Accepted Manuscript

Removal of Cr(VI) and fluoride by membrane capacitive deionization with nanoporous and microporous *Limonia acidissima* (wood apple) shell activated carbon electrode

Mahendra S. Gaikwad, Chandrajit Balomajumder

PII:	S1383-5866(17)32958-1
DOI:	https://doi.org/10.1016/j.seppur.2017.12.006
Reference:	SEPPUR 14239
To appear in:	Separation and Purification Technology
Received Date:	10 September 2017
Revised Date:	13 November 2017
Accepted Date:	3 December 2017



Please cite this article as: M.S. Gaikwad, C. Balomajumder, Removal of Cr(VI) and fluoride by membrane capacitive deionization with nanoporous and microporous *Limonia acidissima* (wood apple) shell activated carbon electrode, *Separation and Purification Technology* (2017), doi: https://doi.org/10.1016/j.seppur.2017.12.006

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

ACCEPTED MANUSCRIPT

Removal of Cr(VI) and fluoride by membrane capacitive deionization with nanoporous and microporous *Limonia acidissima* (wood apple) shell activated carbon electrode

Mahendra S. Gaikwad^{*}, Chandrajit Balomajumder

Department of Chemical Engineering, Indian Institute of Technology Roorkee, Roorkee-247667, Uttarakhand, India.

*Corresponding author.

E-mail: mahendra14g@gmail.com (M. S. Gaikwad)

chandfch@iitr.ernet.in (C. B. Majumder)

Abstract

Membrane capacitive deionization (MCDI) is an attractive desalination technique to achieve pure water by adsorbing ions in charge porous electrodes. In the current study nano-porous and micro porous activated carbon was derived from *Limonia acidissima* shells. The *Limonia acidissima* shell activated carbon (LASAC) prepared with chemical and thermal modification. The characterization techniques such as SEM, TGA, DTA, DTG, and N₂ adsorption–desorption isotherm analysis were used for analysis LASAC. The LASAC electrode was used in MCDI lab scale self-made set up for simultaneous sorption of Cr(VI) and fluoride from the aqueous system. The LASAC electrode performance of was found effective for simultaneous removal of Cr(VI) and fluoride from mixed feed solution with maximum electrosorption capacity of Cr(VI) and fluoride were found to 0.8086 mg/g and 0.7805 mg/g for 10 mg/L; 3.227 mg/g and 2.7554 mg/g Download English Version:

https://daneshyari.com/en/article/7044030

Download Persian Version:

https://daneshyari.com/article/7044030

Daneshyari.com