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Effect of disjoining pressure (Π) on multi-scale modeling for evaporative liquid metal (Na) capillary



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ABSTRACT

This work presents a new multiscale model of an evaporating liquid metal capillary meniscus under nonequilibrium evaporation sustaining a nonisothermal interface. The primary investigation is elaborated on to examine the critical role of the disjoining pressure, which consists of both the traditional van der Waals component and a new electronic pressure component, for the case of liquid metals. The fully extended dispersion force is modeled along with an electronic disjoining pressure component that is unique to liquid metals attributing to their abundant free electrons. For liquid sodium (Na), as a favorable coolant for high temperature two-phase devices, the extended meniscus thin film model (sub-microscale) is coupled to a CFD model of the evaporating bulk meniscus (sub-millimeter scale). Two extreme cases are compared, i.e. with or without incorporation of the electronic disjoining pressure leads towards larger total capillary meniscus surface areas and larger net evaporative mass flow rates. Furthermore, the net evaporative mass flux in the bulk meniscus region is needed to be accounted for to obtain a true picture of the total capillary evaporation transport.

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1. Introduction

Effective cooling schemes for high power density electronics, space-based nuclear reactors, hypersonic vehicle leading edges, and the like share a need for high temperature coolant operation and mass minimization. Liquid metal heat transport devices, such as heat pipes and capillary pumped loops, have the potential to meet these needs [1–4]. Their evaporative capacity and stability, however, depends upon accurate modeling of the evaporating extended meniscus region [5]. The interline or contact line region of an evaporating extended meniscus consists of three subregions in multiscales as schematically shown in Fig. 1: the adsorbed region where a disjoining pressure dominates the local atomic forces; the intrinsic or bulk meniscus region where the interfacial curvature governs the driving physics through surface tension; and the transition or thin-film region in between where both the disjoining pressure and the interfacial curvature share a comparable influence.

Liquid alkali metals, such as sodium, present complexities in the disjoining pressure models that have received little attention in the literature. In contrast to ordinary coolants such as water or

http://dx.doi.org/10.1016/j.ijheatmasstransfer.2014.06.042 0017-9310/© 2014 Elsevier Ltd. All rights reserved. pentane, the presence of abundant free electrons in liquid metals necessitates the additional "electronic pressure" or "electron degeneracy" contributions to the disjoining pressure [6]. This study seeks to address these modeling issues associated with liquid metal in the context of a physically realistic nonisothermal model of the evaporating thin film. To do so, the model must incorporate the near-wall micro/nano-scale phenomena as well as the macro-scale heat and mass transfer effects. Furthermore, a truly multiscale modeling is necessary to couple the submicroscale heat and mass transport in the thin film region and the macroscale transport in the bulk meniscus.

Visualization of a capillary evaporation experiment will be extremely difficult, if not impossible. Sodium lamps are a mature illumination technology that provides useful insight into the detrimental effects of sodium vapor. It is known that sodium vapor reacts chemically with silica (including Pyrex glass and Quartz) to form sodium silicate [7]. As a result, the glass container is chemically etched and develops orange/brown deposits which preclude visualization [8,9]. Commercial lighting applications have developed proprietary coatings of polycrystalline alumina which resist chemical attack but at the expense of transparency [10]. Sapphire tubing appears to be the only material available that can provide transparency and resistance to sodium vapor attack at high temperatures [11]. The

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Nomencla	ature
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$a_{1,2,3}$	thin film boundary conditions	Greek Symbols	
Α	Hamaker constant [J]	α	evaporation coefficient
В	disjoining pressure electronic component constant	γ	surface tension [N/m]
	[N]	£0	permittivity of free space [s ⁴ A ² /m ³ kg]
Ca	capillary number [Ca = $\mu_l u_0 / \gamma$]	ε ₁	relative permittivity of container
C_P	specific heat capacity at constant pressure [J/kg]	£2	relative permittivity of vapor
E_F	Fermi energy [eV]	8 ₃	relative permittivity of liquid thin film
h_{fg}	latent heat of evaporation [J/kg]	η	nondimensional axial coordinate $[\eta = x/x_0]$
Н	film thickness [m]	θ	radial coordinate (cylindrical CS) [rad]
H_0	adsorbed film thickness [m]	θ	nondimensional film thickness $[\theta(\eta) = H/H_0]$
i	imaginary number $\left[\sqrt{-1}\right]$	κ	ratio of evaporative interfacial resistance to conductive
K	thin film curvature $[m^{-1}]$		resistance
m	electron mass [kg]	κ_n	electronic disjoining pressure work function parameter
me m''	evaporative mass flux $[ka/s m^2]$	λ	thermal conductivity [W/m K]
m _{evp}		μ	viscosity[N s/m ²]
M	molar mass [kg/mol]	ξ	nondimensionalized distance mapped for Chebyshev
Μ″	nondimensionalized evaporative mass flux		polynomials
Ne	valence electron number density $[m^{-3}]$	Π	disjoining pressure [N/m ²]
Р	pressure [N/m ²]	П	nondimensional disjoining pressure $[\Pi = \Pi/\Pi_0]$
a,	electron charge [C]	ho	density [kg/m ³]
a"	heat flux $[W/m^2]$	σ	optical conductivity [S/m]
Ч		τ	relaxation time [s]
r	radial coordinate (cylindrical CS) [m]	χ	electronic disjoining pressure boundary condition
R	pore radius [m]	ω	frequency [rad/s]
R	universal gas constant [N m/K mol]	ω_e	plasma frequency of an electron gas [rad/s]
S	surface domain of integration	ω_n	electromagnetic wave frequency [rad/s]
SA	surface area [m ²]		
Т	temperature [K]	Subscript	ts
T^*	nondimensional interfacial temperature	0	reference state
ΔT	liquid overheat [K]	Α	dispersion component of the disjoining pressure
и	horizontal velocity [m/s]	В	electronic component of the disjoining pressure
ν	vertical velocity [m/s]	lv	liquid/vapor interface
v_e	plasma frequency of an electron gas [Hz]	l	liquid
x	axial coordinate (Cartesian CS) [m]	tf	thin film
у	vertical coordinate (Cartesian CS) [m]	ν	vapor
Ζ	horizontal coordinate (cylindrical CS) [m]	w	wall

anisotropic nature of this crystalline material, however, makes it prone to cracks and sealing problems. Indeed, experimental visualization of liquid sodium capillary evaporation will prove difficult.

2. Background

Previous studies have addressed numerical heat and mass transfer solutions to steady extended meniscus evaporation where



Fig. 1. Schematic of a cylindrical capillary geometry identifying the distinct regions of the extended evaporating meniscus.

a static interline region continually replenished by fluid from the intrinsic meniscus. Wayner and Schonberg [12] developed a governing equation for the film height of a symmetric meniscus as a function of distance between two feed ports. Their development draws upon the pioneering thin film experiments of Derjaguin [13], Schrage's [14] relationship for net mass flux across a liquid/ vapor interface, and the evaporating extended meniscus models of Potash and Wayner [15] and Wayner et al. [16]. Later, Hallinan et al. [17] introduced new nondimensional variables and used an explicit Runge–Kutta numerical solution procedure to solve the more complicated (and realistic) case of extended meniscus evaporation under nonisothermal interface conditions.

For the case of neutral atoms and non-polar molecules, London dispersion forces act as a prevailing component of the disjoining pressure in thin films. Dzyaloshinskii, Lifshitz, and Pitaevskii (DLP) used quantum electrodynamics to derive the first general theory of van der Waals forces [18], which includes the London dispersion force. Previous studies [5,19–21] have utilized only approximations to the full DLP theory, so called Hamaker approximations that were valid as long as (a) dielectric materials were used, i.e., for non-metallic coolants with no free electrons presented, (b) the film thicknesses were much greater or less than "the wavelengths which characterize the absorptions spectra of the given bodies", and (c) a temperature requirement was met to be in a moderate range [18]. The case of a high temperature, liquid

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