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Author: Bo zhao Mengyuan Chen Qingxiang Guo Yao Fu

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Electrocatalytic hydrogenation of furfural to furfuryl alcohol using platinum supported on activated carbon fibers

Bo zhao, Mengyuan Chen, Qingxiang Guo\* and Yao Fu\*

Anhui Province Key Laboratory of Biomass Clean Energy, Department of Chemistry, University of Science and

Technology of China, Hefei, Anhui 230026, P. R. China

Abstract

In this work, electrocatalytic hydrogenation (ECH) of furfural to furfuryl alcohol was studied.

Experiments were run in a H-type cell in aqueous solution with a platinum sheet as anode. Platinum

(Pt), nickel (Ni), copper (Cu) and lead (Pb) were used as cathode materials. As expected, the cathode

material, which serves as the hydrogenation catalyst, was found to have a large effect on the ECH of

furfural. Among the cathode materials studied, the Pt gave the best product selectivity. So cathodic

reduction was catalyzed by platinum supported on activated carbon fibers(Pt/ACF), a novel

electrocatalyst. Incipient wetness impregnation and electrodeposition methods were employed to

prepare the electrocatalyst. Catalysts prepared by impregnation method were more active than

prepared using electrodeposition method, presumably because of more active surface area. When

using impregnation method, 3%Pt/ACF showed the best activity and current efficiency, followed by

5%Pt/ACF. Effects of electrolyte on product yield and current efficiency were also investigated. The

yield of furfuryl alcohol was highest in the 0.1M HCl. The initial furfural concentration and the

electrolytic potential also strongly affected the product yield and current efficiency.

**Key word**: furfural; electrocatalytic hydrogenation; activated carbon fibers.

1.Introduction

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