Accepted Manuscript

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PII: DOI: Reference:	S0021-9614(18)30257-X https://doi.org/10.1016/j.jct.2018.03.030 YJCHT 5375
To appear in:	J. Chem. Thermodynamics
Received Date:	17 January 2018
Revised Date:	29 March 2018
Accepted Date:	30 March 2018



Please cite this article as: N. Ebrahimi, B. Farahbod, R. Sadeghi, Salting-in and salting-out effects of organic and inorganic ammonium salts on the aqueous polymer solutions, *J. Chem. Thermodynamics* (2018), doi: https://doi.org/10.1016/j.jct.2018.03.030

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Salting-in and salting-out effects of organic and inorganic ammonium salts on the aqueous polymer solutions

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ABSTRACT

This paper accomplishes a broad survey of the salting effects of different ammonium salts on the aqueous polymer solutions. In order to cover a wide range of hydrophilic and hydrophobic nature of solutes, salts including ammonium chloride (NH₄Cl), ammonium bromide (NH₄Br), tetraethyl ammonium bromide (TEAB), tetrapropyl ammonium bromide (TPAB), tetrabutyl ammonium bromide (TBAB), tetrabutyl ammonium chloride (TBAC), and dodecyl trimethyl ammonium bromide (DTAB), and polymers polyethylene glycol (PEG400 and PEG10000) and polypropylene glycol (PPG400 and PPG1000) have been used. In the first part of this work, cloud point temperatures (T_c) for binary system of PPG1000 in pure water and for ternary systems of PPG1000 in aqueous solutions of ammonium salts have been determined by visual observation. It is shown that while the inorganic ammonium salts (NH₄Cl and NH₄Br) reduce the T_C values of PPG1000 aqueous solutions (salting-out effect), the organic ammonium salts (TEAB, TPAB, TBAB, TBAC, DTAB) because of hydrocarbon portions in their structure favourably interact with PPG1000 and elevate the $T_{\rm C}$ values (salting-in effect). The values of salting coefficient and energetic parameters of clouding process have been calculated. In the second part of this work, the isopiestic equilibrium has been investigated for several ternary {polymer + ammonium salt + water} systems at 298.2 K. Three different types of constant water activity curves deviation from the linear isopiestic relation (LIR) were observed: negative deviations for systems which undergo

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