Accepted Manuscript

Direct visualization of reactant transport in forced convection electrochemical cells and its application to redox flow batteries



Javier Rubio-Garcia, Anthony Kucernak, Alexandra Charleson

PII:	S1388-2481(18)30163-2
DOI:	doi:10.1016/j.elecom.2018.07.002
Reference:	ELECOM 6247
To appear in:	Electrochemistry Communications
Received date:	31 May 2018
Revised date:	2 July 2018
Accepted date:	2 July 2018

Please cite this article as: Javier Rubio-Garcia, Anthony Kucernak, Alexandra Charleson, Direct visualization of reactant transport in forced convection electrochemical cells and its application to redox flow batteries. Elecom (2018), doi:10.1016/j.elecom.2018.07.002

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

ACCEPTED MANUSCRIPT

Direct visualization of reactant transport in forced convection

electrochemical cells and its application to Redox Flow Batteries

Javier Rubio-Garcia, Anthony Kucernak^{*} and Alexandra Charleson

Department of Chemistry, Imperial College London, SW7 2AZ, UK *corresponding author anthony@imperial.ac.uk

Abstract

A novel, simple and low cost electrochemiluminescence imaging method for monitoring mass transport phenomena in a redox flow battery-like system is presented. Luminol solutions were pumped through a flow field (FF) with a given design. At the flowfield/electrode interface light is emitted upon dye oxidation allowing direct visualisation of channels, U-bends and regions of poor wetting. Image analysis allows direct visualisation of reactant distribution and poor mass transport through tortuous materials. These results were compared with the experimental performance of an all-vanadium redox flow battery with different FFs as a function of flow and good correlation achieved.

1.0 Introduction

Redox Flow Batteries (RFB) are one of the most promising systems for medium to large scale energy storage due to its versatility and fast response.[1] Despite recent efforts in improving cell design,[2-4] electrode kinetics[5-7] and energy density;[8] the widespread implementation of RFBs is hindered by high capital cost. Higher electrolyte utilisation – in terms of improved state of ultimate charge (SoC) and increased local reaction rates (leading to lower stoichiometries at constant flow) - would be advantageous to improve energy density and round trip efficiency. This could be achieved with a better understanding of mass transport processes within the RFB which have traditionally been studied using mathematical models Download English Version:

https://daneshyari.com/en/article/6600683

Download Persian Version:

https://daneshyari.com/article/6600683

Daneshyari.com