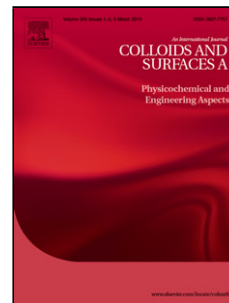


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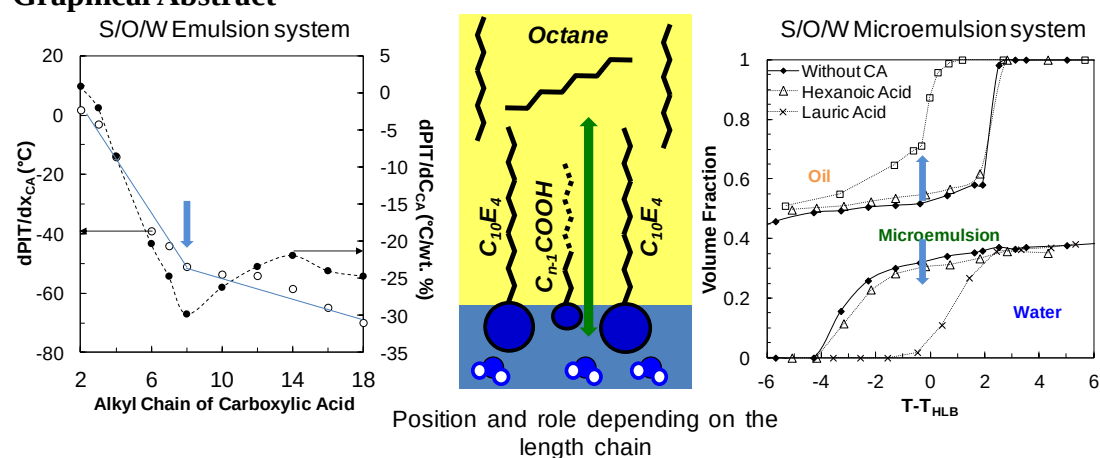
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Graphical Abstract



Highlights

- Influence of carboxylic acids on the PIT of the $C_{10}E_4$ /octane/water reference emulsion system
- Emulsion morphology of Span80/alkane/water system change depending on the added CA.
- Octanoic and higher CA increase solubilization of nonionic SOW equilibrated system.
- Carboxylic acid role in SOW systems depends on alkyl length chain.

Abstract

The influence of various carboxylic acids (from C_2 to C_{18}) on one equilibrated and emulsified surfactant/oil/water (SOW) systems is investigated. The PIT-slope method, based on the influence of an additive on the PIT of the $C_{10}E_4$ / n -octane/ 10^{-2} M $NaCl_{(aq)}$ reference emulsion, shows that from butanoic to hexadecanoic acids, the PIT of the system is decreased. A breakdown on $dPIT/dx$ (considering the molar fraction x) or a minimum on $dPIT/dC$ (taking into account the weight percentage C) is obtained with octanoic acid. The phase behavior of

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