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Ultrastructural organization and micromechanical properties of shark tooth enameloid $\stackrel{\text{\tiny{\scale}}}{=}$

7 Q1 Joachim Enax^a, Anna M. Janus^b, Dierk Raabe^b, Matthias Epple^a, Helge-Otto Fabritius^{b,*}

^a Institute of Inorganic Chemistry and Center for Nanointegration Duisburg-Essen (CeNIDE), University of Duisburg-Essen, Universitaetsstr, 5-7, 45117 Essen, Germany 8 9 ^b Microstructure Physics and Alloy Design, Max-Planck-Institut für Eisenforschung GmbH, Max-Planck-Str. 1, 40237 Düsseldorf, Germany

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ABSTRACT

The outer part of shark teeth is formed by the hard and mineral-rich enameloid that has excellent mechanical properties, which makes it a very interesting model system for the development of new bio-inspired dental materials. We characterized the microstructure, chemical composition and resulting local mechanical properties of the enameloid from teeth of Isurus oxyrinchus (shortfin mako shark) by performing an in-depth analysis using various high-resolution analytical techniques, including scanning electron microscopy, qualitative energy-dispersive X-ray spectroscopy and nanoindentation. Shark tooth enameloid reveals an intricate hierarchical arrangement of thin (50-80 nm) and long (>1 µm) crystallites of fluorapatite with a high degree of structural anisotropy, which leads to exceptional mechanical properties. Both stiffness and hardness are surprisingly homogeneous in the shiny layer as well as in the enameloid: although both tooth phases differ in structure and composition, they show almost no orientation dependence with respect to the loading direction of the enameloid crystallites. The results were used to determine the structural hierarchy of shark teeth, which can be used as a base for establishing design criteria for synthetic bio-inspired and biomimetic dental composites.

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1. Introduction 41

Shark teeth have different macroscopic geometries which are 42 43 evolutionarily optimized for their specific biological function, i.e. 44 the way of hunting and biting. The teeth can be classified by their geometry, e.g. as "tearing-type", "cutting-type" or "cutting-clutch-45 ing type" [1–4]. The chemical and crystallographic composition of 46 shark teeth of such different shape and function are still very sim-47 ilar [5]. Morphologically, the teeth of sharks consist mainly of den-48 49 tin that is covered by an outer hard and highly mineralized layer in 50 the crown area [6]. In, for example, reptiles and mammals, includ-51 ing humans, this outermost layer of the teeth is denoted as enamel [7,8]. In sharks, the external tooth layer has no ectodermal enamel 52 53 and is, therefore, denoted as "enameloid" [9]. The enamel of mammalian teeth, including human teeth, consists of hydroxyapatite 54 (Ca₅(PO₄)₃OH), associated with small amounts (\sim 1 wt.%) of an 55 56 organic matrix composed mainly of the proteins amelogenin and 57 enamelin [10,11]. The mineral phase of shark tooth enameloid con-58 sists of fluoroapatite $(Ca_5(PO_4)_3F)$ [12–15], with a fluoride content

Corresponding author. Tel.: +49 (0) 211 6792 373; fax: +49 (0) 211 6792 333. E-mail address: h.fabritius@mpie.de (H.-O. Fabritius).

nearly as high as that of geological fluoroapatite crystals (3.1 and 3.64 wt.%, respectively) [5]. The total enameloid contains \sim 5–8 wt.% of organic matrix consisting of collagens and enamelins [5,16–18]. In mammalian and human tooth enamel, the hydroxyapatite forms needle-like crystallites organized in bundles ("enamel prisms") that originate at the dentin-enamel junction and are oriented perpendicular to the tooth surface [19–21]. The outermost layer ("prismless layer") of human enamel consists of parallel oriented needle-like crystallites [22]. The fluoroapatite in shark tooth enameloid is also present in the form of elongated crystallites that constitute layers with different structural organization. These layers have been classified by the structural analysis of surface etched sectional samples [23]. Three layers of the enameloid were identified and denoted as "shiny-layered enameloid (SLE)", "parallel-bundled enameloid (PBE)" and "tangled-bundled enameloid (TBE)". Since the PBE and TBE show a gradual transition, the enameloid of sharks is generally thought to be organized into two main structural building blocks: a superficial layer (shiny layer) and an inner layer consisting of crystallite bundles with changing degrees of structural organization from distal to proximal [24]. From exterior to interior, the well-organized crystallite bundles of the PBE change to a less ordered TBE. Reaching the dentin-enameloid junction, no defined bundles are visible in the TBE, but randomly arranged crystallites [24]. The formation of

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83 shark tooth enameloid can be subdivided into three phases of 84 development: first, the formation of the matrix; second, the miner-85 alization of the enameloid; and finally, the maturation of the 86 enameloid [25-27]. The individual fluoroapatite crystallites nucle-87 ate and grow in tubular vesicles formed by the odontoblasts. Dur-88 ing the mineralization stage, the vesicles disappear and individual 89 crystallites in different states of organization can be observed 90 entangled between organic, presumably collagen molecules [9]. 91 However, little is known about the structural organization of 92 matured shark tooth enameloid at a scale smaller than the crystal-93 lite bundles and whether and how the structure and composition 94 influence the local mechanical properties of the material.

Recently, the correlation of structure, composition and mechan-95 ical properties in biological hard tissues [28,29] has become an 96 97 important field of research that aims at the development of novel 98 biomimetic and bio-inspired materials [30,31]. The mechanical 99 properties of shark tooth enameloid have been investigated using 100 mainly micro- and nanoindentation techniques [5,32]. However, 101 this has not been performed with a resolution sufficient to analyze the local mechanical properties of small building units such as 102 103 individual crystallite bundles. Therefore, we have chosen the teeth 104 of the recent shark species Isurus oxyrinchus (shortfin mako shark) 105 to study the ultrastructure and elemental distribution of the enameloid using high-resolution scanning electron microscopy (SEM) 106 107 and qualitative energy-dispersive X-ray spectroscopy (EDX). We 108 have correlated these results with the local mechanical properties 109 obtained by nanoindentation for the shiny layer and fluoroapatite crystallite bundles indented in different directions on both axial 110 and orthogonal planes of the teeth. The results provide valuable 111 112 information that can be used as inspiration for the design of bio-113 inspired dental materials which exploit the advantages of natural 114 shark tooth enameloid.

115 2. Materials and methods

116 *2.1. Sample preparation and analytical methods*

117 Individual teeth of the recent shark species I. oxyrinchus were 118 purchased from commercial sources that obtain their material 119 from a commercial fishery where the teeth were extracted, cleaned 120 and shipped to the retailers. Since the original position of the teeth in the jaw could not be reproduced, we used only specimens with 121 122 shapes and sizes qualifying them as erupted mature teeth. The taxonomical determination of the shark species was verified with 123 124 the help of Dr. A. Gillis, Dalhousie University, Canada. All teeth 125 used for the experiments were stored as delivered in dry state at 126 room temperature. We used SEM to study the microstructure, 127 EDX to map the elemental composition and nanoindentation to 128 determine the local mechanical properties of the enameloid and 129 the shiny layer. The used samples were either fractured or embedded and polished in specific orientations with respect to the geom-130 etry of the tooth. 131

The fractured samples were prepared by creating notches at the 132 133 tip and the cutting edge of two different teeth using a jeweler's saw. A sharp blade was then placed into the notches and splinters 134 135 of the tooth were produced with careful hammer strokes on the back of the blade. To investigate the structure and distribution of 136 the organic matrix within the enameloid, selected sample pieces 137 138 from both teeth were superficially etched by immersion in an 139 aqueous ethylenediaminetetraacetic acid (EDTA) solution (0.1 mol l⁻¹; Waldeck, Germany) with 2.5% glutaraldehyde (Merck, 140 141 Germany) for 2 min, as described by Fabritius et al. [33]. The sam-142 ples were washed in double-distilled water (1 s) followed by 100% 143 methanol (1 s), dehydrated in an ascending series of acetone-144 water (30-50-70-90-100 vol.% acetone; 5 min each) and dried in

a Bal-Tec CPD 030 critical point dryer. The native and etched tooth 145 pieces were mounted with the exposed internal surface pointing 146 upwards onto standard aluminum SEM holders and rotary-shad-147 owed with a 4 nm thick layer of platinum in a Gatan Precision 148 Etching Coating System (PECS 682). Subsequently, the samples 149 were analyzed in a Zeiss Crossbeam XB1560 FIB-SEM at an acceler-150 ation voltage of 5 kV using a 30 µm aperture and an in-lens detec-151 tor at small working distances. Where necessary, contrast and 152 brightness of the SEM micrographs were adjusted using Photoshop 153 CS2 (Adobe Inc.). 154

Four different teeth were prepared for EDX mapping and nanoindentation testing by cutting them with a jeweler's saw. Two teeth were axially cut and embedded in a conductive phenolic resin containing carbon fibers (Polyfast, Struers) with their sagittal planes exposed. The resin was cured using a Buehler SimpliMet 3000 heated press (150 bar, 5 min heating time, 180 °C). The other two teeth were orthogonally cut exposing the cross-section of the crown at about half the length of the tooth. These samples were embedded in a one-component UV-curable methyl methacrylate resin (CEM 4000 Lightfix, Cloeren Technology GmbH, Wegburg) that was cured in a Struers UV-Box (3 min using the bottom source only followed by 6 min with bottom and top source together). The exposed sections of all four teeth were polished using a series of abrasive papers with decreasing grit sizes (220, 400, 600, 1000, 2500 and 4000; Hermes) followed by a 3 µm diamond suspension (Struers) and final polishing with a 0.1 µm silica suspension (Buehler; Saphir 320/330 instrument, ATM).

Nanoindentation measurements were conducted with a Hysi-172 tron TriboIndenter TI 900 equipped with a top-down closed-loop 173 scanner that maximizes the indent-positioning accuracy. We used 174 a Berkovich indenter (Ti 39-01, tip radius 50 nm) in displacement 175 control mode (100 nm indentation depth) and a triangular load 176 function (10 s-10 s). The area function was determined on a fused Q3 177 quartz standard from Hysitron (0.7–194 nm). On the two axially 178 polished samples, a total number of 2575 indents were set in five 179 rectangular patterns that cover the shiny layer and outer enam-180 eloid and were distributed along the tooth margin and one pattern 181 located in the central region of the enameloid. The spacing 182 between the indents was 1 µm for the first and 2 µm for the second 183 sample. On the two orthogonally polished samples, a total number 184 of 435 indents were set in two rectangular patterns placed at the 185 tooth margin that covered the shiny layer and the outer enameloid. 186 The spacing between the indents was 2 µm. The reduced elastic 187 modulus (E_{red}) and the hardness (H) were determined according 188 to the method of Oliver and Pharr [34]. 189

After indentation, the samples were coated with 4 nm of platinum (Gatan PECS 682) and the quality of the indents was inspected using SEM (Zeiss Crossbeam XB1560 FIB-SEM). Elemental maps with a pixel resolution of 512×400 were recorded using the built-in Apollo XL Silicon Drift Detector (EDAX) at an acceleration voltage of 10 kV, 120 µm aperture size and a dwell time of 250 µs. The post-processing of the maps, including background subtraction combined with peak deconvolution, was performed with the EDAX Genesis software package.

Subsequently, the indented areas of selected samples were sub-199 jected to superficial etching by the application of a droplet of aque-200 ous EDTA solution (0.1 M with 2.5 wt.% glutaraldehyde) for 1.5 h, 201 washed with distilled water and then with ethanol, recoated with 202 4 nm of platinum and inspected in the SEM using a backscattered 203 electron (BSE) detector for compositional contrast. For each indent 204 pattern, we inspected the position of every individual indentation 205 and assigned the data point either to the shiny layer, to a certain 206 type of crystallite bundle of the enameloid with known orientation 207 or to areas with exposed organic matrix, respectively. Indents that 208 were not placed on a clearly defined structure like defects in the 209 material or on exposed organic envelopes were excluded from 210

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