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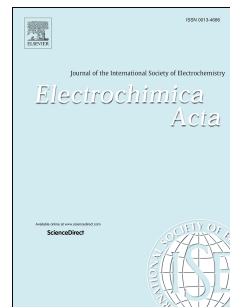
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Band-gap Narrowing and Electrochemical Properties in N-doped and Reduced Anodic TiO₂ Nanotube Arrays

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Abstract

Electrochemical activity of TiO₂ nanotube arrays (NTAs) is restricted by a wide band gap of TiO₂. To overcome this restriction, we considered systematic research on two effective methods of doping of TiO₂ NTAs such as the N-doping and electrochemical reductive doping and predicting the proper application of them. Band gap narrowing was occurred from 3.16 eV for undoped TiO₂ NTAs to 2.9 and 2.7 eV at N-doped and self-doped TiO₂ ones respectively. The electrochemical responses of the TiO₂ NTAs before and after doping were examined by cyclic Voltammetry (CV) curve. To understand the electrochemical behavior of the undoped and doped TiO₂ NTAs, electrochemical impedance spectroscopy (EIS) was used and three equivalent circuit models were also built. The results showed that the undoped TiO₂ NTAs is not strictly capacitive but a small quantity of N in TiO₂ remarkably decreases the surface resistance of TiO₂ electrode. In contrast, self-doped TiO₂ NTAs resistance is reduced to very negligible contents of 0.0001322 Ωcm^{-2} , that not only self-doped sample becomes to completely capacitive but also, it leads to the semiconductor nature of TiO₂ NTAs transforms to semi-metallic one, and the two orders of enhancement in capacitance of blue TiO₂ NTAs are very astonishing and it has outstanding potential for applications like supercapacitors as the electrochemical response of the self-doped TiO₂ NTA sample was found to be a content of about 7 mF cm⁻² that it is improved about 20 times compared with undoped one. Furthermore, it was found that doping of TiO₂ NTAs with nitrogen atoms increases the carrier density about 2.82×10^{21} and self-doped TiO₂ NTAs show the higher carrier density about 1.14×10^{25} compared with N-doped NTAs. These finding help to understand the mechanism of doping in two different methods and select the best one in relevant applications.

Keywords: electrochemical properties, cathodic polarization, blue TiO₂ nanotubes, N-doped TiO₂ nanotubes, supercapacitor

1. Introduction

Electrochemical anodization of titanium is one of the best methods to synthesis TiO₂ NTAs due to easily adjust of the characteristics of TiO₂ NTAs like the wall thickness, the tube diameter and length by varying the anodization parameters. Hence TiO₂ NTAs have attracted notable recent attention because of plain fabrication, large specific surface area and highly ordered structures [1-3]. Because of these significant features of TiO₂ NTAs, they have been extensively employed in various applications including supercapacitors [4], lithium/sodium ion batteries [5], solar cells [6], photocatalysts [7] and hydrogen production through water splitting systems [8].

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