## Accepted Manuscript

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PII: S0013-4686(17)32766-4

DOI: 10.1016/j.electacta.2017.12.192

Reference: EA 30980

To appear in: Electrochimica Acta

- Received Date: 8 September 2017
- Revised Date: 30 December 2017

Accepted Date: 31 December 2017

Please cite this article as: H. Wei, J. Chen, N. Fu, H. Chen, H. Lin, S. Han, Biomass-derived nitrogen-doped porous carbon with superior capacitive performance and high CO<sub>2</sub> capture capacity, *Electrochimica Acta* (2018), doi: 10.1016/j.electacta.2017.12.192.

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## Biomass-derived nitrogen-doped porous carbon with superior

## capacitive performance and high CO<sub>2</sub> capture capacity

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## Abstract

Nitrogen-doped porous carbon is synthesised through a low-cost approach that utilise water chestnut as a carbon source and melamine as a nitrogen source through potassium hydroxide (KOH) activation for 2 h at 600 °C–900 °C. The obtained samples exhibit predominant characteristics with highly developed micropores, an ultralarge specific surface area (3401 m<sup>2</sup> g<sup>-1</sup>) and a high nitrogen content (4.89 at.%). These characteristics endow nitrogen-doped porous carbon with a high specific capacity of 346 F g<sup>-1</sup> and a high energy density of 22.4 W h kg<sup>-1</sup> at 0.5 A g<sup>-1</sup> in 6 mol dm<sup>-3</sup> KOH. It also exhibits an excellent cycling stability with a retention of nearly 97.6% capacity after 5000 cycles at 1 A g<sup>-1</sup>. In addition, the unique pore structure and high nitrogen content of porous carbon provide an important contribution to CO<sub>2</sub> adsorption capacity, which can reach up to 6.0 mmol g<sup>-1</sup> (at 0 °C and 1 bar) and 4.7 mmol g<sup>-1</sup> (at 25 °C and 1 bar), and to high CO<sub>2</sub>/N<sub>2</sub> selectivity. Results show that the synthesised porous carbon exhibit considerable potential in electrochemical energy

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