## Accepted Manuscript

Title: A new "on-off-on" fluorescent sensor for cascade recognition of  $Hg^{2+}$  and  $S^{2-}$  ion in aqueous medium

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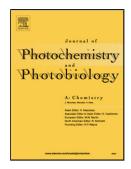
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### ACCEPTED MANUSCRIPT

# A new "on-off-on" fluorescent sensor for cascade recognition of $Hg^{2+}$ and $S^{2-}$ ion in aqueous medium

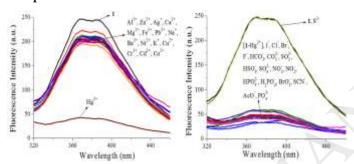
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#### **Graphical Abstract**



A new fluorescent "on-off-on" sensor for cascade recognition of Hg<sup>2+</sup> and S<sup>2-</sup> ion in 100% aqueous medium.

#### **Highlights**

- The sensor 1 we reported can cascade recognition of Hg<sup>2+</sup> and S<sup>2-</sup> ion in 100% aqueous medium.
- The fluorescent detection limits of sensor **1** for Hg<sup>2+</sup> is 0.41 nM.
- This "on-off-on" switching process could be repeated five times with little fluorescent efficiency loss.

**Abstract:** A novel  $Hg^{2+}$  ion fluorescent sensor **1** has been synthesized based on amide group, sensor **1** showed a highly sensitive and selective response towards  $Hg^{2+}$  over other metal ions by reversibly forming a [**1**- $Hg^{2+}$ ] complex in 100% aqueous solution. The fluorescence detection limits was 0.41 nM. The Job's plots, the MS analysis and  $^{1}H$  NMR titration experiments implied that there was only the formation of a  $1/Hg^{2+}$  complex with 1:1 stoichiometry. Moreover,  $S^{2-}$  ion can remove  $Hg^{2+}$  from the complex and restore the spectral signal of **1**. This research may enrich the field of multi-functional chemosensors in natural products.

Keywords: Amide group; Cascade recognition; Mercury ion; Sulfide ion

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