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³ A new rapid method for shale oil and shale gas assessment

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HIGHLIGHTS

- Shale gas and shale oil targets require rapid screening methods.
- 14
 15 16 Pyrolysis-FTIR has been assessed as a 17 quick and information-rich
- 18 technique. 19 Application to Midland Valley shale 20 samples demonstrates efficacy as
- 21 analytical survey tool. 22 -Data reveal actual and potential
- 23 liquid and gas contents.
- 24 Quantification is possible by 25 utilisation of calibration curves.
- 28

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graphical abstract

ABSTRACT

Unconventional hydrocarbons represent the future of fossil fuel supply. Arguably the most exciting 44 unconventional deposits are those provided by shale gas and shale oil, hydrocarbons generated and 45 retained by fine grained sedimentary rocks. Effective exploration for shale gas and shale oil requires 46 screening of large numbers of samples in a time and cost effective manner. The most promising samples 47 are then selected for more sophisticated and time consuming procedures. We have examined a new 48 screening technique for shale gas and shale oil. Pyrolysis-FTIR provides a substantial amount of informa- 49 tion related to shale quality in a single analysis including the types of gases present (including methane) 50 and the nature of any liquid hydrocarbons released. Construction of calibration curves allows the rapid 51 determination of gas quantities and the average chain length of aliphatic hydrocarbons present. 52 Application of pyrolysis-FTIR to Carboniferous oil shales from the Midland Valley of Scotland reveal per-

centage levels of methane. Following pyrolysis at 600 °C, immature Type III kerogen containing shale has 54 centage levels of methane. Following pyrolysis at 600 °C, immature Type III kerogen containing shale has 54 relative gas abundances in the order water > carbon dioxide > methane, mature Type I kerogen contain- 55 ing shales have gas abundances that follow the order water > methane > carbon dioxide and post mature 56 Type I kerogen containing shales have relative abundances in the order carbon diox- 57 ide > water > methane. Multistep pyrolysis-FTIR reveals carbon speciation and the relative responses at 58 low and high temperatures reflect sample maturity. The new pyrolysis-FTIR technique can provide a 59 relatively simple and labour saving, but information-rich, technique for the assessment of shale oil and 60 shale gas targets. 61

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1. Introduction 66

The Earth contains large amounts of hydrocarbons that are not 67 contained within what might be called conventional reservoirs. 68

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2 M.C. Wright et al. / Fuel xxx (2015) xxx–xxx

 These unconventional deposits are often difficult to access, more expensive to develop and require advanced technologies to make production possible [\[1\]](#page--1-0). With conventional hydrocarbon reserves being rapidly utilised, a future in which demand for conventional 73 deposits outstrips supply is inevitable $[2]$. Hence, attention is turn- ing to unconventional resources to satisfy our present and future oil and gas needs.

 Shale gas deposits are unconventional resources in which gas has been generated from a fine grained sedimentary rock rich in organic matter and retained within that same sedimentary rock. Gas is located with the shales either within pore spaces or adsorbed onto organic or inorganic surfaces. In contrast to more conventional petroleum systems, the shales have acted as source 82 rock, reservoir and trap concurrently [\[3\].](#page--1-0) Where maturity levels are less elevated, shale oil deposits may exist, where liquid hydro- carbons have been generated and then retained by the shale. In the case of shale gas and shale oil the technologies that make their recovery economical are hydraulic fracturing and horizontal drilling.

 Although shale oil and shale gas reservoirs share many similarities they also exhibit numerous differences [\[4\]](#page--1-0). Variations are partly the results of the higher relative maturities associated with shale gas reservoirs. In the higher maturity reservoirs swelling clays in the mineral fraction have been transformed to non-swelling clays and organic porosity and permeability is more well-developed owing to the formation of pyrobitumen and char. Also in the higher maturity reservoirs hydrocarbon fluids are less viscous. The variations in rock permeability and fluid viscosity have significant influence on the production characteristics of shale oil and shale gas.

 The exploration for unconventional hydrocarbons involves many of the same methods used in the search for conventional deposits. An issue that is common to both types of deposit is the need to screen large numbers of samples in a relatively rapid and inexpensive manner. Assessment techniques such as total organic carbon (TOC) analysis and programmed pyrolysis provide informa-105 tion that allows the quality of the target rock to be assessed [\[5\].](#page--1-0) The best samples are then selected for more sophisticated and time consuming further analysis by techniques such as gas chromatog- raphy–mass spectrometry and pyrolysis-gas chromatography– mass spectrometry.

 Organic matter is present in rocks in both low molecular weight or ''free'' and high molecular weight or kerogen forms. Organic matter can be liberated from the rock by various methods. Solvent extraction solvates the free compounds and isolates them from the kerogen and mineral matrix. Thermal methods at lower temperatures can also evaporate the free compounds and liberate them from the immobile kerogen and mineral phases. Thermal methods at higher temperatures begin to degrade the kerogen and transform it into low molecular weight fragments. Once organ- ic units are extracted, either by solvent or thermal methods, they can be detected and characterised by a number of techniques. Gas chromatography separates complex mixtures into individual compounds and flame isonisation detection can quantify com-pounds while mass spectrometry can identify any unknowns.

 Current methods for rapid assessment of the required charac- teristics are somewhat discrete and data must be combined to obtain an effective assessment of shale quality. A new method that provides a variety of information on shale quality in a single ana- lysis would be a valuable addition to the screening techniques cur- rently available. Petroleum geochemists need to identify shale gas and shale oil deposits that have potential commercial viability. There is a requirement for a technique that is quick, inexpensive and which provides indications of the rock's ability to host and generate oil and gas. Moreover, a technique that can be applied directly to unextracted whole rock samples would have the

advantage of analytical simplicity and the benefit of reflecting 135 the overall constitution of the organic inventory. 136

A currently unexploited method for the assessment of shales is 137 pyrolysis-Fourier transform infrared (FTIR) spectroscopy. Previous 138 work has demonstrated the utility of pyrolysis-FTIR for the assess-
139 ment of the gas generation potential of extra-terrestrial organic 140 materials $[6-8]$. Pyrolysis-FTIR has also been proposed as an explo- 141 ration tool for the triage of future samples return missions aimed 142 at detecting life on Mars $[9]$. Thermal degradation is an analytical 143 staple of organic geochemical research but its combination with 144 FTIR is relatively uncommon. The technique can be made quantita- 145 tive by constructing calibration curves for products of interest $[10]$. 146 In this paper we assess the utility of pyrolysis-FTIR for the rapid 147 assessment of shale gas and shale oil target rocks. We examine 148 the responses of four organic-rich shale samples of varying matu- 149 rity to establish proof of concept for the method. The approach 150 swiftly provides a great deal of information, some of which would 151 normally only become available following more sophisticated and 152 time consuming analytical steps. 153

2. Methods 154

2.1. Samples 155

A series of shales of various maturities and organic constitution 156 were collected from field exposures in the Midland Valley of 157 Scotland [\(Table 1](#page--1-0)). The shales were deposited in a large rift 158 Valley formed between the Highland Boundary and the Southern 159 Upland faults [\[11\]](#page--1-0). During the Carboniferous a tropical lake com-
160 plex experienced thermal stratification and associated anoxic con- 161 ditions leading to the deposition of organic rich shales $[12]$. All 162 shale samples were obtained from the Lower Carboniferous 163 Subsystem, Visean Stage, Strathclyde Group sedimentary rocks. 164 Port Edgar (PEE), South Queensferry (SQB) and Society Beach (SB) 165 represent shales from the Queensferry Beds, Pumpherston Shale 166 Member. Broxburn Riverside (BR) is from the Upper Oil Shale 167 Group, Broxburn Shale Member. The stratigraphic relationships 168 of these beds are available in the literature [\[12\].](#page--1-0) 169

Comprehensive organic geochemical and microscopic investiga- 170 tions have led to the definition of four organic facies in these shales 171 [\[12\]](#page--1-0). Organic facies 1 occurs in silty mudstones, has a mean TOC of 172 only 2% and Type III–IV gas-prone or inert kerogen indicating oxic 173 conditions that were unsuitable for preservation of planktonic 174 organic matter. Organic facies 2 is transitional between the oxic 175 facies described above and more the anoxic facies described below. 176 Organic matter contents in organic facies 2 are similar to those in 177 the oil shales but are less well preserved owing to more variable 178 redox conditions. Organic facies 3 represent the true oil shales with 179 high TOC values and Type I or Type I/II kerogens formed in a distal, 180 low energy and anoxic environment. Organic facies 4 represents 181 marine band deposition with high sulfur contents. 182

2.2. Sample preparation and screening 183

Each shale was crushed to a fine powder (clay-grade) using a 184 pestle and mortar. Aliquots of the powdered samples were ana- 185 lysed for total organic carbon (TOC), and Rock–Eval (RE6). Any liq- 186 uid hydrocarbon content of the shales was extracted by placing a 187 measured amount of crushed sample in a test tube to which a 188 93:7 v/v dichloromethane/methanol solvent mixture was added. 189 The tube was then placed in an ultrasonic bath for 15 min followed 190 by 5 min at 2500 rpm in a centrifuge. The supernatant solvent was 191 collected by pipette, the process repeated three times, and the 192 extracts combined. The final extract was subjected to a stream of 193 nitrogen gas to remove solvent and weighed when dry. The 194

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