FISEVIER

Contents lists available at ScienceDirect

## **Bioresource Technology**

journal homepage: www.elsevier.com/locate/biortech



# Kinetics and thermodynamics of biodegradation of hydrolyzed polyacrylamide under anaerobic and aerobic conditions \*



Lanmei Zhao a,b, Mutai Bao a,b,\*, Miao Yan a,b, Jinren Lu b

<sup>a</sup> Key Laboratory of Marine Chemistry Theory and Technology, Ministry of Education, Ocean University of China, Qingdao 266100, China <sup>b</sup> College of Chemistry and Chemical Engineering, Ocean University of China, Qingdao 266100, China

#### HIGHLIGHTS

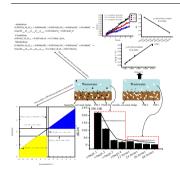
- HPAM biodegradation kinetics were established in the activated sludge system.
- Thermodynamic entropy changes (ΔS) were calculated in biochemical treatment system.
- Thermodynamic windows of opportunity for the HPAM biodegradation were drawn.
- The feasibility of the HPAM biodegradation to final product was demonstrated.
- Growth-process equation confirmed electron equivalence between substrate and product.

#### ARTICLE INFO

Article history: Received 22 March 2016 Received in revised form 12 May 2016 Accepted 14 May 2016 Available online 19 May 2016

Keywords: Hydrolyzed polyacrylamide Biodegradation Kinetics Thermodynamics

#### G R A P H I C A L A B S T R A C T



#### ABSTRACT

Kinetics and thermodynamics of hydrolyzed polyacrylamide (HPAM) biodegradation in anaerobic and aerobic activated sludge biochemical treatment systems were explored to determine the maximum rate and feasibility of HPAM biodegradation. The optimal nutrient proportions for HPAM biodegradation were determined to be  $0.08~\rm g \cdot L^{-1}~\rm C_6H_{12}O_6$ ,  $1.00~\rm g \cdot L^{-1}~\rm NH_4Cl$ ,  $0.36~\rm g \cdot L^{-1}~\rm NaH_2PO_4$  and  $3.00~\rm g \cdot L^{-1}~\rm K_2HPO_4$  using response surface methodology (RSM). Based on the kinetics, the maximum HPAM biodegradation rates were  $16.43385~\rm mg \cdot L^{-1} \cdot d^{-1}$  and  $2.463~\rm mg \cdot L^{-1} \cdot d^{-1}$  in aerobic and anaerobic conditions, respectively. The activation energy ( $E_a$ ) of the aerobic biodegradation was  $48.9897~\rm kJ \cdot mol^{-1}$ . Entropy changes ( $\Delta S$ ) of biochemical treatment system decreased from  $216.21~\rm J \cdot K^{-1}$  to  $2.39~\rm J \cdot K^{-1}$ . Thermodynamic windows of opportunity for HPAM biodegradation were drawn. And it demonstrated HPAM was biodegraded into acetic acid and  $CO_2$  under laboratory conditions. Growth-process equations for functional bacteria anaerobically grown on polyacrylic acid were constructed and it confirmed electron equivalence between substrate and product.

© 2016 Elsevier Ltd. All rights reserved.

E-mail address: mtbao@ouc.edu.cn (M. Bao).

#### 1. Introduction

HPAM-containing oilfield wastewater is a type of relatively complex and special sewage that can cause a series of environmental problems whether through reinjection or discharge. High viscosity and strong emulsification of the wastewater lead to difficulties in its treatment (Guezennec et al., 2015). Treatment

<sup>☆</sup> This is MCTL Contribution No. 121.

<sup>\*</sup> Corresponding author at: Key Laboratory of Marine Chemistry Theory and Technology, Ministry of Education, Ocean University of China, Qingdao 266100, China.

methods generally involve mechanical degradation (Shao et al., 2005; Al Hashmi et al., 2013), chemical oxidation (Caulfield et al., 2002; Zhou et al., 2015), biodegradation (Kay-Shoemake et al., 1998a,b; El-Mamouni et al., 2002; Bao et al., 2010) and new technologies that combine these methods (Lu and Wei, 2011; Pi et al., 2015). The effect of treatment has improved to some extent. Physical/chemical methods have some drawbacks, such as secondary pollution problems and substantially increased costs. Therefore, environmentally friendly, low-cost biological treatments play an important role in the field of HPAM-containing wastewater treatment.

Most of the previous studies have focused on the methods of improving HPAM biodegradation. Yan et al. (2016) investigated HPAM biodegradation in a sequencing batch biofilm reactor, and the HPAM removal ratio reached 54.69%. After anaerobic/aerobic coupling biodegradation, the HPAM removal ratio reached 90%. and HPAM with a relative molecular mass of approximately  $2.2 \times 10^7$  was biodegraded into small molecule segments of approximately  $3.5 \times 10^3$  (Sang et al., 2015). HPAM-containing oilfield wastewater was treated by biological processes combined Fenton oxidation, and the maximum removal rate of COD<sub>Cr</sub> and HPAM reached 94.61% and 91.06%, respectively (Pi et al., 2015). A zero-valent iron/EDTA/air chemical oxidation/sequencing batch activated sludge system for treatment of the wastewater had a total HPAM removal rate of 96% (Lu and Wei, 2011). Other research revealed HPAM was a source of either carbon or nitrogen for microorganisms, and the metabolic pathways of HPAM biodegradation occurred under both anaerobic and aerobic conditions. Kay-Shoemake et al. (1998a,b) showed that soil microorganisms utilized HPAM as a sole N-source and hydrolyze the amide group by amidase, which converted HPAM into ammonia and organic acids. Cross-linked PAM stimulated the activity of microorganisms and the generation of methane under anaerobic conditions in the absence of a nitrogen source (Haveroen et al., 2005). Wen et al. (2010) reported two strains isolated from oil contaminated soil and activated sludge that used PAM as their sole carbon source. Bao et al. (2010) inferred the biodegradation mechanism of microbial enzymes. An amide enzyme hydrolyzed amino into carboxyl and microorganisms attacked the methyl group at the end of the side chain. Finally, the methyl group was oxidized to acid under the catalysis of a single oxygenase. The chain skeleton of HPAM formed acetic acid, propionic acid, butyric acid, isobutyric acid, valeric acid and isovaleric acid from different positions of fracture (Dai et al., 2014). In addition, Dai et al. (2015) also interpreted metabolic pathways for the biological hydrolysis of PAM from the perspective of enzymes.

However, four major issues exist: (i) How is the maximum biodegradation rate of HPAM determined? (ii) What does the model of kinetics for HPAM biodegradation conform to under anaerobic and aerobic conditions? (iii) Can HPAM be degraded completely by microorganisms? (iv) How many opportunities exist for reactions to be carried out?

Kinetics for utilization of different substrate was also researched. Pala-Ozkok et al. (2013) studied the effect of sludge age on acetate utilization kinetics. The maximum growth rate was observed to change from 3.9/d to 8.5/d when the sludge age was decreased from 10 d to 2.0 d. Orhon et al. (2009) found that peptone biodegradation at different sludge ages under aerobic conditions conformed to Monod kinetics. The kinetics of aerobic biodegradation of hydroquinone and catechol were successfully described by a Haldane model (Pramparo et al., 2012). First-order exponential and sigmoidal models were used for the two different stages in biochemical methane potential tests (Fernández-Rodrí guez et al., 2014). In addition, Battley (1998) studied the thermodynamics of microbial growth by calculating ΔS and free energy change. In the same way, the kinetics and thermodynamics can

be used to calculate the maximum rate and feasibility for HPAM biodegradation under anaerobic and aerobic conditions.

The objectives of this study were to (i) build the kinetics models of HPAM biodegradation in anaerobic and aerobic activated sludge biochemical treatment systems, (ii) investigate the effects of temperature on the HPAM biodegradation rate constant and determine  $E_{\rm a}$ , (iii) calculate  $\Delta S$  and demonstrate the feasibility of the biodegradation process from HPAM to CH<sub>4</sub> and CO<sub>2</sub> and (iv) construct growth-process equations for functional bacteria anaerobically grown on polyacrylic acid and confirm electron equivalence between substrate and product.

#### 2. Materials and methods

#### 2.1. Materials

#### 2.1.1. Wastewater composition

The average molecular weight of HPAM in the sample was approximately  $2.2 \times 10^7$ , and it was 10% hydrolyzed. The HPAM wastewater in this study was dispensed to simulate the wastewater of Gu Liu joint station, located in Dongying, China. The composition of the wastewater is shown in Table 1. The optimal nutrient proportions were added as reported in Section 3.1, and microelements were added as nutriments for microorganism growth (Ren et al., 2008). NaHCO<sub>3</sub> was used to regulate the pH value between 7.0 and 7.5.

#### 2.1.2. Inoculated sludge

The inoculated granular sludge used as inoculum was obtained from Gu Liu joint station, where one of the treatment stations of HPAM-containing wastewater is located. The sludge was selected for its high activity. This inoculum contained dominant bacteria, and it adapted to the wastewater. The main characteristics of inoculated sludge were: 1–2 mm average particle size, 1.76 mg·L $^{-1}$  total phosphorus (TP), 3.97 mg·L $^{-1}$  total nitrogen (TN), 32.54 sludge volume index (SVI) and 14.89 mg·L $^{-1}$  volatile suspended solid (VSS).

#### 2.1.3. Functional bacteria

The three strains of bacteria for HPAM biodegradation were isolated from production water after polymer flooding. Partial sequencing revealed that the closest matches, determined by a BLAST search, corresponded to *Bacillus cereus* strain FM-4 EU794727 (95% similarity, PM-2), *Bacillus* sp. M7-23 EU706321 (95% similarity, PM-3) and *Rhodococcus* strain EF028124.1 (99% similarity, PAM-F1). The GenBank accession numbers of PM-2, PM-3 and PAM-F1 are JN713901, FJ598437 and KC476501.1, respectively (Bao et al., 2010; Sang et al., 2015).

#### 2.1.4. Experimental set-up

Inoculated sludge was cultivated as anaerobic and aerobic activated sludge in an anaerobic baffle reactor (ABR) and a

**Table 1** The ingredients of the wastewater.

Ingredients	Concentrations $(mg \cdot L^{-1})$	Ingredients	Concentrations $(\text{mg} \cdot \text{L}^{-1})$
HPAM	500	NaCl	10
$C_6H_{12}O_6$	80	CoCl <sub>2</sub> ·6H <sub>2</sub> O	5
NH <sub>4</sub> Cl	1000	MnCl <sub>2</sub> ·4H <sub>2</sub> O	5
$NaH_2PO_4$	360	AlCl <sub>3</sub>	2.5
$K_2HPO_4$	3000	$(NH_4)_6Mo_7O_{24}$	15
$NaHCO_3$	405	$H_3BO_3$	5
CaCl <sub>2</sub>	50	NiCl <sub>2</sub> ·6H <sub>2</sub> O	5
$MgCl_2 \cdot 6H_2O$	100	CuCl <sub>2</sub> ·5H <sub>2</sub> O	5
FeCl <sub>2</sub>	25	ZnCl <sub>2</sub>	5

### Download English Version:

# https://daneshyari.com/en/article/7070568

Download Persian Version:

https://daneshyari.com/article/7070568

<u>Daneshyari.com</u>