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Vertically aligned NiS nano-flakes derived from hydrothermally prepared Ni(OH)₂ for high performance supercapacitor

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ABSTRACT

In present work, the vertically aligned NiS nano-flakes composed thin film is prepared by anionic exchange process in which hydrothermally prepared Ni(OH)₂ is used as a parent thin film and Na₂S as a sulfide ion source. This synthesis process produced fully transformed and shape-controlled nano-flakes of NiS from nano-flowers of Ni(OH)₂. The electrochemical supercapacitor properties of NiS electrode are studied with cyclic voltammetry (CV), galvanostatic charge discharge (GCD) and electrochemical impedance spectroscopy (EIS) techniques. Highly porous surface area (85 m²/g) of NiS nano-flakes makes large material contribution in electrochemical reaction stretching specific capacitance (C_s) of 880 F/g at scan rate of 5 mV/s and 90% electrochemical stability up to 4000 CV cycles in 2 M KOH electrolyte. Further, the flexible solid-state symmetric supercapacitor device (NiS/PVA–LiClO₄/NiS) has been fabricated using NiS electrodes with polyvinyl alcohol (PVA)–lithium perchlorate (LiClO₄) gel electrolyte. The NiS/PVA–LiClO₄/NiS device exhibits specific capacitance of 56 F/g with specific energy of 14.98 Wh/kg and excellent cycling stability after 2000 cycles. In addition, the NiS/PVA–LiClO₄/NiS device demonstrates illumination of red light emitting diode (LED) for 60 s, which confirms the practical applicability of NiS/PVA–LiClO₄/NiS device in energy storage.

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1. Introduction

In present day, rising demands for power sources of transitory high power density have motivated a great attention in supercapacitor with major uses in digital cameras, electric hybrid vehicles and memory back-up devices, which require higher specific power density (PD) [1]. The fast development of the worldwide economy elevates the enervation of fossil fuels as well as growing environmental pollution. There is a need of proficient, unpolluted, and supportable sources of energy and new technologies connected with energy storage [2]. Supercapacitor exhibits emerging, fascinating and substituting to battery and ordinary capacitor due to its vital properties like fast charging–discharging, higher PD and excellent electrochemical cycling stability [3]. Supercapacitor store electric charges at the interface of electrolyte and electrode. Supercapacitors can be divided in to two types on the basis of

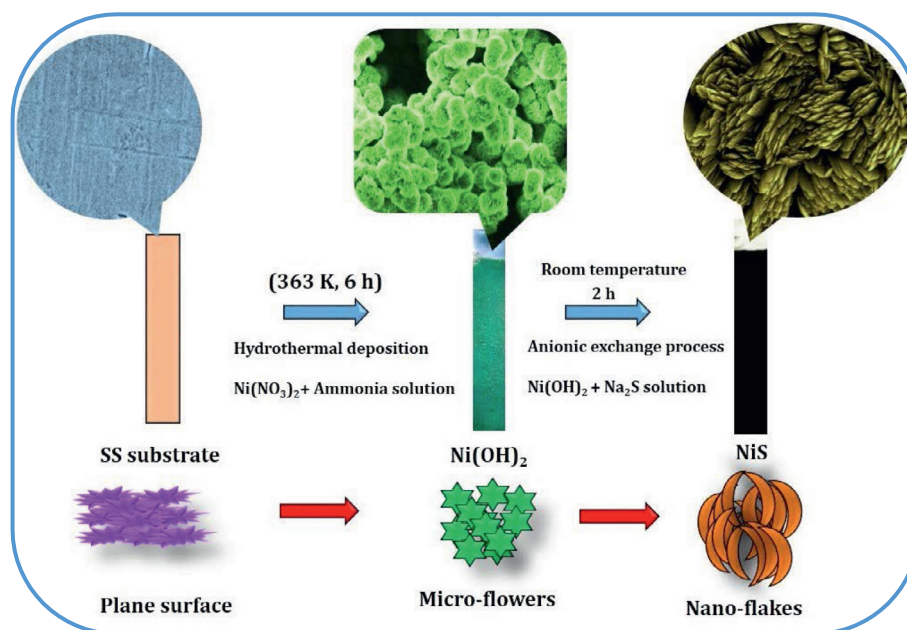
different energy-storage mechanisms as electrochemical double layer capacitor (EDLC) and pseudocapacitor, which store charges by charge separation at electrode–electrolyte and at electrode interface by faradaic charge transfer reaction, respectively. In comparison, the pseudocapacitor offers a higher specific capacitance (C_s) than EDLCs because of their fast charge-discharge faradaic reaction. Generally, carbon materials such as graphene oxide (GO), carbon nano tubes (CNT) and carbon aerogel exhibit the properties of EDLC [4,5] and metal oxides [6,7], metal sulfides [8] and conducting polymers [9] are used as a pseudocapacitive material. To overcome drawbacks like lower specific energy density (ED) and electrochemical cycling stability, a new species of hybrid capacitor is developed.

In order to improve the storing capacity of supercapacitors, there is a need of particular highly porous morphological electrodes. Accordingly, metal sulfides have much attention because of their facile preparation and excellent performance with nanostructured surface morphologies [10–12]. Al-doped β-NiS mesoporous nanoflowers show excellent energy density (36.6 Wh/kg) as well as power density (12,296 W/kg) [13]. Yan et al. [14] synthesized porous NiS nanoflake arrays by ion exchange method and achieved an energy density of 14.1 Wh/kg. Alternatively, results of current

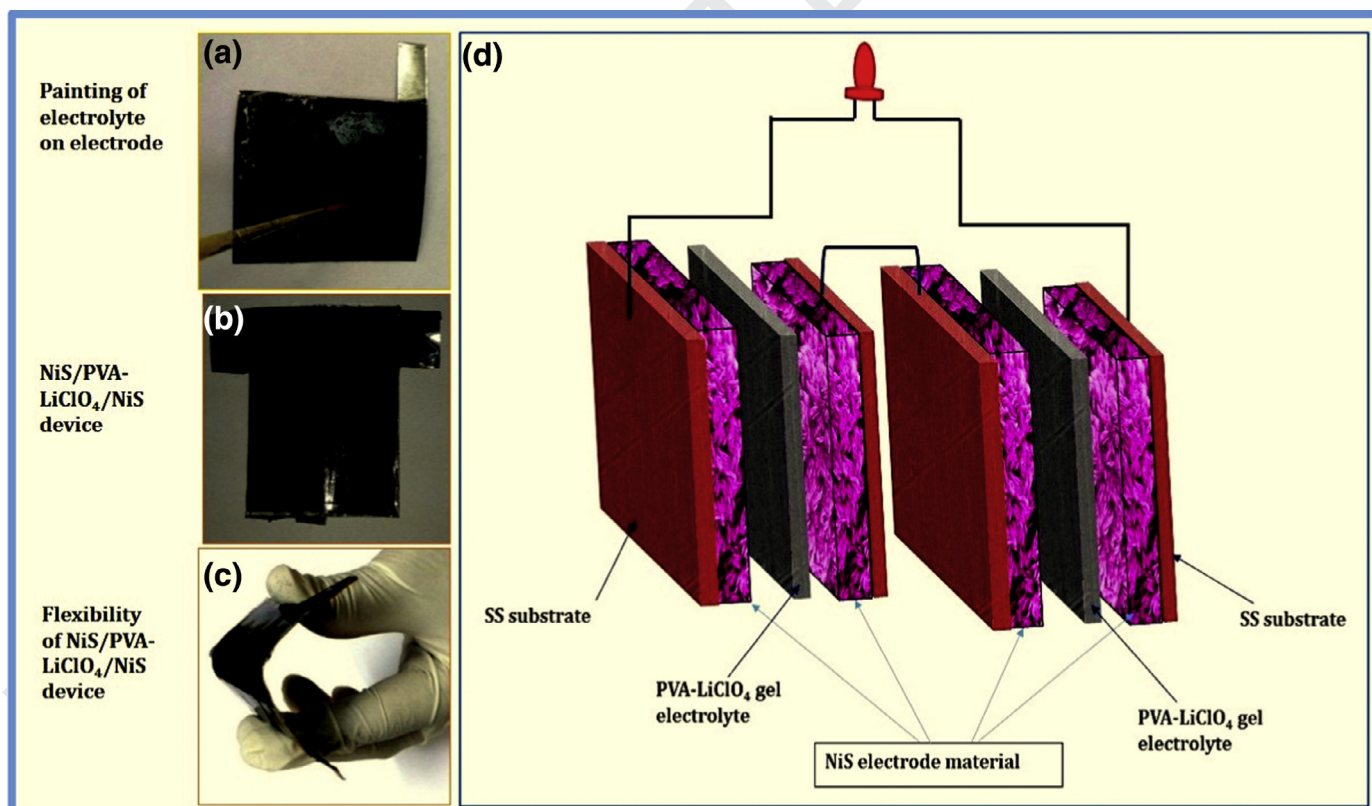
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Schematic 1. Formation of NiS nano-flakes by anionic exchange process using Ni(OH)₂ micro-flowers.



Schematic 2. (a) Painting of PVA-LiClO₄ electrolyte on NiS electrode deposited on flexible SS substrate, (b) symmetric NiS/PVA-LiClO₄/NiS device, (c) flexibility of device and (d) schematic for fabrication of NiS/PVA-LiClO₄/NiS device.

38 research indicate that metal sulfides are applicable for pseudocapacitor applications [15]. Nickel sulfide inaugurate an important
 39 type of metal sulfide having different phases such as NiS, NiS₂,
 40 Ni₃S₂, Ni₃S₄, Ni₇S₆, and Ni₉S₈ with application in dye-sensitized
 41 solar cells, supercapacitors and lithium ion batteries [16–20]. Peng
 42 et al. [21] reported C_s of 845 F/g for NiS nanoparticles synthesized
 43

by microwave-assisted method. Yang et al. [22] prepared NiS nano-
 44 rods, which exhibit C_s of 583.2 F/g. The metal hydroxide/oxide
 45 shows lower electric conductivity compared to metal sulfides.
 46 Because of lower conductivity, metal hydroxide/metal oxides have
 47 lower supercapacitor performance. Zang et al. [23] synthesized
 48 Ni(OH)₂/rGO composite by solvothermal method and reported
 49

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