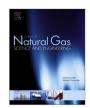


Contents lists available at ScienceDirect

Journal of Natural Gas Science and Engineering

journal homepage: www.elsevier.com/locate/jngse



Effect of gas composition produced by gasification, on the performance and durability of molten carbonate fuel cell (MCFC)



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ARTICLE INFO

Article history:
Received 21 March 2016
Received in revised form
25 July 2016
Accepted 30 August 2016
Available online 13 September 2016

Keywords:
Biomass
Durability
Gas composition
Gasification
MCFC
Performance

ABSTRACT

Utilization of biomass using gasification technology combined with Molten Carbonate Fuel Cell (MCFC), is an option to solve fossil fuels and environmental pollution problem. Various types of biomass give different composition of the gas produced by gasification process and indirectly affect the performance and durability of MCFC.

This research is focused on determination of gas composition produced by gasification process (CO, CO_2 , H_2 and N_2) that gives optimum performance and durability of MCFC either in dry or wet condition. Using mixture design method, the gas composition supplied to MCFC anode is varied in the range of 14% -25% for CO_2 , -25% for CO_3 , and -25% for CO_3 , and -25% for CO_3 . The effect of feed gas composition on the performance and durability of MCFC is characterized by using potentiodynamic method and accelerated load cycle method respectively. SEM and XRD is used in this study to evaluate gas composition effect on MCFC anode morphology.

Based on experiments that are carried out, the optimum result is obtained by using syn-gas from coconut shell gasification with composition of 25% CO, 12% H_2 , 10% CO_2 and 53% N_2 in wet condition. Electrical power of 0.13 W is generated within 228 days of durability or it is decreased by 135 mV/1000 h or 16% (v/v)/1000 h.

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1. Introduction

The increase of electrical energy demand and greenhouse gases are the main factors that drive the urge to utilize alternative source of energy which is both environmentally friendly and capable to increase national electricity production. Indonesia is abundant in some sources of biomass that are potential to be utilized as alternative source of energy as shown in Table 1.

Gasification is a mature technology for industrial scale application to convert biomass into syngas. The composition of syngas produced from gasification depends on its biomass source. Table 2 below shows some studies on syngas production from gasification of biomass that is abundantly available in Indonesia.

Based on the studies listed in Table 2, the composition of biomass-based syngas generally lies in particular compositional range as shown in Table 3.

One of fuel cell that is suitable for converting syngas directly is molten carbonate fuel cell (MCFC). MCFC operates at temperature of 650 °C and able to directly convert CO and H₂ into electrical energy continuously (Bagotsky, 2006). Molten carbonate with LiAlO₂ matrix acts as electrolyte in MCFC, while Ni act as anode and NiO act as cathode. The following is the general reaction mechanism that takes place in MCFC (Bodén, 2007).

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Anode
$$2H_2 \rightarrow 4H^+ + 4e^-$$
 (1)

Gasification process is commonly followed by syngas combustion for power generation. However, combustion causes environmental problem as it produces side flue gas containing greenhouse gases. Another technology option to couple gasification process in power generation is by using fuel cell. Fuel cells are electrochemical devices that is capable in converting chemical energy derived from biomass gasification into electrical energy directly and continuously with water and heat as byproduct (Larminie and Dicks, 2003).

Table 1Potential biomass in Indonesia (EBTKE, 2012).

Biomass	Production (ton/year)	Energy (GJ/year)
Rubber	31,0 million	496,0 million
Logging waste	1,2 million	11,0 million
Residual from wood sawmill industry	1,1 million	10,6 million
Palm empty fruit bunch	3,5 million	15,4 million
Palm fiber	3,7 million	35,3 million
Palm fruit shell	1,3 million	17,2 million
Sugarcane bagasse	6,5 million	78,0 million
Rice husk	14,3 million	179,0 million
Coconut shell	1,1 million	18,7 million
Coconut coir	2,0 million	24,0 million

Table 2Results of biomass gasification (% v/v).

Biomass source	СО	H ₂	CO ₂	N ₂	Author
Rubber wood	18	16	10	54	(Susanto, 1986)
Wood waste	17	16	10	55	(Rajvanshi, 1986)
Palm empty	15	12	20	51	(Lahijani et al., 2013)
fruit bunches	14	9	14	56	(Moni and Sulaiman, 2013)
Palm fruit shells	20	11	10	58	(Susanto, 1986)
Sugarcane bagasse	15	15	12	56	(Rajvanshi, 1986)
Rice husks	16	10	10	57	(Rajvanshi, 1986)
	20	11	11	55	(Susanto, 1986)
Coconut shell	19	10	11	54	(Rajvanshi, 1986)
	25	12	10	51	(Susanto, 1986)
Coconut coir	16	17	10	55	(Rajvanshi, 1986)

Table 3General composition of biomass-based syngas.

Type of gas	CO	H_2	CO ₂	N_2
Composition (% v/v)	14-25	9-17	10-20	51-58

Cathode
$$O_2 + 4H^+ + 4e^- \rightarrow 2H_2O$$
 (2)

Total reaction
$$2H_2 + O_2 \rightarrow 2H_2O$$
 (3)

MCFC has a high efficiency and produce energy on a large scale (100 kW - 10 MW). Therefore, process coupling between gasification and fuel cell technology is a promising solution to produce clean energy.

Studies on the combination of biomass gasification technology and MCFC are mostly focused only on the effect of one component of the syngas on the performance of MCFC. Meanwhile, the influence of biomass source and biomass-based syngas composition to the performance and durability of MCFC has not been intensively examined. Therefore, the purpose of this research includes:

- Determining the most suitable biomass to produce syngas as MCFC anode fuel;
- Estimating the amount of power generated from gasification using potential biomass source in Indonesia;
- Analyzing the influence of CO, CO₂, H₂, and N₂ composition in syngas and their interaction on the performance of MCFC; and
- Predicting the durability of MCFC for each syngas produced from gasification of the potential biomass in Indonesia.

2. Methodology

The overall method of experiment conducted in this research is presented in Fig. 1. The study is started by characterizing MCFC anode using XRD and SEM to determine the initial condition of anodes. The step is followed by designing experiment framework using mixture design method in Minitab 16.1.1 to determine the feed rate at anode. The feed rate at the cathode is set constant at 68 ccm for CO₂ and 162 ccm for compressed air. This is done to determine the optimum conditions for each component presents at anode feed gas, to observe the effect of each component on the

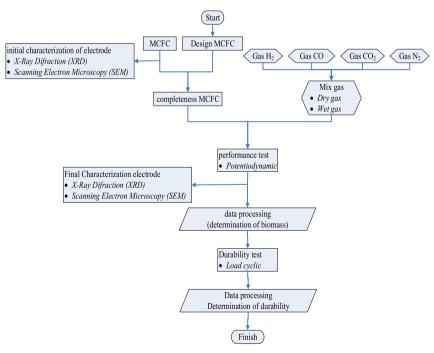


Fig. 1. Experiment method flow diagram.

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