Accepted Manuscript

Title: Adsorption of Fluoride onto activated carbon synthesized from *Manihot esculenta* biomass—Equilibrium, kinetic and thermodynamic studies

Authors: Chubaakum Pongener, Parimal Chandra Bhomick, Aola Supong, Mridushmita Baruah, Upasana Bora Sinha, Dipak Sinha

PII: S2213-3437(18)30117-9

DOI: https://doi.org/10.1016/j.jece.2018.02.045

Reference: JECE 2237

To appear in:

Received date: 2-11-2017 Revised date: 6-2-2018 Accepted date: 24-2-2018

Please cite this article as: Chubaakum Pongener, Parimal Chandra Bhomick, Aola Supong, Mridushmita Baruah, Upasana Bora Sinha, Dipak Sinha, Adsorption of Fluoride onto activated carbon synthesized from Manihot esculenta biomass—Equilibrium, kinetic and thermodynamic studies, Journal of Environmental Chemical Engineering https://doi.org/10.1016/j.jece.2018.02.045

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



ACCEPTED MANUSCRIPT

Adsorption of Fluoride onto activated carbon synthesized from *Manihot esculenta* biomass - equilibrium, kinetic and thermodynamic studies

Chubaakum Pongener, Parimal
Chandra Bhomick, Aola Supong, Mridushmita Baruah, Upasana Bora Sinha and Dipak
Sinha *

Department of Chemistry, Nagaland University, Lumami-798627, Nagaland, India

*e-mail: dipaksinha@gmail.com

Abstract

Removal of fluoride using cheap, effective activated carbon derived from biomass of *Manihot esculenta* has been investigated. To understand the properties of activated carbon, characterization using SEM, BET, total pore volume, zero point charge etc. have been studied. The effect of adsorbent dose, contact time, initial concentration and pH on adsorption has been experimentally studied using batch mode of adsorption. Freundlich model of adsorption is found to be best fitted as compared to Langmuir and Temkin model. Kinetic studies indicate that adsorption follows a pseudo-second order kinetic model. Thermodynamic process on adsorption of fluoride onto activated carbon was found to be endothermic and spontaneous. Presence of coions on fluoride removal follows the order of CO_3^2 - SO_4^2 - NO_3 -CI-Br-. Regeneration of *Manihot esculenta* activated carbon with NaOH was successfully achieved with NaOH solution upto 98.7% could be regenerated.

Key words: adsorption, fluoride, Manihot esculenta, regeneration.

1. Introduction

Fluoride is a double edged sword-like element as its presence in drinking water can be either beneficial or detrimental to human health[1]. Fluoride is beneficial as it promotes the growth and maintenance of bones and teeth if taken at the right permissible level[2,3]. Though fluoride is universally found in earth crust and rocks, we usually come in contact with fluoride through water, food, drug, cosmetics, air and other commercial sources [4]. According to World Health Organization(WHO), fluoride is considered as one of the drinking water contaminants in addition to arsenic and nitrate which cause large-scale health problems through drinking water exposure [5]. Fluoride levels in water in many countries is notably found to be beyond permissible limits (1.5mg/L according to WHO) especially in Sri Lanka, China, the Rift Valley countries in East Africa, Turkey, and parts of South Africa and India[5,6]. Long term consumption of fluoride contaminated water could lead to various health effects because it is

Download English Version:

https://daneshyari.com/en/article/6664037

Download Persian Version:

https://daneshyari.com/article/6664037

<u>Daneshyari.com</u>