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Investigation of iron hexacyanoferrate as a high rate cathode for aqueous batteries: Sodium-ion batteries and lithium-ion batteries

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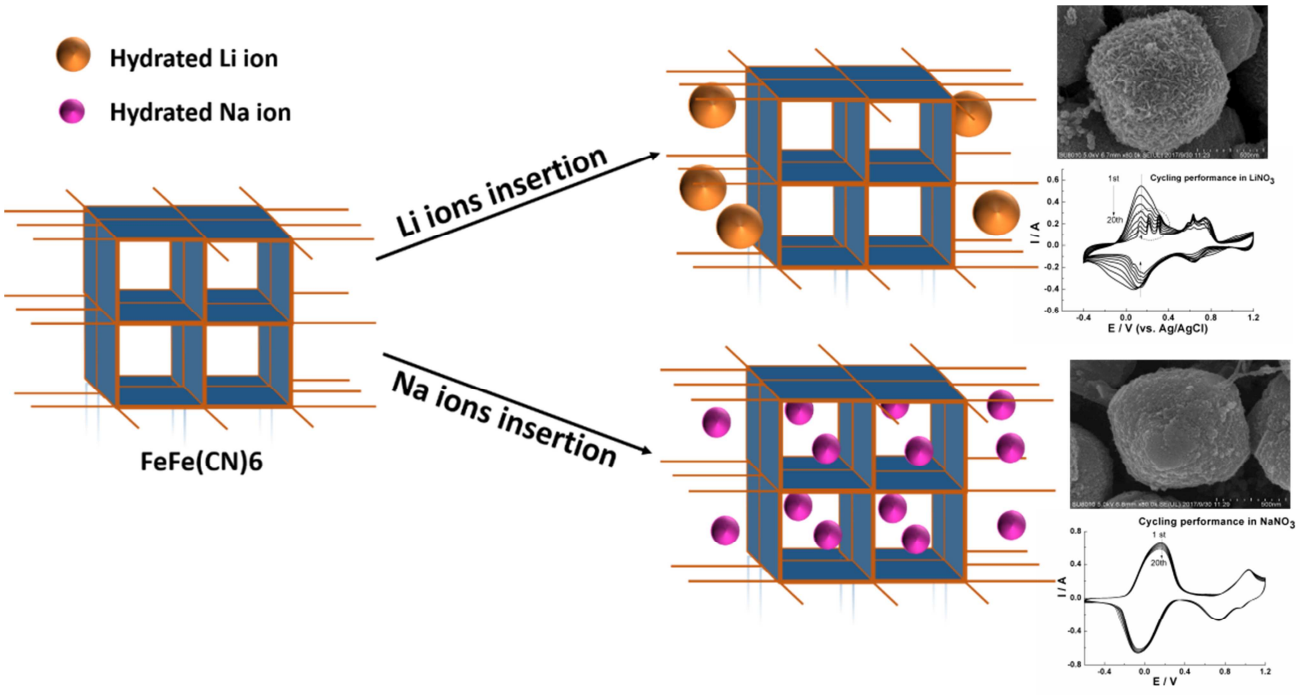
Graphical Abstract

Hydrated Na ions could insert at interstitial sites, while hydrated Li ions are more likely to occupy large open sites. This difference could lead to the better electrochemical behavior of $FeFe(CN)_6$ in storing sodium ions than storing lithium ions for aqueous batteries.



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