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Degradation of Chlortetracycline Using Nano Micro-Electrolysis Materials with Loading Copper

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ABSTRACT

Nano micro-electrolysis materials with loading copper were synthesized by a modified reduction method under inert atmosphere and characterized with XRD, BET, SEM and SEM mapping. The effects of nano micro-electrolysis material types, initial pH, nano micro-electrolysis material with loading copper dosage, and initial chlortetracycline concentration on chlortetracycline removal were evaluated. Batch studies results indicated that the maximum removal and total organic carbon removal of chlortetracycline were found to be 99.1% and 70.7% under the optimal conditions of $m=0.5\text{g/L}$, $\text{pH}=3.6$ and $C_0=50\text{ mg/L}$. The degradation mechanisms and kinetics were investigated, and the two constants model was discovered to be reasonable fit for the experimental data kinetic model with initial chlortetracycline as variables. These results demonstrated that nano micro-electrolysis materials with loading copper could potentially be used for chlortetracycline removal.

Keyword: Nano micro-electrolysis, Chlortetracycline, Degradation mechanism

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