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Effect of TiO₂ Loading on Pt-Ru Catalysts During Alcohol Electrooxidation

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Graphical Abstract

Highlights

TiO₂ can be used to modify Pt-Ru based electrodes for alcohol oxidation.

TiO₂ modified electrodes with lower amount of metals had higher active surface area

than pure Pt-Ru electrodes.

TiO₂ modified electrodes showed comparable performance with pure Pt-Ru electrode

both in a single cell and in a PEM fuel cell under alcohol fuelling.

Abstract

In this study, Pt-Ru based electrodes modified by TiO₂ were prepared by means of

thermal decomposition of chloride and isopropoxide precursors on Ti substrates, characterised

by X-ray Diffraction (XRD), Scanning Electron Microscopy (SEM), X-ray Photoelectron

Spectroscopy (XPS), electrochemical techniques and CO stripping and used as anodes for

alcohol oxidation. The minimization of the metal loading without electrocatalytic activity

losses was also explored. TiO2 was chosen due to its chemical stability, low cost and excellent

properties as substrate for metal dispersion. It was found that TiO₂ loading up to 50% results

in a 3-fold increase of the Electrochemically Active Surface (EAS). This conclusion has been

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