was firstly used to address CO₂ WAG in nanochannel.
- The different roles of scCO₂ slug and water slug in CO₂ WAG injection were revealed.
- The synergistic effects of scCO₂ slug and water slug were proposed.

ARTICLE INFO

Article history:

Received 16 July 2016

Received in revised form 31 October 2016

Accepted 3 November 2016

Available online 9 November 2016

Keywords:

WAG

scCO₂

Nanoslit

EOR

Molecular dynamics simulation

ABSTRACT

CO₂ water-alternating-gas (WAG) injection has been proven as a promising and effective technology to significantly enhance the recovery of the residual oil. However, the detailed displacing process and exquisite interaction mechanism still remain ambiguous. By using Nonequilibrium Molecular Dynamic Simulation (NEMD), three comparative displacement processes of adsorbed oil layer in nanoslit using water, supercritical carbon dioxide (scCO₂), and CO₂ WAG injection were conducted to study the enhanced oil recovery (EOR) mechanism of CO₂ WAG injection. Simulation results indicate that the scCO₂ and water play different roles in CO₂ WAG: the scCO₂ slug could selectively dissolve apolar oil components, and the water slug provides stable water/scCO₂ promoting interface, which compels the miscible oil/scCO₂ slug out of the nanoslit. Furthermore, the synergistic effects of scCO₂ slug and water slug were found to account for a decreased injecting pressure and an improved dissolution capability of scCO₂ slug, which endows the CO₂ WAG higher oil recovery efficiency than that of both scCO₂ injection and water injection. Our work here provides a comprehensive interpretation of EOR mechanism of CO₂ WAG, and the results could provide insights into the optimization of the engineering process.

© 2016 Elsevier Ltd. All rights reserved.

1. Introduction

Enhancing oil recovery (EOR) is an unremitting pursuit in the petroleum field. After an initial quick extraction, most of the residual oil stays inside the reservoir on adsorption status [1,2]. In order to displace the residual oil, numerous oil-displacing agents have been developed [2–5], such as the surfactant, the polymer, nanoparticles, and the scCO₂. Among them, scCO₂ exhibits great advantages due to its unique properties, such as its large diffusion rate approximately equal to gas, and high dissolving capability similar to liquid [6,7]. ScCO₂-EOR has been confirmed to be a successful technique to reduce the residual oil saturation [5,8].

However, due to the extremely low viscosity, the displacement front of scCO₂ is inherently unstable and viscous fingering often

happens and causes an early breakthrough, resulting in low volumetric sweep efficiency [2,9]. To disentangle this problem, CO₂ WAG technique was developed. Experimental researches demonstrated that, in CO₂ WAG injection, the water slug could improve the volumetric sweep efficiency, and the scCO₂ slug could enhance the oil displacement ratio [10–12]. In current opinions, the CO₂ could diffuse into and swell the crude oil to reduce its viscosity [13]; meanwhile, the CO₂ also distributes at the interface between oil and water to reduce the interfacial tension [14,15], which favor forming of their miscible phase. In addition, the water dissolved with CO₂ could corrode and broaden the channel. These advantages will facilitate improving the EOR. However, limited to the detecting method to observe the displacing process of crude oil in opaque and nanoscale channel, the underlying mechanism of microscopic interaction mechanism and the synergistic effect between the scCO₂ and the water slug is not very clear. Therefore, the in-depth understanding the microscopic mechanism of the WAG is

* Corresponding author.

E-mail addresses: yyg@upc.edu.cn (Y. Yan), zhangjun.upc@gmail.com (J. Zhang).

fundamentally significant for improving EOR. Whereas, due to the nano- and micro-scale size of the porous reservoirs, conventional fluid theory (such as Darcy's law) is no longer suitable in such small scale [16]. Further, because of the invisibility of rock reservoirs and the great difficulty in preparing the synthetic core with a comparable porous structure to the real core, it is still a formidable challenge to obtain the detailed information of the displacing process.

In the past decades, molecular dynamics (MD) simulation has been fully developed and exhibits great advantages offering dynamic atomic-level information. In addition, some studies have been conducted to investigate the influence of the residual water [16], surface wettability [17], polarity of oil components [18], and various additives [4,17] on the oil displacement. And, the behavior of the scCO₂-substituted nanopore oil has drawn dramatically increasing research interest [19–21]. These studies shed light on the oil displacement behavior in water injection and scCO₂ injection. However, to the best of our knowledge, there exist very limited studies on the CO₂ WAG injection in nanopore [22,23]. The detailed displacing process and microscopic interaction is still unclear.

In this work, under the employment of NEMD simulations, three parallel displacing processes using scCO₂, water and CO₂ WAG injections were investigated. The roles of scCO₂ and water slugs in the CO₂ WAG injection were analyzed, and some synergistic effects of scCO₂ slug and water slug were found to be responsible for high oil recovery efficiency. Our work provides an atomic-level understanding of the CO₂ WAG process, and the results would be useful to realize some engineering phenomena in CO₂ WAG and guide the application of scCO₂ in the tertiary oil recovery efficiency.

2. Experimental and simulation

2.1. Nanoscale slit

The initial silica lattice was derived from the database of the Material Studio software, and a α -quartz silica supercell with orthorhombic dimensions (xyz) of $24.6 \times 81.1 \times 97.6 \text{ \AA}^3$ was constructed, the detailed building process could be seen in [Supporting information \(SI\)](#). Considering there are abundant nanoscale pores in the size range of 30–80 $\text{\AA}$ in rock reservoir [24], a nanoslit with a height of 63 $\text{\AA}$ along y-axis was built through deleting the middle part of silica supercell, and the exposed surface was (0 0 1). The silica surface was hydroxylated by hydroxyl with a density of 9.6 nm^{-2} , which is consistent with the result of crystal chemistry calculations ($5.9\text{--}18.8 \text{ nm}^{-2}$) [25].

2.2. Adsorbed oil layer

Here, decane and asphaltene were selected as representative apolar and polar components of crude oil. The molecular configurations of decane and asphaltene are displayed in [Fig. 1a](#) and [b](#), respectively. A detailed explanation for the selection of these two oil components could be found in [SI](#). To obtain equivalent adsorbed oil layer, 10 asphaltene and 40 decane molecules was spread evenly onto the silica surface with dimensions (xz) of $24.6 \times 68.0 \text{ \AA}^2$, and an energy minimization and a subsequent 1 ns MD simulation were performed [27]. In order to conveniently observe the displacement consequence, these two adsorbed oil layers were placed onto the left side of the nanoslit. In the adsorbed oil layer, due to the strong interaction between the polar asphaltene and the hydroxylated surface, asphaltene molecules could preferentially adsorb onto the surface, meanwhile the decane molecules adsorb onto the unoccupied silica surface and wrap

the adsorbed asphaltene molecules [26–29], as presented in [Fig. 1d–f](#).

2.3. Fluid

Based on the typical reservoir condition, a temperature of 330 K and pressure of 20 MPa were selected in our simulation, and the corresponding density of scCO₂ (0.743 g/cm^3) and water (0.993 g/cm^3) were adopted in the initial model, according to *National Institute of Standards and Technology's (NIST) database (2011)* [30]. By adding certain number of CO₂ and water molecules into box, the pressure (20 MPa) at 330 K could be maintained. The amount of added CO₂ and water molecules could be calculated from the Peng–Robinson equation according to their density [31], and is listed in [Table S1 in SI](#). As shown in [Fig. 1d](#) (scCO₂) and [e](#) (water), a fluid box with dimensions (xyz) of $24.6 \times 81.1 \times 130.0 \text{ \AA}^3$ was set on the left side of silica nanoslit which was also filled with the fluid. As to the CO₂ WAG model ([Fig. 1f](#)), it is composed of a water slug (left) and a scCO₂ slug (middle) with same length of 65 $\text{\AA}$, and the nanoslit filled with water. In these three models, a silica sheet was placed on the left of the fluid as slab to push the fluid across the channel.

2.4. MD simulation

The NEMD simulation was performed with LAMMPS package [32] through employing EPM2 CO₂ model [33] and a simple point charge (SPC) water [34]. The CLAYFF force field [35] and CHARMM force field [36,37] were used for silica and oil molecules, respectively. Detailed parameters could be found in [Table S2 in SI](#).

The MD simulations were conducted in NVT (the number of molecules, volume and temperature are constant) ensemble at 330 K with the Nosé–Hoover thermostat [38]. During simulation, the leftmost silica sheet was defined as a rigid body and a constant velocity 1 m/s along z axis was applied to it to push the fluid across the channel; and the two silica slabs of nanoslit were kept frozen except for the modified hydroxyl. The periodic boundary condition was applied in all directions. The particle-particle particle-mesh (PPPM) algorithm [39] with a convergence parameter of 10^{-4} was used to treat the long-range electrostatic interactions. The van der Waals interactions were calculated with a cutoff (10 $\text{\AA}$). A time step of 1 fs was used, and the corresponding data was collected in every 5 ps. A 13 ns simulation was conducted for all three systems. The configuration snapshots were rendered by VMD software [40].

3. Results and discussion

In the WAG injection process, the scCO₂ slug and water slug will alternately pass through the reservoir pore. Sometime the pore was filled with the scCO₂ slug, and in the next period, the pore will be occupied by the water slug. Hence, the displacement behaviors in the single scCO₂ injection and single water injection were analyzed firstly to unveil their roles in the displacement process. And then, a detailed CO₂ WAG injection process was studied to unveil the EOR mechanism of the CO₂ WAG.

3.1. The displacement behavior of scCO₂ and water

3.1.1. Displacement phenomena

[Fig. 2](#) presents several sequential snapshots in the scCO₂ injection. As shown, nearly all decane molecules dissolve into the scCO₂ phase, and asphaltene molecules could only slip on but cannot detach from the silica surface. Detailed observation manifests that

Download English Version:

<https://daneshyari.com/en/article/6475582>

Download Persian Version:

<https://daneshyari.com/article/6475582>

[Daneshyari.com](https://daneshyari.com)