



## Facile synthesis of mesoporous VO<sub>2</sub> nanocrystals by a cotton-template method and their enhanced thermochromic properties

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### ABSTRACT

As a very promising thermochromic material, VO<sub>2</sub> (M/R) (Monoclinic/Rutile) has not been widely applied in smart windows due to its intrinsic low solar modulation ( $\Delta T_{sol}$ ) and low luminous transmission ( $T_{lum}$ ). To address this issue, porous structures have been introduced into the VO<sub>2</sub> film. Herein, mesoporous VO<sub>2</sub> powders with pore size of about 2–10 nm were synthesized using cotton as template by hydrothermal methods. The pore and crystal size of the synthesized VO<sub>2</sub> powders can be reliably controlled by the hydrothermal temperature. The mesoporous VO<sub>2</sub> powders were mixed with PVP to prepare the VO<sub>2</sub>-based nanocomposite films by spin coating. The VO<sub>2</sub>-based films show a better performance between  $\Delta T_{sol}$  and  $T_{lum}$  than that appeared in previous reports. Especially, a larger pore size could lead to a higher visible transmittance and a larger crystal size would facilitate the enhancement in the solar modulation. In this sense, the VO<sub>2</sub>-based film obtained at the hydrothermal temperature of 180 °C exhibits an outstanding thermochromic performance with  $\Delta T_{sol}$  of 12.9% and  $T_{lum}$  up to 56.0% due to a larger crystal size and pore size. Therefore, this synthetic route shows a potential method for the application of mesoporous VO<sub>2</sub> powders for solar control coatings.

### 1. Introduction

Monoclinic/rutile (M/R)-phase vanadium dioxide (VO<sub>2</sub>) undergoes a fully reversible metal-semiconductor transition (MST) at a critical temperature ( $T_c$ ) of ca. 68 °C [1]. This phase transition results in dramatic changes in the optical and electrical properties [2–5]. Below the  $T_c$ , VO<sub>2</sub> displays monoclinic crystal structure, which is transparent to near-infrared light due to semiconducting property. Above the  $T_c$ , VO<sub>2</sub> displays the rutile crystal structure and is highly reflective to near infrared light while maintaining visible-light transparency due to semi-metallic property. This property could make VO<sub>2</sub> a promising material for smart windows [6–9]. However, thin films of VO<sub>2</sub> suffer from poor visible light transmission due to strong inner-band absorptions for both the metallic and semiconducting states [1,6]. In addition, the low solar modulation ( $\Delta T_{sol}$ ) of VO<sub>2</sub> also limits its application [1]. To date, a number of chemical and physical approaches have been employed to improve luminous transmission while maintaining high solar modulation [10–31], including film thickness optimization [32], doping [21–23,33,34], multilayer-stack design [27,28,35], introducing pores [28,36,37] and the formation of composite films [16,18,23,25,30,31,38]. Optical calculations suggest that VO<sub>2</sub>

nanoparticles dispersed in a dielectric host are more advantageous than VO<sub>2</sub> continuous thin solid films in smart window applications [39] because they can offer much higher luminous transmission ( $T_{lum}$ ) and enhanced  $\Delta T_{sol}$ . Liu et al. [40] prepared the composite films with VO<sub>2</sub> powders dispersed in Si-Al composite sol. The  $T_{lum}$  and  $\Delta T_{sol}$  of these composite film reached 61% and 12%, respectively. Gao et al. [30,41] prepared composite films with VO<sub>2</sub>@SiO<sub>2</sub> powders dispersed in aqueous solution containing polyurethane (PU). In this study, it is shown that the visible transparency and solar modulation ability of the VO<sub>2</sub> films can be enhanced by the addition of the PU resulting in a composite. Therefore, VO<sub>2</sub> nanoparticles distributed in another dielectric matrix can offer a novel perspective to achieve high values for  $T_{lum}$  and  $\Delta T_{sol}$ .

Inspired by this concept, highly porous VO<sub>2</sub>(M) films have been synthesized. Xie et al. prepared a periodic porous thermochromic VO<sub>2</sub>(M) film [37]. The two-dimensional periodic porous VO<sub>2</sub>(M) film exhibited high  $T_{lum}$  (81% maximum). However, the solar modulation ability was only 7.9% due to the large pore size (~200 nm). Kang et al. [39] employed polymer-assisted deposition technique to obtain a nanoporous film with an average pore size of ~28 nm, which exhibited a high  $\Delta T_{sol}$  of 14.1%. The value of  $T_{lum}$ , however, was only 41% probably

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because of the small pores. In this sense, it is a great challenge to control the pore size in the VO<sub>2</sub> film to obtain a good balance between visible transparency and solar modulation. Especially, most works are focused on the pore size between VO<sub>2</sub> nanoparticles in the film [28,36,37]. However, the effect of pore size in the VO<sub>2</sub> nanocrystals on the thermochromic properties of films based on the VO<sub>2</sub> nanocrystals has not been reported so far, to our knowledge. Therefore, it is of great importance to investigate the effect of pore size in the VO<sub>2</sub> nanocrystals on the thermochromic properties of VO<sub>2</sub> films consisted of nanocrystals.

In this work, porous VO<sub>2</sub> nanocrystals with 2–10 nm in pore size have been prepared by a facile hydrothermal method using cotton as a template. A post heat treatment under ammonia gas produced from ammonium bicarbonate (NH<sub>4</sub>HCO<sub>3</sub>) yields porous VO<sub>2</sub> powders. The crystallinity and pore size of the VO<sub>2</sub> nanocrystals appears to be controlled through the hydrothermal reaction temperature. Spin coated thin films from the VO<sub>2</sub> powders exhibit a good balance between visible transparency ( $T_{lum} = 56.0\%$ ) and solar modulation ability ( $\Delta T_{sol} = 12.9\%$ ) due to the introduction of the porous structure and creating a composite with PVP in the film preparation stage.

## 2. Experiments and characterization

### 2.1. Preparation of mesoporous VO<sub>2</sub> nanoparticles

All reagents were purchased from the Sinopharm Chemical Reagent Co., Ltd., and were used without further purification. 0.4 g (0.0022 mol) of vanadium pentoxide (V<sub>2</sub>O<sub>5</sub>) and 0.832 g (0.0066 mol) of oxalic acid were dissolved in 30 ml of water, and then 30 ml of ethylene glycol was added to increase the viscosity of the solution. As the cotton degrades the acidic solutions, 1 ml of ammonia was added to adjust the pH to *ca.* 7. The solution and 0.3 g of cotton were transferred to a 100 ml Teflon-lined stainless-steel autoclave. The autoclave was maintained at a temperature of 180 °C for 20 h and then cooled to room temperature naturally. The black product was separated and washed with water and ethanol, and then dried in air at 80 °C for 24 h. In order to remove the template to get porous powders, the product was heated, in air, at 400 °C for 1 h, ramp rate *ca.* 3 °C min<sup>-1</sup>. Finally, 0.2 g of the obtained mesoporous V<sub>2</sub>O<sub>5</sub> and 0.1 g of NH<sub>4</sub>HCO<sub>3</sub> were placed in a vacuum tube furnace for 450 °C for 30 min. Ammonia gas produced from the decomposition of NH<sub>4</sub>HCO<sub>3</sub> reduced V<sub>2</sub>O<sub>5</sub> to VO<sub>2</sub>. The obtained sample was noted as Sample d. As comparison, samples were also prepared at hydrothermal temperature of 120 °C, 140 °C and 160 °C, which are noted as Sample a, Sample b and Sample c.

### 2.2. Preparation of films

0.5 g of VO<sub>2</sub> nanoparticles were dispersed in ethanol that contained 0.25 g of PVP by grinding. This solution was then sonicated to ensure goo mixing. After centrifugation at 8000 rpm, a suspension was formed. This suspension was then uniformly cast onto a float glass substrate by spin-coating at the speed of 500 rpm for 10 s and then 1000 rpm for 20 s. Finally the ethanol was removed by drying in an oven at 80 °C for 1 min, yielding VO<sub>2</sub>-based composite films.

### 2.3. Characterization

The crystalline phases of the nanoparticles in the powder form were determined by X-ray diffraction (XRD, D8Advance, CuK $\alpha$ ,  $\lambda = 0.154178$  nm produced under a 3 kW output power). The morphologies of the powders and films were examined by a field emission scanning electron microscopy (SEM, JSM-5610LV, Japan) and field emission transmission electron microscopy (TEM, JEM2100, Japan). Pore size and distribution of the powders were determined by nitrogen adsorption–desorption using a BET analyzer. The compositions of the powders were determined by X-ray photoelectron spectroscopy (XPS). The thermochromic properties of the films from 300 to 2500 nm was

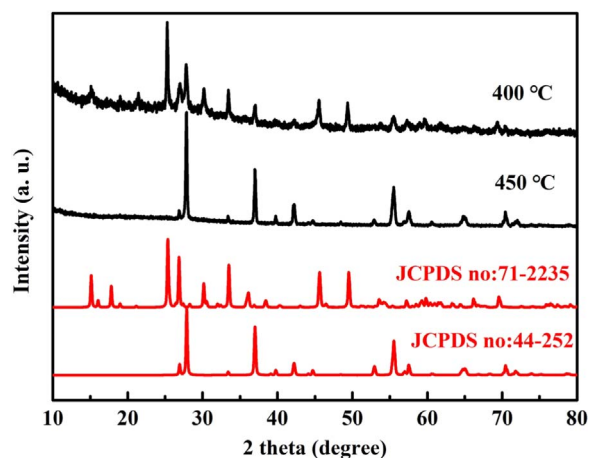


Fig. 1. XRD patterns of Sample d (prepared at the hydrothermal reaction temperature of 180 °C) obtained by annealing V<sub>2</sub>O<sub>5</sub> in an ammonia atmosphere at 400 °C and 450 °C for 30 min.

measured with ultraviolet–visible–near-infrared spectrophotometer (UV–Vis–NIR, UV-3600) at temperature 20 and 90 °C, respectively.

The integrated luminous transmittance ( $T_{lum}$ , 380–780 nm) and solar transmittance ( $T_{sol}$ , 300–500 nm) were obtained from the following equation [42]

$$T_{lum/sol} = \int \varphi_{lum/sol}(\lambda) T(\lambda) d\lambda / \int \varphi_{lum/sol}(\lambda) d\lambda$$

where,  $T(\lambda)$  denotes film transmittance at wavelength  $\lambda$ ,  $\varphi_{lum}(\lambda)$  is the standard luminous efficiency function for the photopic vision of human eyes,  $\varphi_{sol}(\lambda)$  is the solar irradiance spectrum for air mass 1.5 corresponding to the sun standing 37° above the horizon.  $\Delta T_{sol}$  is attained from  $\Delta T_{sol} = T_{sol}(20\text{ °C}) - T_{sol}(90\text{ °C})$ .

## 3. Results and discussion

### 3.1. The structure of VO<sub>2</sub> nanocrystals

The XRD patterns of the powder samples obtained by annealing the V<sub>2</sub>O<sub>5</sub> in an ammonia atmosphere at 400 °C and 450 °C for 30 min are shown in Fig. 1. In addition to the diffraction peaks corresponding to M-phase VO<sub>2</sub> (JCPDS card no.44-252), additional reflections due to V<sub>6</sub>O<sub>13</sub> (JCPDS card no.71-2235) phase were also seen for samples annealed at 400 °C. Only the M-phase VO<sub>2</sub> was detected when the annealing temperature was raised to 450 °C. This suggests that there is an energy barrier that needs to be overcome to obtain M-phase VO<sub>2</sub>.

In order to determine the morphology of the samples synthesized, SEM images of the sample d (produced from hydrothermal reaction at 180 °C and subsequently reduced at 450 °C) are shown in Fig. 2(a), (b). It can be seen from Fig. 2(a) that the samples have a fibrous morphology. The diameter of the VO<sub>2</sub> fibers is estimated to be 5–10  $\mu$ m. Interestingly, the VO<sub>2</sub> fibers consisted of clusters of small nanoparticles with pores present between the particles is seen in an enlarged image which is shown in Fig. 2(b). Further enlarged TEM images of the same sample shown in Fig. 2(c), show that the samples are clusters of nanoparticles with a diameter *ca.* 60 nm, with mesopores (about several tens of nanometers) present between the nanoparticles. This supports the morphology observed in the SEM images in Fig. 2(b). In Fig. 2(d), the lattice planes are clearly seen, these give an interplanar spacing of  $d = 0.320$  nm corresponding to the (011) plane of monoclinic VO<sub>2</sub>. This is consistent with the XRD results. In addition, the clear lattice fringe indicated good crystallinity of the synthesized VO<sub>2</sub>.

In order to examine the oxidation states of the vanadium in the synthesized VO<sub>2</sub> powders, XPS measurements were performed and the results are shown in Fig. 3. The full spectra in Fig. 3(a) suggests the presence of oxygen and vanadium in the samples. C1s is used for

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