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## Acetaldehyde adsorption on TiO<sub>2</sub>: influence of NO<sub>2</sub> preliminary adsorption

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### Abstract

Titanium dioxide (TiO<sub>2</sub>) is a widespread metal oxide used in catalysis and photocatalysis, or coupled to non-thermal plasma for volatile organic compound (VOC) oxidation. Adsorption is a key step in the advanced heterogeneous oxidation processes; adsorption is partially addressed in that it is mainly investigated on fresh TiO<sub>2</sub> surfaces. However, the treatment of real effluents combines various pollutants with VOCs; among them, NO<sub>x</sub> are characterized by reactive adsorption properties on TiO<sub>2</sub> leading to irreversibly adsorbed species on the surface. The aim of this work is to determine the impact of NO<sub>2</sub> preliminary adsorption on the subsequent adsorption of a model VOC: acetaldehyde. Breakthrough methods are used to (i) characterize acetaldehyde adsorption on fresh and NO<sub>2</sub> covered TiO<sub>2</sub> surface, (ii) control the coverage of TiO<sub>2</sub> surface by irreversibly adsorbed NO<sub>x (ads)</sub> species. In a first step, acetaldehyde adsorption on TiO<sub>2</sub> has been characterized. In a second step, acetaldehyde adsorption has been achieved on TiO<sub>2</sub> surface after NO<sub>2</sub> exposure corresponding to different surface coverages. Then, acetaldehyde adsorption parameters have been determined. Subsequently, it has been possible to plot the evolution of acetaldehyde reversibly and irreversibly adsorbed quantities, and reversible fraction adsorption constant, as a function of TiO<sub>2</sub> surface coverage by NO<sub>x (ads)</sub> species, so called TiO<sub>2</sub> surface NO<sub>2</sub> ageing. Interestingly, both adsorbed acetaldehyde fractions are highly impacted by the presence of NO<sub>x (ads)</sub> species on TiO<sub>2</sub> surface. The irreversibly adsorbed fraction is considerably decreased since the lowest values of TiO<sub>2</sub> surface coverage by NO<sub>x (ads)</sub> species. Hypotheses related to competitive adsorption are not sufficient to explain the observed impact, suggesting that NO<sub>x (ads)</sub> species modify TiO<sub>2</sub> surface chemistry and acetaldehyde reactive adsorption. The reversibly adsorbed fraction of acetaldehyde is highly impacted as well; reversibly adsorbed amounts are decreased since the lowest surface coverage by NO<sub>x (ads)</sub> species. The corresponding adsorption constants are abruptly increased as soon as TiO<sub>2</sub> surface has been exposed to NO<sub>2</sub>, suggesting the formation of a new adsorption mode for acetaldehyde on NO<sub>2</sub> exposed TiO<sub>2</sub> surface. This paper evidences the considerable impact of NO<sub>2</sub> on TiO<sub>2</sub> adsorption properties regarding VOCs. Considering that NO<sub>x</sub> may accumulate on TiO<sub>2</sub> the long term behaviour of such processes should be investigated further taking into account the NO<sub>x</sub> vs. VOC interaction for adsorption.

**Keywords:** adsorption; TiO<sub>2</sub>; NO<sub>2</sub>; acetaldehyde; surface coverage



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