

## Accepted Manuscript

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PII: S1385-8947(18)30217-1  
DOI: <https://doi.org/10.1016/j.cej.2018.02.022>  
Reference: CEJ 18501

To appear in: *Chemical Engineering Journal*

Received Date: 30 November 2017  
Revised Date: 2 February 2018  
Accepted Date: 5 February 2018

Please cite this article as: Y. Shang, J. Xiao, H. Weng, F. Li, S. Cheng, S. Yamashita, Y. Muroya, M. Lin, Efficient separation of Re(VII) by radiation-induced reduction from aqueous solution, *Chemical Engineering Journal* (2018), doi: <https://doi.org/10.1016/j.cej.2018.02.022>



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# Efficient separation of Re(VII) by radiation-induced reduction from aqueous solution

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**Abstract:** Rhenium (Re) is a good nonradioactive alternative of technetium-99 (<sup>99</sup>Tc), which is a  $\beta$  emitter causing long-term radioactive toxicity due to its long half-life and high yield in nuclear reactors.  $\text{TcO}_4^-$ , the dominant form of <sup>99</sup>Tc, is very easy to migrate in the environment because of its high solubility. In this work,  $\gamma$  irradiation was applied to separate rhenium from aqueous solution in the presence of isopropanol. Re(VII) was reduced by hydrated electrons ( $e_{aq}^-$ ) generated from  $\gamma$ -radiolysis of water, leading to the formation and precipitation of Re(0) and rhenium oxides ( $\text{ReO}_2$  and  $\text{ReO}_3$ ) nanoparticles. It was the first time that Re(0) was observed in the reduction products of Re(VII) by  $\gamma$  irradiation, which was confirmed by HR-TEM, EDX, XPS and XRD. The reduction of Re(VII) by  $\gamma$  irradiation was very fast and efficient in alkaline condition, whose separation ratio reached 93.6% after 2-hour irradiation and the final separation ratio was as high as 98.1%. Increase in absorbed dose rate, isopropanol concentration, and pH, was conducive to the separation of Re overall. Further studies on the transient species by pulse radiolysis revealed that  $\text{OH}^-$  might coordinate with the intermediate Re(VI) forming different complexes with different concentrations of  $\text{OH}^-$ , thus affecting the reduction of Re(VII). This work not only demonstrates that  $\gamma$  irradiation could be a promising method for the efficient separation of Re(VII), but also gives a better understanding in the reaction mechanism.

**Key words:**  $\gamma$  irradiation; separation; rhenium(VII); technetium(VII); pulse radiolysis

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