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Characterization of a dual fluidized bed gasifier with blended biomass/coal as feedstock



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ABSTRACT

A one-dimensional model is built based on the commercial Aspen Plus software to kinetically simulate the biomass/coal co-gasification process in a dual fluidized bed gasifier. The synergistic effect on the co-gasification kinetics is allowed for, and is coupled with the gas–solid flow hydrodynamics. With the developed model, the effects of different key operating parameters including the biomass blending ratio (R_b), the initial bed temperature (T_g), the feedstock mass flow rate (F_{fs}), the bed material flux (F_{bm}) and the steam to carbon ratio (R_{sc}) on the resultant syngas composition and the supplemental fuel mass flow rate (F_{sf}) are investigated, and the operation parameters are optimized. It is found that increasing R_b and T_g can enhance the gasification, while increasing F_{fs} and R_{sc} restricts the gasification. Increasing F_{bm} has slight effect on the gasification results but can reduce F_{sf} . The cold gas efficiency is up to 78.9% under the proposed optimum condition.

1. Introduction

Biomass currently provides most of the renewable energy (excluding hydroelectricity) consumed in the industrial sector and will continue to do so in the future. It is also the fourth largest fuel source in the world, and shares about 2.7% of the global energy demands (World energy outlook, 2016). Moreover, it is clean and carbon neutral during the thermal conversion. Despite of all these advantages, there are also some inherent disadvantages in terms of biomass utilization. First, the heating value of biomass is low, making its transportation or storage uneconomical. Second, the supply of biomass is seasonally dependent, making it incompetent for large scale utilization. Third, gasification of biomass is apt to generate tar and cause secondary pollution. Compared to biomass, coal is the most abundant but dirtiest fuel in the world. Direct combustion of coal discharges large amount of pollutants and CO₂, which is adverse to the living environment. Although the gasification technology offers a compelling route for clean coal utilization, the relatively low activity of coal char brings about another issue. Moreover, as the non-renewable resource, coal will be exhausted one day in the future.

Co-gasification of biomass and coal is prospective to solve all the above issues. First, the coal consumption per unit power generation can be lowered when biomass is added to the feedstock (Yan et al., 2016a). Second, discharge of pollutants and CO_2 per unit power generation can be reduced (Yan & He, 2017). Third, the co-gasfication temperature is relatively higher, beneficial to the tar decomposition (Xiang et al.,

2017). Fourth, the synergistic effect may happen and promotes the char conversion (Yan et al., 2018). Since most of the gasification processes are endothermic, part of the feedstock is burnt to meet the required heat consumption in industry. If pure oxygen is chosen as oxidant, great efficiency penalty will be imposed. If air is chosen as oxidant, the gasification products can be greatly diluted by large amount of N₂. This contradiction can be perfectly solved by the dual fluidized bed (DFB) gasification technology (Yan et al., 2016b), in which the gasification and combustion processes are well separated from each other to avoid the dilution of syngas, and the heat required by the endothermic gasification is transported from the combustor by the circulating bed material like silica sand.

Co-gasification of biomass/coal in the DFB gasifiers draws upon the inherent advantages of both the co-gasification technology and the dual bed gasification technology, making it very promising for the clean and efficient utilization of the biomass and coal resources. Till now, many concerns have been attracted to this topic and many contributions have been done by researchers worldwide. Chen et al. (2017) simulated the biomass/coal co-gasification in an internally circulating fluidized bed (ICFB) gasifier using the kinetic theory of granular mixture. The variations of granular temperatures of the coal and biomass particles against the solid volume fraction, gasification temperature and carbon concentration were particularly investigated. The synergistic effect on the co-gasification kinetics was, however, not considered. Wang and Chen (2013) experimentally investigated the effect of temperature on the composition and lower heating value (LHV) of the syngas from an

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Nomenclature		R_{g}	gas constant, 8.3145, [J/K]
		\vec{R}_{H2O}	steam conversion, [-]
а	decay constant	R _{sc}	steam-to-carbon ratio, [-]
$A_{\rm b}$	bubble phase cross-sectional area, [m ²]	$S_{\rm cf}$	sand circulation flux, [kg/(m ² ·s)]
$A_{\rm d}$	ash content in dried basis, [-]	$T_{\rm b}$	temperature in bubble, [K]
A_{i}	pre-exponential factor [1/s]	T_{e}	temperature in emulsion, [K]
Ar	Archimedes number	T_{g}	initial gasification temperature [°C]
$A_{\rm t}$	bed cross-sectional area, [m ²]	U_0	superficial velocity, [m/s]
$C_{\rm b}$	concentration in bubble, [mole/m ³]	$U_{\rm br}$	bubble rise velocity, [m/s]
$C_{\rm e}$	concentration in emulsion, [mole/m ³]	$U_{\rm mb}$	minimum bubbling velocity, [m/s]
Ci	concentration of species j , [kmole/m ³]	$U_{\rm mf}$	minimum fluidization velocity, [m/s]
$C_{p,p}$	particle specific heat, [J/(kg·k)]	V _{ad}	volatile in air-dried basis, [-]
$d_{\rm p}$	particle diameter, [m]	$V_{\rm b}$	volume of bubble phase, [m ³]
$D_{\rm B}$	bubble diameter, [m]	Ve	volume of emulsion phase, [m ³]
$\overline{D}_{\rm BO}$	initial bubble diameter, [m]	x	conversion [%]
$D_{\rm BM}$	maximum bubble diameter, [m]	$Y_{\rm biom}$	yield from biomass [%]
D_{i}	diffusion coefficient, [m ² /s]	Y _{coal}	yield from coal [%]
D_{t}	column diameter, [m]	Y _{tv}	volatile fraction by calculation [%]
Ei	activation energy [kJ/mol/K]	Y	volatile fraction by proximate analysis [%]
$F_{\rm b}$	molar flow rate in bubble, [mole/s]	z	length of PFR, [m]
$F_{\rm bm}$	circulating flux in combustor [kg/m ² /h]	Z_f	height above distributor, [m]
F _e	molar flow rate in emulsion, [mole/s]	J	
$F_{\rm f}$	feedstock mass flow rate [kg/h]	Greek sy	mbols
g	acceleration of gravity, $[m/s^2]$		
$h_{\rm be}$	bubble-to-emulsion heat transfer coefficient, [W/(m ² ·K)]	$\delta_{ m b}$	bubble phase volume fraction, [-]
$h_{\rm gp}$	gas-to-particle heat transfer coefficient, [W/m ² /K]	$\varepsilon_{\rm b}$	bubble phase voidage, [-]
Ĥ	total enthalpy flow, [kJ/s]	$\varepsilon_{\rm mf}$	emulsion voidage at minimum fluidization, 0.4, [-]
ΔH	enthalpy generation, [J/mole]	μ_{g}	gas viscosity, [kg/(m·s)]
k	reaction rate constant	ρ _g	gas density, [kg/m ³]
k,	gas thermal conductivity, [W/(m·K)]	ρ _p	particle density, [kg/m ³]
ks	solid thermal conductivity, [W/(m·K)]	τ	residence time of particle in bubble, [s]
K _{be}	bubble-to-emulsion mass transfer coefficient, [1/s]	$\phi_{ m d}$	dense phase solids volume fraction, [-]
$L_{\rm d}$	dense bed height, [m]	ϕ^*	saturation carrying capacity, [-]
L_{f}	freeboard height, [m]	$\phi_{ m f}$	freeboard solids volume fraction, $[-]$
$M_{ m ad}$	moisture in air-dried basis, [-]	$\psi_{\rm char}$	one minus char conversion, [-]
M_s	mass of bed material, [kg]	Ψ_i	structure parameter
$M_{s,i}$	mass of solids in ith stage, [kg]		
n	total number of species, [-]	Abbreviations	
n _d	number of orifices in distributor, $[-]$		
р	operation pressure, [Pa]	AAEM	alkali and alkaline metal
P_{j}	partial pressure of species <i>j</i> , [bar/atm]	BAM	bubble assemblage model
Pr_{p}	Prandtl number, [–]	BFB	bubbling fluidized bed
$r_{(i),j}$	consumption rate of species <i>j</i> in stage <i>i</i> , [kmole/($m^3 \cdot s$)]	CFB	circulating fluidized bed
r _{blend}	net increment of r _{coal} [1/s]	CSTR	continuous stirred tank reactor
$r_{\rm coal}$	conversion rates of coal char [1/s]	DFB	dual fluidized bed
$r_{\rm mix}$	conversion rates of blended char [1/s]	LHV	lower heating value
$R_{\rm b}$	biomass blending ratio	PFR	plug flow reactor
$R_{\rm c}$	char conversion	WGS	water–gas shift
Re_{p}	Reynolds number, [–]	ICFB	internally circulating fluidized bed
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autothermal fluidized bed gasifier, and found that higher reaction temperature leaded to higher H_2 and CO contents, as well as higher energy and exergy efficiencies. But, the syngas LHV decreased with the temperature increment. Saw and Pang (Saw & Pang, 2013) studied the influence of the lignite blending ratio on the producer gas composition and tar content in a 100 kW DFB steam gasifier, and found that increasing the lignite blending ratio could increase the H_2 /CO ratio and reduce the tar concentration. In addition, the synergistic effect was detected in the research. Aigner et al. (2011) studied the co-gasification of coal and wood in a 100 kW DFB gasifier, and found that the producer gas composition varied linearly with linear variation of the wood blending ratio. The contents of NH₃ and H₂S in the producer gas fell with decreasing coal blending ratio. Masnadi et al. (2015) investigated the synergistic effects during the steam co-gasification of switchgrass and coal in a pilot-scale bubbling fluidized bed gasifier, and found that the hydrogen concentration, the cold gas efficiency, the gas yield and the higher heating value (HHV) of the producer gas were enhanced remarkably relative to single-fuel gasification with the switchgrass added to coal. The alkali metals in biomass ash synergized with coal and catalyzed the gasification. Miccio et al. (2012) experimentally studied the co-gasification of biomass and brown coal in an ICFB gasifier, and found that the hydrogen concentration in the producer gas was up to 35% in the best case. The tar content was reduced greatly when changing from inert to partially catalytic bed.

From the literature review, it is known that research on the biomass/coal co-gasification in the DFB gasifiers is mainly restricted to the experimental level. Kinetic simulations of the co-gasification process in DFB, especially those considering the synergistic effect, are rarely Download English Version:

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