## Accepted Manuscript

Title: Recent advances in the mechanism of detoxification of genotoxic and cytotoxic Cr VI by microbes

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PII: S2213-3437(18)30288-4

DOI: https://doi.org/10.1016/j.jece.2018.05.042

Reference: JECE 2407

To appear in:

Received date: 15-3-2018 Revised date: 20-5-2018 Accepted date: 24-5-2018

Please cite this article as: Parvaze Ahmad Wani, Javid Ahmad Wani, Shazia Wahid, Recent advances in the mechanism of detoxification of genotoxic and cytotoxic Cr VI by microbes, Journal of Environmental Chemical Engineering https://doi.org/10.1016/j.jece.2018.05.042

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Recent advances in the mechanism of detoxification of genotoxic and cytotoxic Cr (VI) by

microbes

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**Abstract** 

Chromium has different oxidation states which ranges between -2 to +6. Among all these

oxidation states, Cr (VI) and Cr (III) are stable but vary in biological, geochemical and

toxicological properties. Cr (VI) is more toxic than Cr (III), causes allergies, irritation and

respiratory disorders. When Cr (VI) is taken up by the cell, its interaction with DNA-protein

complex results in the formation of DNA-DNA cross links, and thus ultimately may cause

mutagenic and carcinogenic effects. Chromium results in the formation of ROS, its interaction

with DNA-protein complex and cellular components causes DNA alterations and arrests their

physiological functions of the cell. ROS also may causes lipid peroxidation, DNA and protein

damage and cause cancer in the cells. Chromium (VI) result in oxidative stress, genotoxicity,

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