



Bio-desulfurization of biogas using acidic biotrickling filter with dissolved oxygen in step feed recirculation



Sumate Chaiprapat^{a,*}, Boonya Charnnok^b, Duangporn Kantachote^c, Shihwu Sung^d

^a Energy Technology Research Center, Department of Civil Engineering, Faculty of Engineering, Prince of Songkla University, Hat Yai Campus, Hat Yai, Songkhla 90110, Thailand

^b Faculty of Environmental Management and PSU Energy System Research Institute (PERIN), Prince of Songkla University, Hat Yai Campus, Hat Yai, Songkhla 90110, Thailand

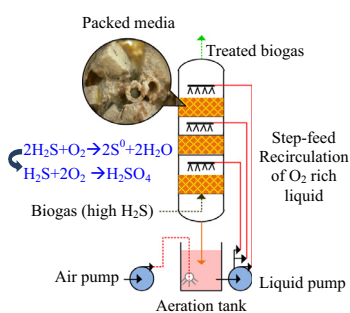
^c Department of Microbiology, Faculty of Science, Prince of Songkla University, Hat Yai Campus, Hat Yai, Songkhla 90110, Thailand

^d College of Agriculture, Forestry and Natural Resource Management, University of Hawaii at Hilo, Hilo, Hawaii 96720-4091, USA

HIGHLIGHTS

- Triple stage biotrickling filter was superior to single stage biotrickling filter.
- H₂S removal efficiency was enhanced by step feeding O₂ rich liquid recirculation.
- 1st order removal rate coefficient of T-BTF was 47% higher than S-BTF.
- Larger liquid recirculation shifted O₂/H₂S balance system to complete oxidation.
- Sulfuric acid production improved with an increasing liquid recirculation velocity.

GRAPHICAL ABSTRACT



ARTICLE INFO

Article history:

Received 20 October 2014
 Received in revised form 17 December 2014
 Accepted 20 December 2014
 Available online 27 December 2014

Keywords:

Desulfurization
 Sulfuric acid
 Dissolved oxygen
 Liquid recirculation
 Step feed

ABSTRACT

Triple stage and single stage biotrickling filters (T-BTF and S-BTF) were operated with oxygenated liquid recirculation to enhance bio-desulfurization of biogas. Empty bed retention time (EBRT 100–180 s) and liquid recirculation velocity (q 2.4–7.1 m/h) were applied. H₂S removal and sulfuric acid recovery increased with higher EBRT and q . But the highest q at 7.1 m/h induced large amount of liquid through the media, causing a reduction in bed porosity in S-BTF and H₂S removal. Equivalent performance of S-BTF and T-BTF was obtained under the lowest loading of 165 gH₂S/m³/h. In the subsequent continuous operation test, it was found that T-BTF could maintain higher H₂S elimination capacity and removal efficiency at 175.6 ± 41.6 gH₂S/m³/h and 89.0 ± 6.8% versus S-BTF at 159.9 ± 42.8 gH₂S/m³/h and 80.1 ± 10.2%, respectively. Finally, the relationship between outlet concentration and bed height was modeled. Step feeding of oxygenated liquid recirculation in multiple stages clearly demonstrated an advantage for sulfide oxidation.

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1. Introduction

Waste to energy approach is now becoming a standard practice, where various waste material feedstocks are converted to different forms of energy. Biogas from waste is among the most economical

methods to transform liquid or semi-solid wastes through anaerobic digestion to methane (CH₄), which can be used as fuel to substitute conventional sources such as natural gas and oil. Hydrogen sulfide (H₂S) is an odorous and corrosive gas that is inevitably formed with the biogas when sulfur is present in the feedstock. It has detrimental effects in environment. At 1000–3000 ppm, human exposure to H₂S can cause instantaneous death (Wang et al., 2005). Combustion of the biogas containing H₂S

* Corresponding author. Fax: +66 74 459396.

E-mail address: sumate.ch@psu.ac.th (S. Chaiprapat).

generates SO₂ which is an acid rain precursor leading to damages to vegetation, architecture and building structure. For suitable biogas utilization in power generation, H₂S level below 500 ppm is generally a recommended upper limit while the ideal H₂S-free biogas is desired in order to avoid corrosion of biogas engines and fast degradation of engine lube oil.

Since the governmental subsidy from Thailand Ministry of Energy for small renewable power generation was put in place in 1995, anaerobic digestion for industrial wastewaters had flourished due to the maturation of technology and environmental benefit it provided. So far, factories in Thailand have installed over 1145 biogas systems across various sectors (Thailand Energy Policy and Planning Office, 2012). Among, the concentrated latex and ethanol industries have shown little interest in biogas technology although their wastewaters have a potential. Their high sulfate wastewaters are often problematic through the offensive odor of H₂S emitted from waste storage ponds to surrounding communities. If anaerobic digester is used, there will be an intense H₂S production with the biogas, reaching as high as 13,000–26,000 ppm for concentrated latex industry (Chaiprapat et al., 2011; Saelee et al., 2009) and 30,000 ppm for ethanol industry (Frankel, 1986). In addition, some specific factories have also produced the biogas with high H₂S such as some seafood processing plants whose biogas containing 5000–8000 ppm as a result of alum use in the process.

Traditional technologies of gas decontamination are based on physical, chemical and thermal principles that normally incur high operating cost and generate waste to be disposed. Biological treatments are economically advantageous especially when the gas stream contains biodegradable or bio-convertible compounds (Devanny et al., 1999; Gabriel and Deshusses, 2003; Tomàs et al., 2009). In biotrickling filtration system, a subgroup of biofiltration, active microbes are immobilized onto bed material while the liquid moves downward typically in countercurrent with the gas stream. The mixture of gaseous H₂S and O₂ coming from air injection is absorbed to liquid phase then dispersed to the attached biofilm. That H₂S is biochemically oxidized to the metabolic products of elemental sulfur (S⁰) or sulfuric acid (H₂SO₄) depending on whether partial or complete oxidation is taking place (Devanny et al., 1999; Maier et al., 2009; Yamanaka, 2008). Although both reactions are carried out by sulfide oxidizing bacteria (SOB), to what degree each reaction proceeds depends on the reactor configuration and operating condition that will dictate the micro environment interphase within the filter bed. However, previous studies have mostly focused on the biodesulfurization at low concentrations (Jin et al., 2005; Tang et al., 2009). Only a few works on biofiltration were conducted at elevated H₂S concentrations, i.e. above 4000 ppmv (Fortuny et al., 2008, 2011; Rattanapan et al., 2009), but even fewer were done using real biogas with high H₂S as inlet (Chaiprapat et al., 2011; Rodriguez et al., 2014; Tomàs et al., 2009). pH of recirculating liquid, liquid recirculation rate and empty bed retention time are among the key operating parameters in biotrickling filtration influencing the biochemical H₂S oxidation performance. Because sulfuric acid is a product of H₂S oxidation, acidic biofiltration was considered suitable. Although some alkaline biotrickling filtration was proposed (González-Sánchez et al., 2008), higher operating cost to maintain pH against the acidifying state becomes a big obstacle for industrial biogas cleanup. Some sulfide oxidizing bacteria (SOB) such as *Acidithiobacillus* sp. are capable of living in extreme pH due to their ability for adaptations that involve changes to the cytoplasmic membrane controlling ion transport (McArthur, 2006). In this work, extreme acidic condition was employed since the recovery of sulfuric acid by microbial activity was aimed as a byproduct from the process.

Oxygen availability and mass transfer could become a limiting factor for biological H₂S oxidation because of its low solubility in

water, only 8.24 mg/L at 25 °C (Colt, 1984). Over supply of air, which composed of only 21% oxygen, through air mix with biogas to biotrickling filter will cause dilution of methane (Chaiprapat et al., 2011). Duangmanee et al. (2009) succeeded in control of O₂ at lower than 5% in off gas by micro-aeration for removing H₂S with partial oxidation although it was strictly chemical reaction. O₂ could be delivered in other means to avoid over injecting of air. In biofiltration, O₂ could be delivered in dissolved form (dissolved oxygen, DO) through the recirculating liquid coming in contact with the biogas stream (Rodriguez et al., 2014). The liquid recirculation velocity (q , m/h), defined as the amount of recirculating liquid applied per unit surface area of a packed bed, could play an important role in regulating such O₂ supply to the system. This method not only reduced CH₄ dilution in biogas but also supplied moisture and nutrients to the microorganisms inside the reactor, and wash the microbial metabolic products off the biofilm (Charnnok et al., 2011). However, a long travel path of biogas and recirculating liquid with single inlet point from the opposite ends cause uneven distribution of the reactants. By dividing the bioreactor into multiple stages with different reactant injection points, the biochemical reactions can be distributed more evenly (Metcalf and Eddy, 2004). In this case, it may shift the reaction toward complete H₂S oxidation by mitigating mass transfer limitation. This bed division or staging configuration also benefited from the lower weight of media pressing down to the lower level that can prevent compaction of the material and clogging of microbial mass leading to lower pressure drop (Elías et al., 2000).

EBRT represents a ratio of empty bed volume of a biotrickling filter to its gas inflow rate. It was reported by many researchers that longer resident time of the H₂S stream within a bioreactor gave higher removal efficiency (Chaiprapat et al., 2011; Charnnok et al., 2013; Montebello et al., 2014). It is worth noticing that the range of EBRT in those studies varied greatly from a few seconds to around 5 min. This was depending mainly on the loading to their systems and concentration of H₂S inlet, oxygen availability, and other operating regimes. Although longer EBRT could benefit the removal of H₂S, excessively large reactor volume typically resulted in higher construction costs. Effects of EBRT as a base parametric variable in interaction with staging and flow pattern on biotrickling filter system performance are of interest in this study.

The aim of this research was to investigate the effects of EBRT and liquid recirculation velocity (q) on H₂S removal efficiency of the triple stage step-feed biotrickling filter in light of a better oxygen distribution through reactor staging configuration. In depth comparison to the single stage biotrickling filter was conducted to identify the degree of improvement in terms of sulfate–sulfide speciation from the proposed multi stage step-feed liquid recirculating system. Finally, a simplified equation with minimal variable to predict H₂S removal was developed and used to compare the single and triple stages biotrickling filters.

2. Method

2.1. Biotrickling filter system and operation

A laboratory single stage biotrickling filter (S-BTF) and triple stage step-feed biotrickling filter (T-BTF) (Fig. 1) were made of cylindrical acrylic with an inner diameter of 4.4 cm and 90 cm in height. The total packed bed media depth was 30 cm equivalent to 456.3 cm³ empty bed volume. The packed bed of T-BTF was divided into 3 stages (lower, middle and upper) at 10 cm high in each stage, to improve dissolved O₂ distribution to the bed. S-BTF with equal size of total packed bed volume was operated for comparison to T-BTF. Coconut husk (the outer layer of the coconut fruit, consisting of long fibers and sponge-like pith particles) was chosen

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