Accepted Manuscript

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PII: S0013-4686(18)31785-7

DOI: 10.1016/j.electacta.2018.08.017

Reference: EA 32451

To appear in: Electrochimica Acta

Received Date: 21 May 2018

Revised Date: 24 July 2018

Accepted Date: 5 August 2018

Please cite this article as: B.J. Matsoso, B.K. Mutuma, C. Billing, K. Ranganathan, T. Lerotholi, G. Jones, N.J. Coville, The effect of N-configurations on selective detection of dopamine in the presence of uric and ascorbic acids using surfactant-free N-graphene modified ITO electrodes, *Electrochimica Acta* (2018), doi: 10.1016/j.electacta.2018.08.017.

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The effect of N-configurations on selective detection of dopamine in the presence of uric and ascorbic acids using surfactant-free N-graphene modified ITO electrodes

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Abstract

Nitrogen doped graphene based electrodes are increasingly gaining attention for application in voltammetric detection of dopamine (DA). However, little attention has been paid on the influence of the relative concentration of N-configurations on the electrocatalytic performance of these electrodes. Therefore in this study, sensitive and selective surfactantfree indium tin oxide (ITO) electrodes, modified with N-graphene films of varying nitrogen configurations, were employed for detection of DA in the presence of uric (UA) and ascorbic acids (AA). The N-graphene films, initially characterized using XPS, consisted of 38-60% pyridinic-N, 4-31% pyrrolic-N, 13-56% graphitic-N, and 2-9% oxidised pyridinic-N (NOx). The results showed that the NGr/ITO electrodes exhibited excellent selective electrocatalytic performance for the oxidation of DA in the presence of UA and AA. As a result, the detection limits (S/N = 3) of DA were determined to be $0.131 \pm 0.005 \,\mu$ M, $0.153 \pm 0.03 \,\mu$ M, and 0.645 \pm 0.07µM at the NGr-2/ITO, NGr-10/ITO and NGr-20/ITO electrodes, respectively. The excellent electrocatalytic activity of NGr-2/ITO electrode towards the oxidation of DA was ascribed mainly to an improved π - π interaction of the NOx (9.2%) and pyrrolic Nconfigurations (59.6%) with the hydroxyl and amine groups on DA. Generally, this study has provided a procedure for designing new surfactant-free 2D N-graphene based electrochemical sensors with specific electrocatalytic performance.

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