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Baskar Senthilkumar, Ramakrishnan Kalai Selvan

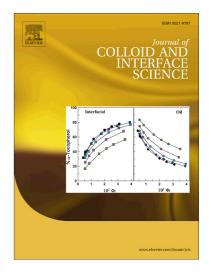
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Hydrothermal synthesis and Electrochemical performances of 1.7 V

NiMoO₄.xH₂O||FeMoO₄ aqueous hybrid supercapacitor

Baskar Senthilkumar and Ramakrishnan Kalai Selvan*

Solid State Ionics and Energy Devices Laboratory, Department of Physics, Bharathiar

University, Coimbatore-641 046, India

Abstract

One-dimensional (1D) NiMoO₄.xH₂O nanorods and βFeMoO₄ microrods are

successfully synthesized by simple hydrothermal method without using any organic solvents.

X-ray diffraction (XRD) patterns reveal the single phase formation of nickel molybdate

(NiMoO₄,xH₂O) and pure monoclinic phase of βFeMoO₄. The growth of one dimensional

morphology of both the molybdate is identified from scanning and transmission electron

microscopic (SEM & TEM) images. The cyclic voltammogram envisage the

pseudocapacitance behaviour of NiMoO₄.xH₂O and βFeMoO₄ through the reversible redox

reactions of Ni³⁺/Ni²⁺ and Fe³⁺/Fe²⁺ ions. An asymmetric supercapacitor is fabricated using

NiMoO₄.xH₂O nanorods and βFeMoO₄ as a positive and negative electrode, respectively. The

βFeMoO₄||NiMoO₄.xH₂O asymmetric supercapacitor delivers a capacitance of 81 F g⁻¹ at a

current density of 1 mA cm⁻². The cell exhibits a high energy density of 29 W h kg⁻¹ and good

cycling stability even after 1000 cycles.

Keywords: Nickel molybdate, Iron molybdate, Asymmetric capacitor, Pseudocapacitance,

Energy Density

*Corresponding author: selvankram@buc.edu.in (R.K.Selvan)

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