



The drying rates of spray freeze drying systems increase through the use of stratified packed bed structures



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ABSTRACT

It is shown that when in the spray freeze drying process stratified packed beds are employed where the smallest in size particles are located near the surface of the lower heating plate while the largest particles are occupying the upper part of the stratified packed bed whose outermost surface is in contact with the atmosphere of the drying chamber of the freeze dryer, the duration times of the primary and secondary drying stages can be substantially reduced when compared with those required by a spray freeze drying process which uses a single packed bed formed by particles all having the same particle size and the total amount of solids in the packed bed is the same as that in the stratified packed bed.

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1. Introduction

Spray freeze drying (SFD) provides a separation process that can produce porous particles whose physical and chemical characteristics are considered to be ideal for use in pulmonary drug delivery, the production of powders for epidermal immunization, in the processing of low water soluble drugs, the preparation of particles for microencapsulation, the processing of particles for use in chemical catalysis and biocatalysis, the preparation of nanopowders and ceramic electronic parts, in the cryogenic processing of chemicals and powder based materials, in the production of high value porous food particles, and in the synthesis of Li-ion battery cathode materials [1–7]. SFD involves the packing of frozen particles in containers and, thus, a packed bed [2,3,7–13] of frozen particles is formed which has a porous structure and makes the frozen region of the material to be unsaturated during primary drying [2] because the space of the frozen region formed by the packed frozen particles is partially filled with gas (inert gas and solvent (e.g., water) vapor) which moves through the pores of the unsaturated porous frozen region by convection, Knudsen diffusion, and bulk diffusion during primary drying, as per Fig. 1a. During secondary drying, the bound (sorbed) solvent (e.g., water) is desorbed from the surface of the pores of the particles being dried and is transported through the pores of the porous structure of the particles

as well as through the pore space of the packed bed by convection, Knudsen diffusion, and bulk diffusion [2,3].

It has been shown theoretically by Liapis and Bruttini [2,3] that the longer drying times required in SFD when compared to the drying times required in classical freeze drying, are mainly due to the reduced heat and mass transfer capabilities of the porous packed beds formed by the packed frozen particles and can also lead to the formation of a secondary dried layer near the surface of the lower heating plate during the primary drying stage, as per Fig. 2. The formation of this secondary dried layer has been confirmed experimentally [4,5] and contributes to the deterioration of the drying rate. Furthermore, Liapis and Bruttini [2] have shown that in the SFD process the drying rate during the primary drying stage increases as (i) the product height decreases, (ii) the particle diameter increases, and (iii) the value of the packing porosity increases. Liapis and Bruttini [2] have also presented the physico-chemical mechanisms which provide the reasons for obtaining the results in items (i)–(iii) above.

It is important to indicate here that for the SFD processing of pharmaceutical products and ceramic powders the magnitude of the ratio of the size of the packed bed of particles to the particle diameter is most often larger than 1000 and the SFD model of Liapis and Bruttini [2] was found to describe the physics and the dynamic behavior of the SFD process satisfactorily and its predictions are consistent with and describe appropriately the experimental results and observations published in the literature [4–6]. The SFD process could also be used in the production of high value porous food particles. But the food particles have diameters of

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