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## ACCEPTED MANUSCRIPT

# Utilization of Renewable Durian Peels for Biosorption of Zinc from Wastewater

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#### **HIGHLIGHTS**

- Zinc ion removal by renewable durian peels from contaminated water is proposed.
- Biosorbent modification with HCl acid improved the Zinc biosorption capacity.
- Temkin, Langmuir and pseudo-second-order models were obeyed in batch scale.
- Biosorbent could be utilized up to 5 cycles and beyond.
- HCl is a good modification reagent and best eluent to recover Zinc from wastewater.

#### **Abstract**

Durian peel is among the renewable biomass wastes abundantly available in Malaysia. An implication of untreated biological materials for biosorption process was intensively reported, that prioritize our work towards sorbent modification. The biosorption potentials of hydrochloric acid (HCl) modified durian peels (HAMDP) for removal of Zn (II) from simulated wastewater was investigated. Characterization of HAMDP was performed by ATR-FTIR, SEM and BET. Spectroscopic studies showed the predominant contributors for Zn (II) biosorption on HAMDP is attributed to hydroxyl, carbonyl, carboxyl and amides groups. Batch adsorption studies revealed optimum conditions of pH 8, 0.5 g biosorbent dose, 4 hours contact time and reaction temperature of 313 K. Non-linear isotherm models suggested applicability of Tempkin and Langmuir models at 313 K. The Langmuir maximum adsorption capacity was 36.73 mg/g. Kinetic studies revealed

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