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Fabrication of polyoxometalate/GO/PDDA hybrid nanocomposite modified electrode and electrocatalysis for nitrite ion, ascorbic acid and dopamine

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ABSTRACT: Based on self-assembly and electrodeposition, a two-dimensional $\text{PMo}_{10}\text{V}_2/\text{GO}/\text{PDDA}$ hybrid nanocomposite modified glass carbon electrode (GCE) was successfully fabricated for the first time. Graphene oxide (GO), PDDA, and $\text{PMo}_{10}\text{V}_2$ have been modified on GCE surface. Analysis by SEM and FTIR, GO has provided larger specific surface area for improving electrocatalysis activity of $\text{PMo}_{10}\text{V}_2$. The electrochemical behavior and electrocatalysis of the modified electrode were studied by detailed characterization of cyclic voltammetry (CV) and amperometric (i-t) methods. The experimental results demonstrated that $\text{PMo}_{10}\text{V}_2/\text{GO}/\text{PDDA}/\text{GCE}$ has high stability, fast response, and good electrocatalytic activity towards the oxidation of nitrite ion (NO_2^-), ascorbic acid (AA), and dopamine (DA) compared to GCE, PDDA/GCE, and GO/PDDA/GCE respectively. Their concentrations are linearly proportional to the amperometric signals. The correlation coefficients of NO_2^- , AA, and DA are 0.999, 0.998, and 0.998 respectively. The detection limits were estimated to be 0.45 μM , 0.03 μM , and 0.18 μM for NO_2^- , AA, and DA with the signal to noise ratio of 3. Almost no interference effect was observed for sensing of NO_2^- , AA, and DA in the presence of each other. The proposed modified electrode was employed for the determination of the NO_2^- , AA, and DA level in human blood serum with recoveries between 97.35% and 104.2%.

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