



## Modeling and simulation of fructooligosaccharides synthesis in a batch basket reactor



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### ABSTRACT

Fructooligosaccharides (FOS) production was carried out in a batch basket reactor with immobilized fructosyltransferase from *Rhodotorula* sp. from  $500 \times 10^3 \text{ g m}^{-3}$  of sucrose in 50 mM sodium acetate buffer at pH 6.0, 48 °C at 85 rpm and with an activity of  $22.44 \times 10^3 \text{ U m}^{-3}$ . The experimental data were well adjusted to the mathematical model for FOS production using SIMULINK® (MATLAB®). The highest regression coefficient ( $R^2 > 90\%$ ) and the lowest percentual residual standard deviation (%RSD < 4.0) and *chi*-square ( $\chi^2 < 1.0$ ) were obtained for sucrose (GF), kestose (GF<sub>2</sub>) and total FOS. The mass transfer coefficient ( $k_L$ ) was determined as  $5.6 \times 10^{-5} \text{ m h}^{-1}$  and the diffusivity ( $D_S$ ) was  $2.11 \times 10^{-11} \text{ m}^2 \text{ s}^{-1}$ . The best predicted FOS yield (after 96 h) was 60.62%, with an equivalent productivity of  $3.16 \times 10^3 \text{ g m}^{-3} \text{ h}^{-1}$ . These results reaffirm the good potential of this enzyme for industrial application and, in addition, are in conformation to other studies conducted with the same enzyme from the same and different microbial sources.

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### 1. Introduction

Fructooligosaccharides (FOS) are polymers frequently used as a functional food ingredient, as a dietary fiber or as a prebiotic, due to its health benefits such as the improvement in gastrointestinal health, amongst other effects (Gibson et al., 2004; Gibson and Roberfroid, 1995; Goldberg, 1994; Passos and Park, 2003). They are polymers of 1–3 fructosyl units (F) bounded to the β-2,1 position of sucrose (GF) and they are generally represented by: GF<sub>n</sub>, in which “n” is the polymerization degree or the number of fructosyl units linked in chain. FOS can be obtained from several plants and vegetables (Benkeblia, 2013) but their production by enzymes is far more studied, even applied, since it can avoid the use of

food substrates to obtain purified FOS. The most common enzymatically produced FOS are: 1-kestose (GF<sub>2</sub>), nystose (GF<sub>3</sub>) and 1<sup>F</sup>-fructofuranosyl-nystose (GF<sub>4</sub>) called short-chain FOS (sc-FOS); the ideal composition of each, in order to optimize its health effects, is still debatable. According to Stewart et al. (2008), longer FOS (like GF<sub>4</sub>) result in a longer permanence inside the gut and, consequently, can cause better effects; on the other hand, Suzuki et al. (2006) affirmed that the shorter molecules (like GF<sub>2</sub>) are more available to the probiotics and so, more effective. Regardless the perfect composition, it is a common sense that FOS consumption is beneficial in many aspects.

β-fructofuranosidases (EC 3.2.1.26) and fructosyltransferases (EC 2.4.1.9) are enzymes which catalyze the production of FOS from sucrose and they can be obtained from different vegetable and microbial sources; the yield for the conversion of sucrose into FOS is variable for each source but in general, fungi like *Aspergillus*, are responsible for the highest industrial yields, around 60%

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**Nomenclature**

$D_b$	Basket diameter (m)
$D_i$	Impeller diameter (m)
$d_p$	Biocatalyst particle diameter (m)
$D_S$	Diffusivity coefficient of the solute in solvent ( $\text{m}^2 \text{h}^{-1}$ )
$D_T$	Reactor diameter (m)
$F$	Fructose concentration ( $\text{g m}^{-3}$ )
$G$	Glucose concentration ( $\text{g m}^{-3}$ )
$GF$	Sucrose concentration ( $\text{g m}^{-3}$ )
$GF_0$	Sucrose feed concentration ( $\text{g m}^{-3}$ )
$GF_2$	1-Kestose concentration ( $\text{g m}^{-3}$ )
$GF_3$	Nystose concentration ( $\text{g m}^{-3}$ )
$GF_4$	1 <sup>F</sup> -Fructosylnystose concentration ( $\text{g m}^{-3}$ )
$h_b$	Basket height (m)
$H$	Liquid height (m)
$k_B$	Boltzmann constant ( $\text{g m}^2 \text{K}^{-1} \text{h}^{-2}$ )
$k_{cat}$	Catalytic constant ( $\text{h}^{-1}$ )
$K_{gk}$	Competitive inhibition constant by glucose for 1-kestose uptake ( $\text{g m}^{-3}$ )
$K_{gs}$	Competitive inhibition constant by glucose for sucrose uptake ( $\text{g m}^{-3}$ )
$K_{gn}$	Competitive inhibition constant by glucose for nystose uptake ( $\text{g m}^{-3}$ )
$k_L$	Mass transfer coefficient ( $\text{m h}^{-1}$ )
$K_m$	Saturation constant ( $\text{g m}^{-3}$ )
$K_{mk}$	Saturation constant for 1-kestose ( $\text{g m}^{-3}$ )
$K_{mnh}$	Saturation constant for nystose for the hydrolyzing reaction ( $\text{g m}^{-3}$ )
$K_{ms}$	Saturation constant for sucrose ( $\text{g m}^{-3}$ )
$K_{mn}$	Saturation constant for nystose ( $\text{g m}^{-3}$ )
$K_{snh}$	Inhibition constant for the hydrolyzing reaction for nystose uptake ( $\text{g m}^{-3}$ )
$K_{ss}$	Inhibition constant for sucrose uptake ( $\text{g m}^{-3}$ )
$N$	Agitation speed (rpm)
$n_e$	Number of experimental data
$R_o$	Solute radius (m)
$P$	Power number (N)
$Q$	Flow rate ( $\text{m}^3 \text{s}^{-1}$ )
$R^2$	Correlation coefficient
$T$	Reaction temperature ( $^{\circ}\text{C}$ , K)
$V$	Reactor volume ( $\text{m}^3$ )
$V_{in}$	Input variable
$V_L$	Liquid volume ( $\text{m}^3$ )
$V_{mk}$	Maximum transfructosylating rate for 1-kestose uptake ( $\text{g m}^{-3} \text{h}^{-1}$ )
$V_{mn}$	Maximum transfructosylating rate for nystose uptake ( $\text{g m}^{-3} \text{h}^{-1}$ )
$V_{mnh}$	Maximum hydrolyzing rate for nystose uptake ( $\text{g m}^{-3} \text{h}^{-1}$ )
$V_{ms}$	Maximum sucrose uptake rate ( $\text{g m}^{-3} \text{h}^{-1}$ )
$V_{out}$	Output variable
$V_R$	Reactor total liquid volume ( $\text{m}^3$ )
$V_S$	Solid volume ( $\text{m}^3$ )
$SF$	Sensitivity factor (%)
$y_e$	Experimental value
$y_e$	Average experimental values
$y_m$	Predicted value

**Greek letters**

$\alpha$	Liquid fraction of the reactor
$\epsilon$	Energy dissipation rate per mass unit ( $\text{m}^2 \text{h}^{-1}$ )
$\mu$	Sucrose viscosity ( $\text{g m}^{-1} \text{h}^{-1}$ )

$\nu$	Sucrose cinematic viscosity ( $\text{g m}^{-1} \text{h}^{-1}$ )
$\rho$	Sucrose density ( $\text{g m}^{-3}$ )
$\chi^2$	Chi-square

**Subscript letters**

$c$	Altered condition
$m$	Bulk phase
$s$	Surface of the biocatalyst particle; standard condition

(Yun, 1996; Yun, 1996). The complex kinetic mechanism of FOS production from sucrose by fructosyltransferases was firstly proposed by Jung et al. (1989) and Duan et al. (1994) and it fits perfectly for the most fructosyltransferases, although, other mechanisms has been proposed (Vega and Zuniga-Hansena, 2014), since the same enzyme, from different sources, can perfectly present different mechanisms.

Based on previous studies regarding the application of a partially purified extracellular fructosyltransferase from *Rhodotorula* sp. immobilized by adsorption in a solid-acid support, composed mainly by niobium, it is possible to affirm that this enzyme has a good potential for industrial application (Aguiar-Oliveira and Maugeri, 2010–2013; Aguiar-Oliveira et al., 2012). For that reason, and to complement some data obtained so far about FOS production with the free and immobilized enzyme in different bioreactors (Alvarado-Huallanco and Maugeri-Filho, 2014, 2012, 2011, 2010), this work was conducted aiming to present the mathematical modeling for FOS production, from sucrose, by the immobilized fructosyltransferase from *Rhodotorula* sp. in a batch basket reactor, operated under optimized conditions for FOS synthesis.

**2. Material and methods****2.1. Microorganism and immobilized enzyme**

*Rhodotorula* sp. (LEB-V10, Laboratory of Bioengineering Process, UNICAMP, Campinas, São Paulo, Brazil) was cultivated in a sugar cane molasses and corn steep liquor based medium, its extracellular fructosyltransferase was obtained by ethanol precipitation. Aguiar-Oliveira and Maugeri (2010) described all these methodologies in details. The same authors described the methodology for enzymatic immobilization by adsorption and crosslinking onto a solid-acid support (niobium ore); the randomly adsorbed biocatalyst obtained was submitted to a sequential step – crosslinking with glutaraldehyde and sucrose (oriented adsorption) – in order to strengthen the enzymatic immobilization.

**2.2. Determination of the enzymatic activity**

According to the fundamentals presented by Chen and Liu (1996) for the free enzyme and later adapted by Aguiar-Oliveira and Maugeri (2010) for the immobilized enzyme, one transfructosylation activity unit (U) was defined as 1  $\mu\text{mol}$  of transferred fructose per minute.

**2.3. FOS production**

The optimization of FOS production by the immobilized (randomly adsorbed) fructosyltransferase from *Rhodotorula* sp. was conducted previously by Aguiar-Oliveira et al. (2012). Based on that study, the FOS synthesis were carried out in a batch basket reactor, described below. The reaction conditions of 48  $^{\circ}\text{C}$ , 500  $\times 10^3 \text{ g m}^{-3}$  of sucrose in 50 mM sodium acetate buffer at pH 6.0 were applied,

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