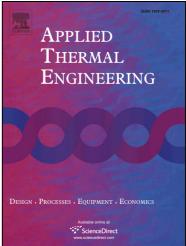
Accepted Manuscript

Phase change behaviour of latent heat storage media based on calcium chloride hexahydrate composites containing strontium chloride hexahydrate and oxidation expandable graphite

Xiang Li, Yuan Zhou, Hongen Nian, Xiufeng Ren, Ouyang Dong, Chunxi Hai, Yue Shen, Jinbo Zeng

PII:	S1359-4311(16)30402-1
DOI:	http://dx.doi.org/10.1016/j.applthermaleng.2016.03.098
Reference:	ATE 7966
To appear in:	Applied Thermal Engineering
Received Date:	4 January 2016
Accepted Date:	23 March 2016



Please cite this article as: X. Li, Y. Zhou, H. Nian, X. Ren, O. Dong, C. Hai, Y. Shen, J. Zeng, Phase change behaviour of latent heat storage media based on calcium chloride hexahydrate composites containing strontium chloride hexahydrate and oxidation expandable graphite, *Applied Thermal Engineering* (2016), doi: http://dx.doi.org/ 10.1016/j.applthermaleng.2016.03.098

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

ACCEPTED MANUSCRIPT

Phase change behaviour of latent heat storage media based on

calcium chloride hexahydrate composites containing strontium

chloride hexahydrate and oxidation expandable graphite

Xiang Li^{a, b, c,} *, Yuan Zhou^{a, c,}*, Hongen Nian^{a, b, c}, Xiufeng Ren^{a, b, c}

Ouyang Dong ^{a,c}, Chunxi Hai ^{a,c}, Yue Shen ^{a,c}, Jinbo Zeng ^{a, c}

^a Qinghai Institute of Salt Lakes, Chinese Academy of Sciences, Xining, 810008, China, ^b University of Chinese Academy of Sciences, Beijing, 100049, China,

^cKey Laboratory of Salt Lake Resources Chemistry of Qinghai Province, Xining, 810008, China

Abstract:

Calcium chloride hexahydrate(CaCl₂·6H₂O) composites PCMs containing strontium chloride hexahydrate (SrCl₂·6H₂O) and oxidation expandable graphite(EGO) were prepared and phase change behavior of CaCl₂·6H₂O/EGO/SrCl₂·6H₂O composite PCMs, including supercooling degree, phase change temperature, latent heat, density, thermal conductivity and thermal stability were systematically studied. Results demonstrate that thermal conductivity, heat transfer and the latent capacities of the CaCl₂·6H₂O/SrCl₂·6H₂O/EGO composite PCMs are significantly enhanced, supercooling of CaCl₂·6H₂O/SrCl₂·6H₂O/EGO composite PCMs are suppressed, mainly ascribe to that the EGO are homogenously dispersed in PCMs due to the existence of oxygen-containing functional groups in EGO, meanwhile, the well-dispersed EGO serving as nuclei to promote the heterogeneous nucleation and crystallization process of CaCl₂·6H₂O/EGO composites PCMs contained with 3wt.% SrCl₂·6H₂O/SrCl₂·6H₂O/EGO possesses enhanced thermal behavior with no phase segregation observed; the latent heat was calculated to be 172.26 J/g.

Keywords: Phase change material (PCM); Oxidation expandable graphite (EGO); CaCl₂·6H₂O/SrCl₂·6H₂O/EGO composites; Phase change behavior

1. Introduction

Phase change materials (PCMs) offer an effective route for the efficient usage of latent thermal energy. PCMs are commonly adopted for thermal energy storage field applications e capability of operating under low/medium temperatures and unstable energy storage fields, such as solar energy, industrial waste heat, intermittent electric heating energy, ect [1-4]. The study and development of PCMs that demonstrate high performance is of great significance in developing more efficient routes of energy

^{*} Corresponding author. Address: Qinghai Institute of Salt Lakes, Chinese Academy of Sciences, Xining 810008, China.

Tel.: +86 971 6338403; fax: +86 971 6338403.

E-mail address: lixiang@isl.ac.cn (X. Li), zhouy@isl.ac.cn (Y. Zhou).

Download English Version:

https://daneshyari.com/en/article/7047825

Download Persian Version:

https://daneshyari.com/article/7047825

Daneshyari.com