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Effect of interconnection position of bicarbazole-triazine type bipolar host materials on the photophysical and device performances

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ABSTRACT

Organic compounds prepared from 1- position activated 9-phenylcarbazole interconnected with diphenyltriazine functionalized carbazole were examined as bipolar host materials of phosphorescent and thermally activated delayed fluorescent organic light-emitting diodes. Interconnection position of the 9-phenylcarbazole was 1- position, and that of diphenyltriazine functionalized carbazole was changed from 1- position to 4- position. Main influence of the interconnection position was current density of the device and 2- or 3- position interconnection increased current density of the devices. All host materials derived from the 1- position activated 9-phenylcarbazole performed well as the host of phosphorescent and thermally activated delayed fluorescent organic light-emitting diodes.

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Introduction

Host materials of organic light-emitting diodes (OLEDs) play a key role of assisting light emission of emitters by transporting carriers to the emitters (charge trapping) or transferring emission energy to emitters (energy transfer) [1–4]. Although no photon is directly generated by the host materials, the host material is as important as the emitters because carrier balance and exciton formation are governed by the host material. Many works reporting host materials touched those functions of host in the emitting layer [5–8].

In many publications, bipolar host materials were of main interest to researchers based on the knowledge that bipolar character is essential for carrier balance, efficient exciton generation, and carrier injection [9–14]. Many bipolar host materials were published in the literature as exemplified in several review papers and many of them were compatible with emitters [15,16]. General chemical structure of the reported bipolar host materials was a donor–acceptor backbone structure, and the engineering of the donor and acceptor moieties offered various compounds which have the energy level, carrier mobility, and emission energy required for the host materials [17]. One of the donor–acceptor type building blocks is carbazole–triazine type backbone structure which takes advantage of hole transport character of carbazole and

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electron transport character of triazine. Fundamentally, the carbazole-triazine building block is ideal as the core structure of the bipolar host materials because of good carrier injection. carrier transport and chemical stability [18-21]. Several modification methods of the carbazole-triazine core were suggested and one example is to add an extra carbazole unit for intense hole carrying property and thermal stability. A few compounds were reported as the derivatives of the carbazole modified carbazoletriazine core and they showed satisfactory material and device results as the hosts of green phosphorescent OLEDs (PhOLEDs) [22-24]. It was reported that the properties of material and device performances were changed by engineering the substitution position of diphenyltriazine to carbazole unit by different geometrical structure and highest occupied molecular orbital (HOMO)/lowest unoccupied molecular orbital (LUMO) distribution. However, chemical structure of the hosts was not diversified and systematic exploration of the effect of chemical structure on the material and device parameters has not been reported.

Herein, we explain our systematic investigation of four carbazole modified carbazole-triazine type hosts prepared from 1- position activated 9-phenylcarbazole interconnected with diphenyltriazine functionalized carbazole. The change of the interconnect position between 1-position activated 9-phenylcarbazole and carbazole-triazine core mostly affected carrier transport properties of the host materials. The interconnection of the carbazole unit via 2- or 3- position was advantageous for good carrier transport properties and high carrier density in the emitting layer. The synthesized host materials performed as the

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host materials in the green phosphorescent and thermally
activated delayed fluorescent (TADF) devices.

58 Experimental

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⁵⁹ General information

60 Sodium tert-butoxide (NaO(t-bu)), tri-tert-butylphosphine (P 61 $(t-bu)_3$), tris(dibenzylideneacetone)dipalladium(0) (Pd₂(dba)₃), 62 tetrakis(triphenylphosphine)palladium(0) (Pd(PPh₃)₄) (P&H Tech 63 Co.), 1-bromo-9-phenyl-9H-carbazole, 2-(4,4,5,5-tetramethyl-64 1,3,2-dioxaborolan-2-yl)-9H-carbazole, 3-(4,4,5,5-tetramethyl-65 1,3,2-dioxaborolan-2-yl)-9H-carbazole, 4-(4,4,5,5-tetramethyl-66 2-chloro-4,6-1,3,2-dioxaborolan-2-yl)-9H-carbazole, and 67 diphenyl-1,3,5-triazine (JH Trade.) were used as received. Sodium 68 hydroxide (NaOH), toluene, and tetrahydrofuran (THF) (Duksan 69 Sci. Co.) were also used as received.

70 Basic analysis of the materials was similar to the analytical 71 procedure in other works [25-27]. Nuclear magnetic resonance 72 (NMR) spectra were obtained by Varian (¹H NMR: 400 MHz), Avance-73 500 (¹³C NMR: 125 MHz). Ultraviolet-visible (UV-vis) absorption 74 spectra were obtained by a Shimadzu UV-2501PC and fluorescence 75 spectra were obtained by Hitachi F-7000. Low temperature PL 76 measurement of the synthesized materials was carried out at 77 K 77 using a dilute solution of materials in tetrahydrofuran. The mass 78 spectra were obtained by a Waters Xevo TQ-S spectrometer in API 79 mode. The HOMO energy levels were measured with a cyclic 80 voltammetry (CV). CV measurement of organic materials was carried 81 out in acetonitrile solution with tetrabutylammonium perchlorate at 82 0.1 M concentration. Pt was counter electrode and Ag was used as the 83 reference electrode. Ferrocene was used as the standard material for 84 the CV measurement.

Synthesis

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86 9-Phenyl-9H,9'H-1,2'-bicarbazole (1) 1-Bromo-9-phenyl-9H-87 carbazole (1.56 g, 4.84 mmol), 2-(4,4,5,5-tetramethyl-1,3,2-dioxa-88 borolan-2-yl)-9H-carbazole (1.7 g, 5.80 mmol), NaOH (0.58 g, 89 14.5 mmol) and Pd(PPh₃)₄ (0.28 g, 0.24 mmol) were dissolved in 90 a mixed solvent of THF (40 ml) and water (20 ml) under a nitrogen 91 bubbling. The reaction mixture was continuously stirred for 3 h 92 under reflux condition and then was cooled down to room 93 temperature. At room temperature, it was diluted with ethyl 94 acetate and washed with water using a separator funnel. The ethyl 95 acetate layer was separated and was dried using anhydrous MgSO₄. 96 Evaporation of ethyl acetate after filtering off MgSO₄ produced a 97 crude product which was purified by column chromatography on 98 silica gel using an ethyl acetate/n-hexane eluent. Purified product 99 was obtained as a white powder (1.8 g, Yield 91%). 100

¹H NMR (400 MHz, DMSO): δ 10.95 (s, 1H), 8.28 (d, 2H, J=7.2 Hz), 7.97 (d, 1H, J=8.0 Hz), 7.67 (d, 1H, J=8.0 Hz), 7.39 (m, 4H), 7.30 (m, 2H), 7.17 (d, 1H, J=8.0 Hz), 7.14 (s, 1H), 7.08 (m, 3H), 6.95 (t, 2H, J=7.6 Hz), 6.85 (d, 1H, J=7.6 Hz), 6.78 (t, 1H, J=6.8 Hz).

104 9'-(4,6-Diphenyl-1,3,5-triazin-2-yl)-9-phenyl-9H,9'H-1,2'-105 bicarbazole (PCz2TCz) 9-Phenyl-9H,9'H-1,2'-bicarbazole (1.8 g, 106 2-chloro-4,6-diphenyl-1,3,5-triazine 4.42 mmol). (1.8 g. 107 6.63 mmol), NaO(t-bu) (1.3 g, 13.2 mmol), P(t-bu)₃ (50%) (0.18 g, 108 0.44 mmol) and $Pd_2(dba)_3$ (0.20 g, 0.22 mmol) were dissolved in 109 toluene (40 ml) under a nitrogen flow. After refluxing the reaction 110 mixture for 2 h, it was diluted with dichloromethane at room 111 temperature followed by washing with water. The dichloro-112 methane layer was separated, dried using anhydrous MgSO₄, and 113 evaporated in vacuo. A crude product was obtained, purified by 114 column chromatography on silica gel using an ethyl acetate/n-115 hexane eluent, and sublimed under vacuum. A white powder was 116 obtained as a final product (1.4g, Yield 52%).

¹H NMR (400 MHz, DMSO): δ 9.08 (d, 1H, J = 8.0 Hz), 9.03 (s, 1H), 8.55 (d, 4H, J = 8.0 Hz), 8.38 (d, 2H, J = 7.6 Hz), 8.16 (d, 1H, J = 7.6 Hz), 7.93 (d, 1H, J = 8.0 Hz), 7.64 (t, 1H, J = 7.6 Hz), 7.60 (m, 3H), 7.50 (t, 1H, J = 7.6 Hz), 7.44 (t, 1H, J = 7.6 Hz), 7.34 (m, 7H), 7.15 (m, 1H), 6.98 (d, 2H, J = 7.2 Hz), 6.84 (m, 2H), 6.73 (m, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 172.45, 165.18, 142.62, 139.50, 138.79, 138.60, 138.43, 138.33, 136.27, 132.83, 129.21, 128.90, 128.30, 127.66, 127.55, 126.92, 126.86, 126.45, 126.30, 125.52, 124.93, 124.43, 123.60, 123.45, 120.46, 120.31, 120.25, 119.81, 119.75, 119.61, 118.61, 118.24, 110.61. MS (API+) m/z: 640.9 [(M+H)⁺]

9-Phenyl-9H,9'H-1,3'-bicarbazole (2) Synthetic method of 2 was the same as that of 1 except the 1-bromo-9-phenyl-9H-carbazole (2.7 g, 8.38 mmol) and 3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-9H-carbazole (3.0 g, 10.1 mmol) starting materials. A beige powder was obtained as a product (2.0 g, Yield 59%).

¹H NMR (400 MHz, DMSO): δ 11.13 (s, 1H), 8.43 (d, 1H, J = 8.4 Hz), 8.32 (d, 1H, J = 7.2 Hz), 8.11 (t, 1H, J = 8.0 Hz), 7.77 (d, 1H, J = 8.4 Hz), 7.76 (d, 1H, J = 8.4 Hz), 7.63 (m, 4H), 7.45 (m, 6H), 7.33 (m, 2H), 7.23 (t, 1H, J = 7.6 Hz), 7.14 (t, 1H, J = 8.0 Hz). MS (API+) m/z: 409.9 [(M +H)⁺]

9'-(4,6-Diphenyl-1,3,5-triazin-2-yl)-9-phenyl-9H,9'H-1,3'bicarbazole (PCz3TCz) Synthetic procedure of PCz3TCz was the same as that of PCz2TCz except for the 9-phenyl-9H,9'H-1,3'bicarbazole (2.0 g, 4.91 mmol) reactant. Purification by column chromatography on silica gel using a dichloromethane/n-hexane eluent produced a white powder as a target material (1.9 g, Yield 61%).

¹H NMR (400 MHz, CDCl₃): δ 9.15 (d, 1H, J = 8.0 Hz), 8.76 (m, 5H), 8.23 (t, 2H, J = 8.0 Hz), 7.86 (d, 1H, J = 8.0 Hz), 7.69 (s, 1H), 7.64 (m, 6H), 7.58 (m, 1H), 7.47 (d, 1H, J = 7.2 Hz), 7.41 (m, 3H), 7.33 (m, 2H), 7.28 (m, 1H), 7.10 (d, 2H, J = 7.6 Hz), 6.91 (t, 2H, J = 8.0 Hz), 6.70 (t, 1H, J = 7.6 Hz). ¹³C NMR (125 MHz, DMSO): δ 171.78, 164.48, 141.87, 138.46, 138.03, 137.34, 136.68, 135.48, 133.61, 133.22, 129.18, 129.03, 128.77, 128.37, 127.96, 127.42, 127.18, 126.73, 126.36, 126.04, 125.91, 125.10, 124.38, 123.46, 122.63, 120.89, 120.41, 120.35, 120.23, 120.11, 119.57, 117.13, 116.18, 109.91. MS (API+) m/z: 640.8 [(M+H)⁺]

9-phenyl-9H,9'H-1,4'-bicarbazole (3) 1-Bromo-9-phenyl-9H-carbazole (1.5 g, 4.66 mmol) and 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-9H-carbazole (1.64 g, 9.75 mmol) were used in the same synthetic procedure of 1. The product was obtained as a beige powder (1.45 g, Yield 76%).

¹H NMR (400 MHz, DMSO): δ 11.00 (s, 1H), 8.39 (d, 1H, J=6.4 Hz), 8.31 (d, 1H, J=6.4 Hz), 7.42 (m, 3H), 7.34 (d, 1H, J=8.0 Hz), 7.27 (m, 2H), 7.17 (t, 1H, J=8.0 Hz). 7.12 (m, 2H), 6.82 (m, 4H), 6.62 (t, 1H, J=8.0 Hz), 6.46 (d, 1H, J=7.6 Hz), 6.38 (m, 1H), 5.96 (m, 1H).

9'-(4,6-Diphenyl-1,3,5-triazin-2-yl)-9-phenyl-9H,9'H-1,4'bicarbazole (PCz4TCz) 9-Phenyl-9H,9'H-1,4'-bicarbazole (1.45 g, 3.56 mmol) was a reactant in the synthetic procedure of PCz2TCz. A white powder was obtained as a target product (1.3 g, Yield 56%).

¹H NMR (400 MHz, DMSO): δ 8.99 (d, 1H, J = 7.2 Hz), 8.77 (d, 1H, J = 7.6 Hz), 8.69 (d, 4H, J = 7.6 Hz), 8.46 (d, 1H, J = 7.2 Hz), 8.34 (d, 1H, J = 7.6 Hz), 7.73 (m, 6H), 7.48 (m, 4H), 7.29 (m, 2H), 7.16 (d, 1H, J = 7.6 Hz), 6.94 (t, 1H, J = 7.6 Hz), 6.88 (m, 1H), 6.83 (d, 1H, J = 7.6 Hz), 6.79 (m, 1H), 6.74 (t, 1H, J = 7.2 Hz), 6.50 (d, 1H, J = 8.0 Hz), 6.42 (m, 1H), 6.05 (m, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 172.59, 165.31, 142.69, 139.18, 138.86, 138.26, 137.15, 136.54, 133.81, 132.96, 129.34, 129.07, 128.42, 128.05, 127.87, 127.59, 127.29, 126.90, 126.57, 126.18, 125.71, 125.21, 124.48, 124.42, 123.26, 123.09, 121.88, 120.34, 120.23, 120.13, 120.07, 117.04, 116.24, 110.35. MS (API+) m/z: 640.9 [(M+H)⁺]

Device fabrication and measurements

The phosphorescent devices were constructed on a 150 nm thick indium tin oxide (ITO) coated substrate which was cleaned

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