Contents lists available at ScienceDirect



International Journal of Heat and Mass Transfer

journal homepage: www.elsevier.com/locate/ijhmt

Optimization of adsorption processes for climate control and thermal energy storage



Shankar Narayanan, Sungwoo Yang, Hyunho Kim, Evelyn N. Wang*

Massachusetts Institute of Technology, 77 Massachusetts Avenue, Cambridge, MA 02139, United States

ARTICLE INFO

Article history: Received 3 December 2013 Received in revised form 3 May 2014 Accepted 9 May 2014 Available online 7 June 2014

Keywords: Adsorption chiller Adsorption optimization Adsorption heat pump Thermal energy storage Adsorption dynamics HVAC

ABSTRACT

Adsorption based heat-pumps have received significant interest owing to their promise of higher efficiencies and energy savings when coupled with waste heat and solar energy compared to conventional heating and cooling systems. While adsorption systems have been widely studied through computational analysis and experiments, general design guidelines to enhance their overall performance have not been proposed. In this work, we identified conditions suitable for the maximum utilization of the adsorbent to enhance the performance of both intermittent as well as continuously operating adsorption systems. A detailed computational model was developed based on a general framework governing adsorption dynamics in a single adsorption layer and pellet. We then validated the computational analysis using experiments with a model system of zeolite 13X-water for different operating conditions. A dimensional analysis was subsequently carried out to optimize adsorption performance for any desired operating condition, which is determined by the choice of adsorbent-vapor pair, adsorption duration, operational pressure, intercrystalline porosity, adsorbent crystal size, and intracrystalline vapor diffusivity. The scaling analysis identifies the critical dimensionless parameters and provides a simple guideline to determine the most suitable geometry for the adsorbent particles. Based on this selection criterion, the computational model was used to demonstrate maximum utilization of the adsorbent for any given operational condition. By considering a wide range of parametric variations for performance optimization, these results offer important insights for designing adsorption beds for heating and cooling systems.

© 2014 Elsevier Ltd. All rights reserved.

1. Introduction

In the past three decades, the total primary energy consumption in the US has increased by $\sim 25\%$ [1], in which the buildings sector account for more than 40% of the overall consumption. In particular, within the buildings sector, heating and cooling applications account for more than half of all the energy consumed. With the significant demand for energy, especially in the building heating ventilation and air-conditioning (HVAC) systems, it is essential to develop new technologies that can improve energy efficiency.

Adsorption-based thermal energy storage, heat-pumps and airconditioning systems offer numerous advantages including the use of simple construction, reduced number of moving parts, and being environmentally friendly. A schematic diagram illustrating the operation of a basic two-bed adsorption heat pump is shown in Fig. 1. The heat pump operates by pumping a refrigerant from a storage tank or reservoir to the evaporator, where evaporation takes place by absorbing heat, which may be provided by a liquid-based heat exchanger, wherein a coolant enters the evaporator at high temperature and exits at low temperature. The vaporized refrigerant diffuses from the evaporator into one of the adsorption beds, where it is adsorbed by high-capacity adsorptive materials. A significant amount of heat is released due to spontaneous adsorption of vapor, which is dissipated via liquid-cooling as illustrated in the figure. With continuous adsorption of vapor, the bed is gradually transformed from a dry to vapor-saturated state. This can result in a cessation of any further cooling or heating using the adsorption system. Consequently, to facilitate a continuous operation, the system can utilize two beds, as shown in Fig. 1. While one of the beds is saturated with vapor, the other bed is regenerated by heating. A portion of the total heat necessary for regenerating the bed can be supplied by the other bed generating the heat of adsorption. However, additional heating is necessary for complete regeneration. The desorbed vapor is condensed within the condenser and the condensate is subsequently returned to the reservoir. In Fig. 1, the heat of condensation is also shown to be dissipated by liquid cooling. With this simple operational mechanism, adsorption-based heat pumps can provide year-round

^{*} Corresponding author. Tel.: +1 6173243311. E-mail address: enwang@mit.edu (E.N. Wang).

Nomencialure			
а, а _о	constant for Toth isotherm [mol/kg Pa]	r _s	radiusof adsorbent sample [m]
b, b_o	constant for Toth isotherm [1/Pa]	Ŕ	universal gas constant [J/mol K]
со	constant for Toth isotherm [K]	R_c	radius of a long cylinder [m]
С	concentration of vapor phase [mol/m ³]	R _s	radius of a sphere [m]
C_{b}	concentration of vapor inside the chamber [mol/m ³]	t	time [s]
C_i	initial concentration of vapor phase [mol/m ³]	to	total duration of adsorption [s]
C_{ea}	equilibrium concentration of the adsorbed phase [mol/	Т	temperature [K]
- 1	m ³]	To	reference temperature [K]
C_{μ}	concentration of adsorbed phase [mol/m ³]	T_i	initial temperature [K]
$C_{\mu b}$	boundary concentration of adsorbed phase [mol/m ³]	T_{ν}	chamber vapor temperature [K]
$C_{\mu i}$	initial concentration of adsorbed phase [mol/m ³]	T_w	chamber wall temperature [K]
C_n	specific heat []/kg K]	u	vapor velocity inside the adsorption particle [m/s]
$C_{p,c}$	specific heat of the adsorbent crystal []/kg K]	ū	dimensionless vapor velocity
$C_{n \mu}$	specific heat of the adsorbed vapor phase []/kg K]	V	volume [m ³]
$C_{n,v}$	specific heat of the vapor phase []/kg K]		
d_n	characteristic intercrystalline pore diameter [m]	Greek s	vmhols
$\dot{D_{\mu}}$	intercrystalline vapor diffusivity [m ² /s]	v	dimensionless transport parameter
$D_{\mu0}$	intercrystalline vapor diffusivity at $T_0 [m^2/s]$	Γ Γ	dimensionless vanor concentration
D_{v}^{\prime}	intracrystalline vapor diffusivity [m ² /s]	Γ	dimensionless adsorbed vapor concentration
E	constant for Toth isotherm [K]	$-\mu$ δ_m	thickness of the metal substrate [m]
E_{μ}	activation energy for vapor diffusion [J/mol]	δ	thickness or height of the adsorbent sample [m]
g, g ₀	constant for Toth isotherm	E	porosity of the adsorption bed
h_{ad}	enthalpy of adsorption [J/mol]	EUCD	porosity of hexagonally closed packed spheres
h_{v}	heat transfer coefficient during adsorption/desorption	E-	emissivity of aluminum surface
	$[W/m^2 K]$	ea Ea	emissivity of adsorbent surface
k	thermal conductivity [W/m K]	U.,	dynamic viscosity of vapor phase [kg/ms]
Κ	permeability of the adsorption bed [m ²]	ω	vapor adsorbed per unit mass of adsorbent [kg/kg]
1	characteristic length [m]	0	effective density [kg/m ³]
ls	thickness of a long slab [m]	P 0-	average density of dry adsorption bed $[kg/m^3]$
M	molecular weight of the adsorbate [kg/mol]	Pa Da	dry density of the adsorbent crystal $[kg/m^3]$
р	pressure [Pa]	σ	dimensionless parameter
Pe	Peclet number	τ	dimensionless time
p_i	initial vapor pressure [Pa]	γ	nondimensional intercrystalline pore diameter
p_{v}	chamber vapor pressure [Pa]	λ Yaua	nondimensional average intercrystalline pore diameter
$P(\chi)$	probability of a spherical void of diameter between χ	∇	gradient operator [m ⁻¹]
	and $\chi + d\chi$	$\overline{\nabla}$	dimensionless gradient operator
r _c	radius of adsorbent crystal [m]	·	operator

climate control by interfacing liquid-based heat exchangers with the system to deliver both heating and cooling.

N - --- - -1 - 4 -----

Compared to a vapor compression cycle, adsorption systems have the benefit of saving energy if the beds can be regenerated using waste heat or solar energy [2–6]. Furthermore, using novel materials with superior adsorption capacities, such as synthetic zeolites [7] and metal organic frameworks [8,9], adsorption systems have the potential to deliver large energy and power densities. The thermodynamic cycle representing the operation of an adsorption-based heat-pump is shown in Fig. 2(a), wherein Q_{evap} and Q_{cond} are the net heat exchanged in the evaporator and condenser, respectively. Meanwhile, the heat released from the bed during isobaric adsorption at p_{evap} is Q_{ads} , and the heat supplied to regenerate the bed at a constant pressure, p_{cond} is Q_{des} . During continuous operation of the system, the two beds are operated out of phase. This can be seen in Fig. 2(b), which shows the cyclic variation of the temperature of the two beds with a characteristic half-cycle or adsorption time of t_0 . It can be seen that a higher switching frequency, while ensuring complete utilization of each bed, can result in a much larger heating and cooling power for a two-bed adsorption system. While the overall energy density delivered by an adsorption-based heat pump is proportional to the adsorption capacity, the overall transport characteristics set the limit on the maximum extractable power density. Consequently, the net rate of heat exchange and vapor diffusion taking place in the characteristic time scale, t_o , is important and controls the maximum power density achievable using the adsorption-based climate control system. In addition to continuously operating multi-bed adsorption systems, thermal energy storage is also feasible using an intermittently operating, single-bed adsorption system, which also relies on efficient transport characteristics to deliver maximum power. Nevertheless, to optimize the overall performance of an adsorption-based thermal system, a detailed analysis of adsorption dynamics is necessary.

Numerical modeling in the form of thermodynamic, lumpedparametric, and full scale heat and mass transfer analysis has been extensively used in the past to predict the performance of adsorption-based heat pumps [10]. However, a thermodynamic model, based solely on first and second-law analysis, does not consider limitations due to heat transfer [11]. A lumped parametric model, although predicts adsorption dynamics, assumes uniform temperature and refrigerant concentration, and therefore neglects the intrinsic heat and mass transfer limitations [12]. A full-scale heat and mass transfer based model, while incorporating transport limitations to determine the overall performance, is complex, where the temperature-dependence of adsorption-capacity, thermophysical and transport properties are incorporated. Consequently, the non-linearity of such coupled phenomenon is generally addressed Download English Version:

https://daneshyari.com/en/article/657792

Download Persian Version:

https://daneshyari.com/article/657792

Daneshyari.com