### Accepted Manuscript

Preparation and Characterization of a Reactive Filter for Groundwater Defluoridation

N.A. Oladoja, Y. Liu, J.E. Drewes, B. Helmreich

PII: S1385-8947(15)01141-9

DOI: http://dx.doi.org/10.1016/j.cej.2015.08.059

Reference: CEJ 14051

To appear in: Chemical Engineering Journal

Received Date: 7 July 2015 Revised Date: 9 August 2015 Accepted Date: 10 August 2015



Please cite this article as: N.A. Oladoja, Y. Liu, J.E. Drewes, B. Helmreich, Preparation and Characterization of a Reactive Filter for Groundwater Defluoridation, *Chemical Engineering Journal* (2015), doi: http://dx.doi.org/10.1016/j.cej.2015.08.059

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

### **ACCEPTED MANUSCRIPT**

#### Preparation and Characterization of a Reactive Filter for Groundwater Defluoridation

## N.A. <sup>#</sup>Oladoja, Y. Liu, J. E. Drewes, B. Helmreich

Technische Universität München, Chair of Urban Water Systems Engineering, Am Coulombwall, 8, 85748 Garching, Germany

#### **Abstract**

In order to make anionic clay amenable in fixed bed operations, granular anionic clay composite was prepared by in-situ hybridization of synthetic magnesium aluminum layered double hydroxide (Mg-Al-LDH) onto sand and used for groundwater defluoridation. The Mg-Al-LDH with the best metal ion ratio composition was incorporated onto sand and the defluoridation efficiency of the granular composite was studied in batch and column reactor. Sequel to the operation parameters obtained from synthetic feed water defluoridation, the defluoridation of real groundwater (GW) was performed in the column reactor. The time-concentration profiles of the process showed that the rate of defluoridation was fast and the process was controlled by either chemisorption or ion exchange. Experimental evidences affirmed the role of anion exchange as the underlying mechanism of defluoridation. The defluoridation efficiency of the fixed bed was far higher with synthetic feed water than with GW. The spent fixed bed was regenerated with 0.5M NaOH and reused in different defluoridation cycles.

**Keywords:** Anionic clay; sand; composite material; defluoridation; groundwater; fixed-bed

\*Corresponding Author. Tel.:+491521749097

E-mail address: bioladoja@yahoo.com; n.oladoja@tum.de

Permanent Address: Department of Chemistry, Adekunle Ajasin University, Akungba-Akoko,

Nigeria

#### Download English Version:

# https://daneshyari.com/en/article/6583936

Download Persian Version:

https://daneshyari.com/article/6583936

<u>Daneshyari.com</u>