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A new rapid method for shale oil and shale gas assessment

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GRAPHICAL ABSTRACT

- Shale gas and shale oil targets require rapid screening methods.
- Pyrolysis-FTIR has been assessed as a
 quick and information-rich
 technique.
- Application to Midland Valley shale
 samples demonstrates efficacy as
 analytical survey tool.
- Data reveal actual and potential
 liquid and gas contents.
- Quantification is possible by
- 25 utilisation of calibration curves.

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ABSTRACT

Unconventional hydrocarbons represent the future of fossil fuel supply. Arguably the most exciting unconventional deposits are those provided by shale gas and shale oil, hydrocarbons generated and retained by fine grained sedimentary rocks. Effective exploration for shale gas and shale oil requires screening of large numbers of samples in a time and cost effective manner. The most promising samples are then selected for more sophisticated and time consuming procedures. We have examined a new screening technique for shale gas and shale oil. Pyrolysis-FTIR provides a substantial amount of information related to shale quality in a single analysis including the types of gases present (including methane) and the nature of any liquid hydrocarbons released. Construction of calibration curves allows the rapid determination of gas quantities and the average chain length of aliphatic hydrocarbons present. Application of pyrolysis-FTIR to Carboniferous oil shales from the Midland Valley of Scotland reveal percentage levels of methane. Following pyrolysis at 600 °C, immature Type III kerogen containing shale has relative gas abundances in the order water > carbon dioxide > methane, mature Type I kerogen containing shales have gas abundances that follow the order water > methane > carbon dioxide and post mature Type I kerogen containing shales have relative abundances in the order carbon dioxide > water > methane. Multistep pyrolysis-FTIR reveals carbon speciation and the relative responses at low and high temperatures reflect sample maturity. The new pyrolysis-FTIR technique can provide a relatively simple and labour saving, but information-rich, technique for the assessment of shale oil and shale gas targets.

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1. Introduction

The Earth contains large amounts of hydrocarbons that are not contained within what might be called conventional reservoirs.

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These unconventional deposits are often difficult to access, more expensive to develop and require advanced technologies to make production possible [1]. With conventional hydrocarbon reserves being rapidly utilised, a future in which demand for conventional deposits outstrips supply is inevitable [2]. Hence, attention is turning to unconventional resources to satisfy our present and future oil and gas needs.

Shale gas deposits are unconventional resources in which gas has been generated from a fine grained sedimentary rock rich in organic matter and retained within that same sedimentary rock. Gas is located with the shales either within pore spaces or adsorbed onto organic or inorganic surfaces. In contrast to more conventional petroleum systems, the shales have acted as source rock, reservoir and trap concurrently [3]. Where maturity levels are less elevated, shale oil deposits may exist, where liquid hydrocarbons have been generated and then retained by the shale. In the case of shale gas and shale oil the technologies that make their recovery economical are hydraulic fracturing and horizontal drilling.

88 Although shale oil and shale gas reservoirs share many 89 similarities they also exhibit numerous differences [4]. Variations 90 are partly the results of the higher relative maturities associated 91 with shale gas reservoirs. In the higher maturity reservoirs 92 swelling clays in the mineral fraction have been transformed to 93 non-swelling clays and organic porosity and permeability is more 94 well-developed owing to the formation of pyrobitumen and char. 95 Also in the higher maturity reservoirs hydrocarbon fluids are less 96 viscous. The variations in rock permeability and fluid viscosity 97 have significant influence on the production characteristics of shale oil and shale gas. 98

99 The exploration for unconventional hydrocarbons involves 100 many of the same methods used in the search for conventional deposits. An issue that is common to both types of deposit is the 101 102 need to screen large numbers of samples in a relatively rapid and 103 inexpensive manner. Assessment techniques such as total organic 104 carbon (TOC) analysis and programmed pyrolysis provide informa-105 tion that allows the quality of the target rock to be assessed [5]. 106 The best samples are then selected for more sophisticated and time 107 consuming further analysis by techniques such as gas chromatog-108 raphy-mass spectrometry and pyrolysis-gas chromatography-109 mass spectrometry.

Organic matter is present in rocks in both low molecular weight 110 or "free" and high molecular weight or kerogen forms. Organic 111 112 matter can be liberated from the rock by various methods. Solvent extraction solvates the free compounds and isolates them 113 114 from the kerogen and mineral matrix. Thermal methods at lower 115 temperatures can also evaporate the free compounds and liberate 116 them from the immobile kerogen and mineral phases. Thermal 117 methods at higher temperatures begin to degrade the kerogen 118 and transform it into low molecular weight fragments. Once organ-119 ic units are extracted, either by solvent or thermal methods, they 120 can be detected and characterised by a number of techniques. Gas chromatography separates complex mixtures into individual 121 compounds and flame isonisation detection can quantify com-122 123 pounds while mass spectrometry can identify any unknowns.

124 Current methods for rapid assessment of the required charac-125 teristics are somewhat discrete and data must be combined to obtain an effective assessment of shale quality. A new method that 126 127 provides a variety of information on shale quality in a single ana-128 lysis would be a valuable addition to the screening techniques cur-129 rently available. Petroleum geochemists need to identify shale gas 130 and shale oil deposits that have potential commercial viability. 131 There is a requirement for a technique that is quick, inexpensive 132 and which provides indications of the rock's ability to host and 133 generate oil and gas. Moreover, a technique that can be applied 134 directly to unextracted whole rock samples would have the

advantage of analytical simplicity and the benefit of reflecting the overall constitution of the organic inventory.

A currently unexploited method for the assessment of shales is 137 pyrolysis-Fourier transform infrared (FTIR) spectroscopy. Previous 138 work has demonstrated the utility of pyrolysis-FTIR for the assess-139 ment of the gas generation potential of extra-terrestrial organic 140 materials [6-8]. Pyrolysis-FTIR has also been proposed as an explo-141 ration tool for the triage of future samples return missions aimed 142 at detecting life on Mars [9]. Thermal degradation is an analytical 143 staple of organic geochemical research but its combination with 144 FTIR is relatively uncommon. The technique can be made quantita-145 tive by constructing calibration curves for products of interest [10]. 146 In this paper we assess the utility of pyrolysis-FTIR for the rapid 147 assessment of shale gas and shale oil target rocks. We examine 148 the responses of four organic-rich shale samples of varying matu-149 rity to establish proof of concept for the method. The approach 150 swiftly provides a great deal of information, some of which would 151 normally only become available following more sophisticated and 152 time consuming analytical steps. 153

2. Methods

2.1. Samples

A series of shales of various maturities and organic constitution were collected from field exposures in the Midland Valley of Scotland (Table 1). The shales were deposited in a large rift Valley formed between the Highland Boundary and the Southern Upland faults [11]. During the Carboniferous a tropical lake complex experienced thermal stratification and associated anoxic conditions leading to the deposition of organic rich shales [12]. All shale samples were obtained from the Lower Carboniferous Subsystem, Visean Stage, Strathclyde Group sedimentary rocks. Port Edgar (PEE), South Queensferry (SQB) and Society Beach (SB) represent shales from the Queensferry Beds, Pumpherston Shale Member. Broxburn Riverside (BR) is from the Upper Oil Shale Group, Broxburn Shale Member. The stratigraphic relationships of these beds are available in the literature [12].

Comprehensive organic geochemical and microscopic investiga-170 tions have led to the definition of four organic facies in these shales 171 [12]. Organic facies 1 occurs in silty mudstones, has a mean TOC of 172 only 2% and Type III-IV gas-prone or inert kerogen indicating oxic 173 conditions that were unsuitable for preservation of planktonic 174 organic matter. Organic facies 2 is transitional between the oxic 175 facies described above and more the anoxic facies described below. 176 Organic matter contents in organic facies 2 are similar to those in 177 the oil shales but are less well preserved owing to more variable 178 redox conditions. Organic facies 3 represent the true oil shales with 179 high TOC values and Type I or Type I/II kerogens formed in a distal, 180 low energy and anoxic environment. Organic facies 4 represents 181 marine band deposition with high sulfur contents. 182

2.2. Sample preparation and screening

Each shale was crushed to a fine powder (clay-grade) using a 184 pestle and mortar. Aliquots of the powdered samples were ana-185 lysed for total organic carbon (TOC), and Rock-Eval (RE6). Any liq-186 uid hydrocarbon content of the shales was extracted by placing a 187 measured amount of crushed sample in a test tube to which a 188 93:7 v/v dichloromethane/methanol solvent mixture was added. 189 The tube was then placed in an ultrasonic bath for 15 min followed 190 by 5 min at 2500 rpm in a centrifuge. The supernatant solvent was 191 collected by pipette, the process repeated three times, and the 192 extracts combined. The final extract was subjected to a stream of 193 nitrogen gas to remove solvent and weighed when dry. The 194

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