



Operating conditions of lead-acid batteries in the optimization of hybrid energy systems and microgrids



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HIGHLIGHTS

- Analyzing the behavior of electrochemical storage devices is a hot research topic.
- Lead-acid batteries may work under stress conditions, negatively influencing lifetime.
- GA method combined with weighted Ah ageing model is improved in this paper.
- A hybrid energy system to be installed in Zaragoza, Spain, is analyzed as case study.
- Near-optimal design in a reduced computational time is obtained.

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ABSTRACT

The promotion and deployment of storage technologies in autonomous and grid-connected systems plays a relevant part in the massive integration of renewable power sources required for the worldwide development of a sustainable society. In this regard, analyzing the behavior of electrochemical storage devices such as lead-acid batteries installed on hybrid energy systems and microgrids in terms of their lifetime and economic profitability is an important research topic. Since renewable generation is characterized by its random nature, lead-acid batteries typically work under stress conditions, which directly influence their lifetime in a negative way by increasing the net present cost. Due to the fast growing of renewable sources as a consequence of governmental policies and incentives, the number of manufacturers to be considered worldwide is becoming really high, so that optimization techniques such as genetic algorithms (GAs) are frequently used in order to consider the performance of a high number of manufacturers of wind turbines, photovoltaic panels and lead-acid batteries subject to the environmental conditions of the location under analysis to determine a cost-effective design. In this paper, GA method combined with weighted Ah ageing model is improved by including expert experiences by means of stress factors and the categorization of operating conditions, as a new contribution to earlier studies. The effectiveness of the proposed method is illustrated by analyzing a hybrid energy system to be installed in Zaragoza, Spain, resulting in a near-optimal design in a reduced computational time compared to the enumerative optimization method.

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1. Introduction

Within energy system categories, Hybrid Energy Systems (HESs) and Microgrids (MGs) have gained attention due to their

high integration of renewable resources, which, from an environmental viewpoint, is crucial for the development of a sustainable society. HESs typically operate in isolated regions with abundant natural resources [1], while MGs are provided with modern capabilities to integrate Battery Storage Systems (BSSs), distributed generation, and demand response programs, which increase the efficiency of the system, allowing for the active participation of the consumers [2]. In a general sense, these systems are mainly

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Nomenclature

List of symbols

t	index for time ($t = 1, \dots, T$) with $T = 8760$ h	k_{sp}	corrosion speed parameter
m	index for each interval of discharging-current histogram ($m = 1, \dots, M$)	n	number of poor charge cycles
s	index of each individual of the population ($s = 1, \dots, S$)	Δn	weighting factor for the number of poor charge cycles
ν	index for the generation of GA under analysis ($\nu = 1, \dots, V$)	C_P	coefficient of the increment of acid stratification
l	index for SOC operating ranges ($l = A, B, C, D, E$)	C_M	coefficient of the reduction of acid stratification
Δt	time step (Typically 1 h)	C_Z	parameter for the capacity loss due to degradation
$U_{(t)}$	battery voltage per cell at time t (A)	C_F	counter for number of full charge events
U_0	open-circuit voltage of a single cell (V)	T_F	stress factor for time between full charge (days)
$U_{(t)}^c$	corrosion voltage per cell at time t (V)	T_S	stress factor for time at low SOC (%)
U_0^c	corrosion voltage under open-circuit and fully-charged conditions (V)	D	diffusion constant (m^2/s)
U_0^G	voltage of gassing process (V)	L	amount of intervals required to fulfill 1% of Q_{thr}
U_r	reference voltage for the reduction of acid stratification (V)	K	population of GA
$I_{(t)}$	battery current per cell at time t (A) (charge: $I_{(t)} > 0$; discharge: $I_{(t)} \leq 0$)	CF	charge stress factor
$I_{(t)}^G$	gassing current at time t (A)	FL	rated battery lifetime under float operation (year)
I_0^G	normalized value of gassing current at U_0^G and T_0^G (A)	PC	stress factor for partial cycling operation
$I_{0(t)}^G$	normalized value of gassing current at time t (A)	$A - E$	SOC operating ranges
ΔI_0^G	increment rate of gassing current (A)	$H(\cdot)$	Heaviside function
$I_{(t)}^d$	discharging current at time t	Z_l	operating cycles of the battery under standard conditions
$I_{1\%}^{\max}$	stress factor for the highest discharge rate ($\times I_{10}$)	Z_D	typical value for battery height (cm)
I_r	reference current (A)	c_0^{SOC}	coefficient of SOC impact factor (function slope)
$I_{(t)}^{1th}$	discharging current at the beginning of the partial cycling operation (A)	c_{min}^{SOC}	coefficient of SOC impact factor (minimum SOC)
I_{10}	discharging current after 10 h (A)	ΔC_{max}^c	limit value of capacity loss due to corrosion process
w	intermediate variable	$\Delta C_{(t)}^c$	loss of capacity due to corrosion process
g	coefficient of variation of open-circuit voltage with SOC	$\Delta C_{(t)}^d$	capacity loss due to degradation process
C_{10}	rated capacity of the battery after 10 h (Ah)	ΔC_{max}^d	limit value of capacity loss due to degradation process
M_c/M_d	resistance of charge-transfer process	$\Delta t_{(t)}^{SOC}$	time since the last full charge (h)
$C_c/C_{(t)}^d$	normalized capacity of a single cell	$\Delta f_{(t)}^+$	factor for the increment of acid stratification
C_U	voltage parameter of gassing process (1/V)	$\Delta f_{(t)}^-$	factor for the reduction of acid stratification
C_T	temperature parameter of gassing process (1/K)	$\Delta f_{G(t)}^-$	factor for the reduction of acid stratification due to gassing
$T_{(t)}^a$	ambient temperature at time t (K)	$\Delta f_{D(t)}^-$	factor for the reduction of acid stratification due to reduction
T_0^G	temperature of gassing process (K)	$f_{(t)}^{SOC}$	SOC impact factor
$R_{(t)}^c$	internal resistance during charging process (Ω Ah)	$f_{(t)}^I$	current factor
$R_{(t)}^d$	internal resistance during discharging process (Ω Ah)	$f_{(t)}^{ST}$	factor for degree of acid stratification
$\Delta R_{(t)}$	resistance of corrosion layer (Ω Ah)	$f_{(t)}^A$	factor for total impact of acid stratification
ΔR_{max}	limit value of resistance of corrosion layer (Ω Ah)	Q_{thr}	normalized total Ah discharged per year ($\times C_{10}$)
$\Delta W_{(t)}$	corrosion layer thickness	$l_{(m)}$	interval m of discharging-current histogram
ΔW_{max}	limit value of corrosion layer thickness	$f_{(m)}$	frequency of interval m of discharging-current histogram
$DOD_{(t)}$	depth of discharge at time t	$P_{(l)}$	partial cycling factor of range l
$SOC_{(t)}$	state of charge at time t	$P_{(l)}^{min}$	inferior limit of range l
SOC_{lim}^{low}	limit value for operation at low SOC	$P_{(l)}^{max}$	superior limit of range l
SOC_{lim}^{max}	limit value between full charging and partial cycling operation	$F_{f(s)}$	value of fitness function for individual s
k_{max}^{sp}	limit value of corrosion speed parameter	$\vec{k}_{(s)}$	individual s on the population
		$k_{(s)}^1 - k_{(s)}^4$	chromosomes of individual s of GA
		$k_{min}^1 - k_{min}^4$	inferior limit for each chromosome
		$k_{max}^1 - k_{max}^4$	superior limit for each chromosome

powered by renewable sources such as wind and solar energies, a conventional generating unit that operates during a short period of time or is dispatched producing a reduced amount of energy. A BSS is typically based on lead-acid technology [3] that provides flexibility to the system by storing the excess energy to be consumed later during those periods of low renewable energy [4]. On the one hand, BSS is the weakest element of the system, as its lifetime is difficult to understand and predict. On the other hand, the variability of renewable resources and the operating philosophy imposes stress conditions on the storing system, reducing its lifetime and increasing its Net Present Cost (NPC).

This problem has been widely studied for a long time, resulting in quantitative and qualitative approaches. Quantitative approaches could be classified in three different categories: physico-chemical ageing models, weighted Ah ageing models, and event-oriented ageing models. Physico-chemical ageing models are based on detailed knowledge of the relationship between the state variables and their corresponding effects on the ageing mechanisms, specifically, the degradation of the active mass and the loss of grid conductivity as a consequence of the corrosion process. This relationship is based on the mathematical equations of chemical reactions; hence, the operating conditions can be

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