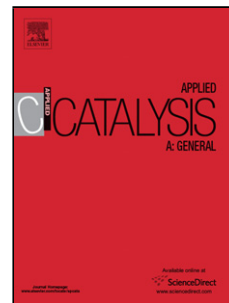


## Accepted Manuscript

Title: Rapid Room Temperature Synthesis of Tin-based Mesoporous Solids: Influence of the Particle Size on the Production of Ethyl Lactate

Authors: Nicolas Godard, Xavier Collard, Alvis Vivian, Lucia Anna Bivona, Sonia Fiorilli, Luca Fusaro, Carmela Aprile



PII: S0926-860X(18)30072-3  
DOI: <https://doi.org/10.1016/j.apcata.2018.02.014>  
Reference: APCATA 16554

To appear in: *Applied Catalysis A: General*

Received date: 8-12-2017  
Revised date: 4-2-2018  
Accepted date: 13-2-2018

Please cite this article as: Godard N, Collard X, Vivian A, Bivona LA, Fiorilli S, Fusaro L, Aprile C, Rapid Room Temperature Synthesis of Tin-based Mesoporous Solids: Influence of the Particle Size on the Production of Ethyl Lactate, *Applied Catalysis A, General* (2018), <https://doi.org/10.1016/j.apcata.2018.02.014>

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

## Rapid Room Temperature Synthesis of Tin-based Mesoporous Solids: Influence of the Particle Size on the Production of Ethyl Lactate

Nicolas Godard,<sup>a</sup> Xavier Collard,<sup>a</sup> Alvise Vivian,<sup>a</sup> Lucia Anna Bivona,<sup>a</sup> Sonia Fiorilli,<sup>b</sup> Luca Fusaro,<sup>a</sup> Carmela Aprile<sup>\*,a</sup>

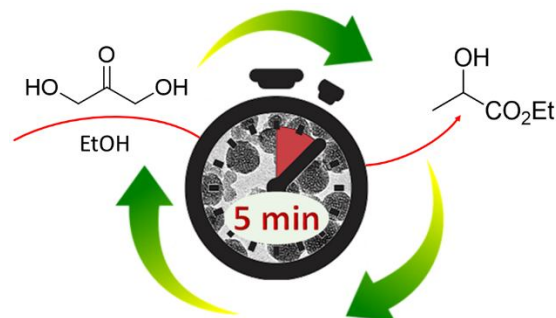
<sup>a</sup> Laboratory of Applied Material Chemistry (CMA), University of Namur, 61 rue de Bruxelles, 5000 Namur (Belgium)

<sup>b</sup> Department of Applied Science and Technology, Polytechnic of Turin, Institute of Chemistry, Corso Duca degli Abruzzi, 24, 10129 Torino, Italy

✉ Corresponding author at Laboratory of Applied Material Chemistry (CMA), University of Namur, 61 rue de Bruxelles, 5000 Namur (Belgium)

E-mail address: carmela.aprile@unamur.be

Graphical abstract



### Highlights

- A rapid room temperature synthesis protocol was successfully conceived
- A series of porous solids bearing Sn inserted as single site was prepared
- The solids were obtained with a precise control of the particle size
- An excellent correlation between particles size and catalytic activity was observed
- The best catalyst allows obtaining a full conversion of DHA with total selectivity

### Abstract

A series of tin-based mesoporous catalysts was prepared via a novel straightforward sol-gel procedure leading to an extremely short synthesis time decreased up to 5 min at room temperature. This synthesis, together with the precise control of the selected particle size, represents an advancement compared to the state of the art and can be easily applied to large scale production. Characterization of the materials revealed the presence of MCM-41 like architecture with a high specific surface area, narrow pore size distribution, insertion of tin in tetrahedral coordination and a good balance of Lewis/Brønsted acidity. Moreover, all these characteristics were almost identical for the entire series thus making these solids an ideal case study to investigate the influence of the particle size on the catalytic behavior. The Sn containing materials were tested as catalysts in the conversion of dihydroxyacetone into ethyl lactate. The excellent correlation between particles size and catalytic performances proves the importance of the size control. In addition, the absence of leaching was proved via hot filtration experiments and the materials preserved their activity in multiple catalytic cycles.

Keywords: mesoporous materials; fast synthesis; size control; heterogeneous catalysis

### 1. Introduction

Mesoporous materials containing different metal elements as single sites within the silica structure may act as highly efficient catalysts for various applications including epoxidations [1], Baeyer-Villiger oxidations [2], Meerwein-Ponndorf-Verley-Oppenauer redox reactions [3],

Download English Version:

<https://daneshyari.com/en/article/6496837>

Download Persian Version:

<https://daneshyari.com/article/6496837>

[Daneshyari.com](https://daneshyari.com)