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## Assessment of existing two phase heat transfer coefficient and critical heat flux correlations for cryogenic flow boiling in pipe quenching experiments



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ABSTRACT

To enable efficient design and analysis of cryogenic propellant transfer systems, high accuracy models are required for predicting two phase flow boiling and heat transfer at reduced temperatures. The penalty for poor models translates into higher margin, safety factor, and ultimately cost in design. Recently, there has been a drive towards developing universal correlations to cover a broad range of fluids, tube diameters, and thermodynamic conditions for predicting heat flux and pressure drop. These correlations do not, however, cover cryogenic fluids like liquid hydrogen. Therefore the purpose of this paper is to apply popular two phase heat transfer correlations used in commercial codes against available flow boiling data for cryogenic fluids. Specifically, quenching test data for critical heat flux and two phase heat transfer coefficient are compared against the correlations. Results show that existing correlations over-predict heat transfer by as much as 20,000% and that significant model improvements are warranted.

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#### 1. Introduction

#### 1.1. Role of cryogenic fluids in modern world

Cryogenic fluids, which are substances that exist as liquids at extremely low temperatures, are employed in a wide variety of applications throughout industry. Liquid nitrogen (LN<sub>2</sub>) is used to fast freeze food [1], to preserve tissues and blood [2], and to kill unhealthy tissues in cryosurgery [3]. Liquid oxygen (LOX) is used in the medical industry, life support systems, and fuel cells [4]. In the space industry, liquid helium (LHe) is used to chill down Earth-orbiting telescopes and satellites [5,6]. Liquid hydrogen (LH<sub>2</sub>) is used to chill down superconducting magnets [7,8] and as rocket fuel to prechill [9] and ignite high performance engines such as the Shuttle [10]. Perhaps the most prolific use of cryogenic fluids is in the proposed fuel depots [11,12]. A depot is defined as an Earth-orbiting propellant storage vessel that will be used to store LOX and LH<sub>2</sub> in Low Earth Orbit (LEO) indefinitely to refuel spacecraft [13]. This technology will enable long duration human and robotic missions beyond LEO because a higher percentage of spacecraft mass can be used for payload or for larger engines, and the vehicle can achieve higher velocities once outside the gravity well of Earth.

Before cryogenic liquid can flow, the transfer line and associated hardware must be chilled down or "quenched" to temperatures below the fluid saturation temperature. The most direct, repeatable, and reliable method to remove heat is to use the cryogen itself to quench the transfer system. Due to the ultralow normal boiling point of cryogens, phase change, complex flow patterns, two-phase flow boiling, and heat transfer are inevitable during the chilldown process.

#### 1.2. Importance of accurate cryogenic flow boiling predictive tools

Experimental and numerical studies on two-phase flow have been carried out for nearly a century. Hundreds of carefully controlled experiments have been performed, resulting in a large database that covers multiple fluids, flow geometries, heat input, and fluid quality. As a result, numerous empirical correlations have been proposed to model two-phase heat transfer coefficient (HTC), pressure drop, and heat flux.

The complexity of two-phase flow features that occur during chilldown makes it difficult to provide correlations that are valid

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#### Nomenclature

A	A	area [m <sup>2</sup> ]	$\Delta P_{sat}$	saturation pressure difference based on wall and fluid
E	30	boiling number		[Pa]
С	P	specific heat [J/kg K]	3	emissivity
a	l	diameter [m]	μ	viscosity [kg/m s]
F	7	two phase multiplier	ρ	density [kg/m <sup>3</sup> ]
F	III	view factor	σ	Stefan–Boltzmann constant [W/m <sup>2</sup> K <sup>4</sup> ]
F	ř	Froude number	x	quality
(	- J	mass flux [kg/m <sup>2</sup> s]	Xe	equilibrium quality
g	Ţ	gravity [m/s <sup>2</sup> ]	Ψ	two phase constant
Ĭ	i	heat transfer coefficient [W/m <sup>2</sup> K]	,	
k	lfo	latent heat []/kg]	Subscript	S
k	li	enthalpy []/kg]	h	hoiling
Ι		current [A]	ch	convection boiling
k	ć	thermal conductivity [W/m K]	C	critical
I		length [m]	CHE	critical heat flux
r	'n	mass flow rate [kg/s]	d	based on diameter
N	ЛАF	mean absolute error	u ovp	ovporimental
N	лw	molecular weight	exp f	experimental finid
N	J	number of data points	J ED	film boiling
ŀ	)	pressure	ГD	
ŀ	0_	wetted perimeter [m]	g	gds
ŀ	F D,,	heated perimeter [m]	gascona	gas conduction
F	H Dr	Prandtl number	ні ;	ineated tube
1	י ר	hand humber	1	inner
	۲ ۳	heat flux [W/m <sup>2</sup> ]		based on length
9	1	radius [m]	l, lo	liquid only
	20	Pounolde number	lv	liquid/vapor
r c	le l	nucleate boiling suppression factor	тіс	macro-convective
נ ר	PN	tomporature [K]	mic	micro-convective
1		time [e]	mix	mixture
l	T	une [s]	NB	nucleate boiling
	J ,	uncertainty	0	outer
V	/	voltage [V]	R	reduced
V	,	velocity [m/s]	rad	radiation
V	ve	Weber number	S	surface
2	K <sub>tt</sub>	Martinelli parameter	sat	saturated
Y		mass fraction	solidcond	solid conduction
Z		distance along pipe [m]	spl	single phase liquid
			TB	transition boiling
(	Greek		tp	two phase
0	ζ	thermal diffusivity [m <sup>2</sup> /s]	ν	vapor
2	,	surface tension [N/m]	w	wall
Z	$\Delta T_e$	temperature difference between wall and fluid satura- tion [K]		

x	quality	
Xe	equilibrium quality	
$\psi$	two phase constant	
Subscripts	1	
b	boiling	
cb	convection boiling	
С	critical	
CHF	critical heat flux	
d	based on diameter	
exp	experimental	
f	fluid	
FB	film boiling	
g	gas	
gascond	gas conduction	
ΗT	heated tube	
i	inner	
L	based on length	
l, lo	liquid only	
ĺv	liquid/vapor	
тіс	macro-convective	
тіс	micro-convective	
mix	mixture	
NB	nucleate boiling	
0	outer	
R	reduced	
rad	radiation	
S	surface	
sat	saturated	
solidcond	solid conduction	
spl	single phase liquid	
ŤВ	transition boiling	
tp	two phase	
v	vapor	
w	wall	

over a wide range of conditions. Typically, separate heat transfer correlations must be used for modeling different boiling regimes and different flow patterns. For example, the Chen [14] correlation is valid for modeling the nucleate boiling (NB) regime over a wide range of fluids:

$$h_{\rm NB} = h_{mic}S_N + h_{mac}F \tag{1}$$

where  $h_{mac}$  is the "macro-convective" HTC and  $h_{mic}$  is the "microconvective" HTC based on bubble nucleation growth rate where  $S_N$  is the NB suppression factor that accounts for differences between measured superheat and superheat from surface cavities caused by the presence of the thermal boundary layer, and F is the two-phase multiplier. Both constants are fit to experimental data:

$$S_N = \frac{1}{1 + 0.00000253 \text{Re}_m^{1.17} F^{1.4625}} \tag{2}$$

where the two-phase Re number is defined as:

$$\operatorname{Re}_{tp} = \frac{Gd_i}{\mu_l} (1 - x_e) \tag{3}$$

$$F = \left(\frac{1}{X_{tt}} + 0.213\right)^{0.736}$$
(4)

where the Martinelli parameter is used:

$$X_{tt} = \left(\frac{1-x}{x}\right)^{0.9} \left(\frac{\rho_{\nu}}{\rho_l}\right) \left(\frac{\mu_l}{\mu_{\nu}}\right)^{0.1}$$
(5)

The two HTCs are defined as:

$$h_{mic} = 0.00122 \frac{k_l^{0.79} c_{p,l}^{0.45} \rho_l^{0.49} \Delta T_e^{0.24} \Delta P_{sat}^{0.75}}{\gamma_{l\nu}^{0.29} \mu_l^{0.29} h_{fg}^{0.24} \rho_{\nu}^{0.24}}$$
(6)

$$h_{mac} = 0.023 \operatorname{Re}_{tp}^{0.8} \operatorname{Pr}_{l}^{0.4} \left( \frac{k_{l}}{d_{i}} \right)$$
(7)

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