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PII: S1385-8947(18)31430-X

DOI: https://doi.org/10.1016/j.cej.2018.07.181

Reference: CEJ 19579

To appear in: Chemical Engineering Journal

Received Date: 2 April 2018 Revised Date: 3 July 2018 Accepted Date: 27 July 2018



Please cite this article as: S. Singh, D. Bhatia, Role of NH₃ oxidation in NO_x reduction performance of dual-layer and mixed LNT-SCR configurations, *Chemical Engineering Journal* (2018), doi: https://doi.org/10.1016/j.cej. 2018.07.181

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Role of NH₃ oxidation in NO_x reduction performance of dual-layer and mixed LNT-SCR configurations

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Abstract

A global kinetic model for NH₃ oxidation is developed and incorporated into an existing LNT-SCR kinetic model to study its effect on the performance of dual-layer and mixed LNT-SCR configurations. The model predicts a decrease in the NO_x conversion and an increase in the NO_x storage with the inclusion of NH₃-oxidation reactions for both the configurations. This is due to the selective oxidation of NH₃ on the LNT catalyst instead of its consumption by the NO_x reduction reactions on the SCR catalyst. The decrease in NOx conversion is negligible at low temperatures and is more prominent at intermediate temperatures (~300 °C). The limited supply of stored NH₃ for the oxidation reactions lowers the negative effect of NH₃ oxidation at high temperatures. The fractional contribution of NH₃-oxidation reactions towards the consumption of NH₃ is calculated to be higher for the mixed configuration, which is further confirmed by the uniform axial profiles of stored NH₃ in the mixed configuration as compared to the dual-layer configuration. This results in a more prominent decrease in NO_x conversion for the mixed configuration as compared to the dual-layer configuration. The formation of NO_x at high temperatures by the oxidation of NH₃ in the combined configurations can result in a NO_x conversion even lower than the standalone LNT catalyst. A higher NH₃ slip is predicted for the mixed configuration as compared to the dual-layer configuration. It is highlighted that the composition of the LNT catalyst, its activity for NH3-oxidation and the selectivity of products

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