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Enhanced discharge performance of electrolytic manganese anode

slime using calcination and pickling approach

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Abstract:

Electrolytic manganese anode slime (EMAS) is a solid waste found in the anode chamber

during the production of electrolytic manganese metal, which mainly contains high concentration

of manganese and other impurities. In EMAS, γ-MnO₂ and δ-MnO₂ turned into α-MnO₂ after

calcination, and the oxide can be dissolved by pickling. The results showed that the specific

surface area of EMAS increased from 33.11 m²·g⁻¹ to 93.53 m²·g⁻¹, and the micropore volume

increased from $0.0129~\text{cm}^3 \cdot \text{g}^{-1}$ to $0.0362~\text{cm}^3 \cdot \text{g}^{-1}$, as well as the specific capacity and the Open

Circuit Voltage (OCV) of EMAS were 249.65 mAh·g⁻¹ and 1.60 V, respectively, under the

condition of calcination temperature 400 °C, calcination time 1 h, sulfuric acid 1 mol·L⁻¹, pickling

temperature 25 °C, pickling time 0.5 h, and liquid-to-solid ratio 8:1.

Keywords: electrolytic manganese anode slime; calcination; pickling; discharge performance

1 Introduction

In recent years, China has been the top contributor in global electrolytic metal manganese

production, accounting for over 98.5% of the total world capacity in 2016 (1.17 million tons)

[1-6]. Electrolytic manganese anode slime (EMAS) is a solid waste found in the anode chamber

during the production of electrolytic manganese metal [7-10]. Producing 1 ton of electrolytic

metal manganese would generate 50–150 kilogram of EMAS using current electrolytic technology

[11,12]. In China, EMAS was hardly used as an additives in steel making, and the vast majority

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