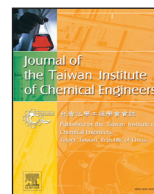




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Synthesis of mechanically robust epoxy cross-linked silica aerogel membranes for CO₂ capture

Yi-Feng Lin^{a,*}, Yen-Ju Lin^a, Chang-Chun Lee^b, Kun-Yi Andrew Lin^{c,**}, Tsair-Wang Chung^a, Kuo-Lun Tung^d

^a Department of Chemical Engineering and R&D Center for Membrane Technology, Chung Yuan Christian University, Chungli, Taoyuan 320, Taiwan, ROC

^b Department of Power Mechanical Engineering, National Tsing Hua University, Hsinchu 30013, Taiwan, ROC

^c Department of Environmental Engineering, National Chung Hsing University, Taichung 250, Taiwan, ROC

^d Department of Chemical Engineering, National Taiwan University, Taipei 106, Taiwan, ROC

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ABSTRACT

Mesoporous SiO₂ aerogels are multi-functional porous materials with diverse applications. However, the fragile characteristics and low mechanical strength of mesoporous SiO₂ aerogels limits their development for use in industrial applications. The mechanical strength of SiO₂ aerogels can be greatly enhanced by the addition of tri-epoxy cross-linkers. Tri-epoxy cross-linked SiO₂ aerogels with tri-epoxy concentrations of 7.5%, 15%, 30% and 45% were coated on macroporous Al₂O₃ membrane supports. The surface of the tri-epoxy cross-linked SiO₂ aerogel membranes became superhydrophobic after fluoroalkylsilane (FAS) modification. Compared to native SiO₂ aerogel membranes in the absence of tri-epoxy cross-linker, the pore size of the tri-epoxy cross-linked SiO₂ aerogel membranes with tri-epoxy concentrations of 15% and 30% increased to approximately 6 nm, resulting in an increase in the CO₂ absorption flux. The highest CO₂ absorption flux of 1.4 mmol/m²s was reached for the tri-epoxy cross-linked SiO₂ aerogel membrane with a tri-epoxy concentration of 15%. The durability of the as-prepared SiO₂ aerogel membranes for CO₂ absorption is also demonstrated in this work. Consequently, the mechanically robust tri-epoxy cross-linked SiO₂ aerogel membranes have potential applications in membrane contactor systems for CO₂ capture.

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1. Introduction

SiO₂ aerogels [1] have attracted great research interests due to their high porosity (at least 90%), large specific surface area (400–1200 m²/g), large pore volume and mesoporous structures (pore size between 2 and 50 nm). SiO₂ aerogels have been successfully prepared via sol–gel processes using a variety of precursors, such as tetraethyl orthosilicate (TEOS) [2], tetramethyl orthosilicate (TMOS) [3], and methyltrimethoxysilane (MTMS) [4]. Diverse applications of SiO₂ aerogels, such as thermal insulation [5], drug release [6], drug delivery [7], waveguides [8], catalysis [9], catalytic supports [10] and adsorption [11], have been studied in the previous literature. The SiO₂ aerogels [12] were also added in the chitosan (CS) polymers to form mixed-matrix membranes (MMMs), just like the zeolite/CS MMMs [13,14], for the ethanol/water separations. However, the SiO₂ aerogels suffer from low mechanical strength and fragile structural features, which

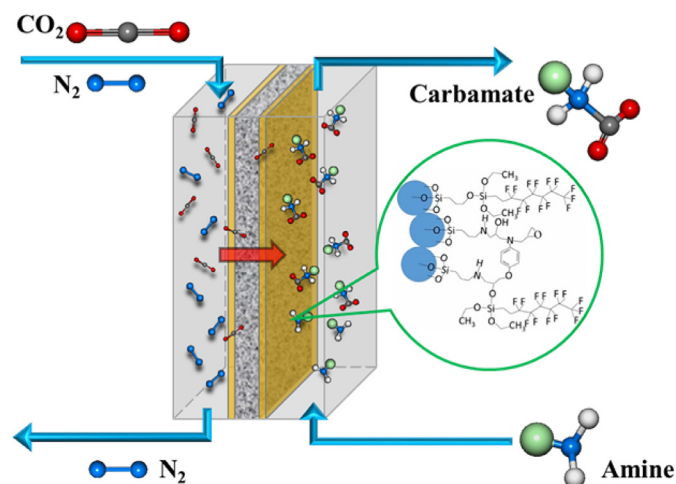
limits their applications. To improve the mechanical strength and fragile structural features of SiO₂ aerogels, Meador et al. [15,16] successfully synthesized mechanically strong, lightweight SiO₂ aerogels via cross-linking of amine-modified SiO₂ aerogels with epoxides. The modulus of the epoxy cross-linked SiO₂ aerogels (10–100 MPa) [15] is 2 orders of magnitude higher than that of native SiO₂ aerogels (0.5–1.0 MPa) [15]. Consequently, the mechanical strength of SiO₂ aerogels can be effectively increased by the addition of epoxy cross-linkers.

In our previous study, for the first time, highly porous SiO₂ aerogels modified with hydrophobic fluorocarbon functional groups (–CF₃) by fluoroalkylsilane (FAS) were successfully coated onto a macroporous Al₂O₃ membrane [17–19]. The FAS-modified SiO₂ aerogel membranes were further applied in a membrane contactor for CO₂ absorption. The hydrophobic SiO₂ aerogel membrane served as an interface between an aqueous amine solution and a CO₂/N₂ gas mixture. CO₂ gases passed through the pores of the hydrophobic SiO₂ aerogel membranes and were absorbed by the aqueous amine solution. The hydrophobic SiO₂ aerogel membrane prevented the membrane from being wet by the aqueous amine solution in order to increase the CO₂ absorption flux. The CO₂

* Corresponding author.

** Corresponding author.

E-mail addresses: yflin@cycu.edu.tw (Y.-F. Lin), linky@nchu.edu.tw (K.-Y.A. Lin).



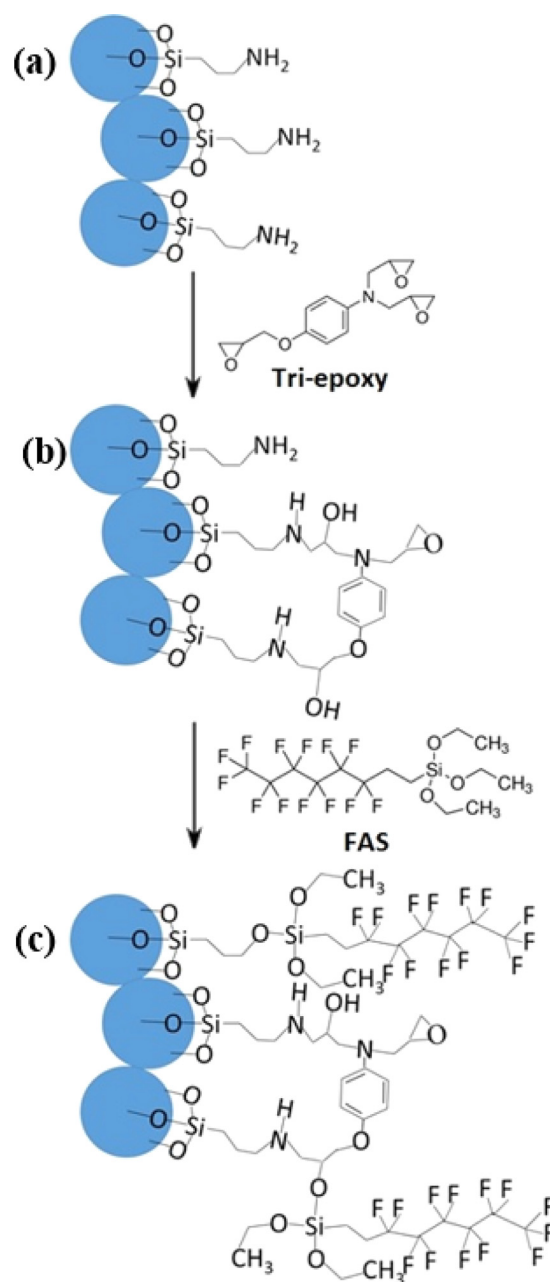
Scheme 1. Scheme of CO₂ absorption in a membrane contactor using the FAS-modified tri-epoxy cross-linked SiO₂ aerogel membrane.

absorption flux of the FAS-modified SiO₂ aerogel membranes reached a stable value of approximately 1.1 mmol/m²s. The durability of the FAS-modified SiO₂ aerogel membranes was also demonstrated by their continuous operation for at least 4 days. To avoid the FAS hydrophobic post-modification, water-proof SiO₂ aerogel membranes were directly coated onto the macroporous Al₂O₃ membranes using hydrophobic MTMS precursors [20,21]. The MTMS-derived SiO₂ aerogel membranes were also utilized in membrane contactor for CO₂ absorption. The CO₂ absorption flux remained at a stable value of approximately 1.2 mmol/m²s with continuous operation for 4 days. Even though the utilization of SiO₂ aerogel membranes in membrane contactors for CO₂ absorption was well established, the mechanical strength of the SiO₂ aerogel membranes was not investigated in the previous literatures.

In this work, the addition of tri-epoxy cross-linkers to the SiO₂ aerogels successfully improved the mechanical strength of the SiO₂ aerogel membranes to overcome the fragility of the aerogels. The tri-epoxy cross-linked SiO₂ aerogel membranes were further used for CO₂ absorption in membrane contactor systems, as shown in Scheme 1. The effects on the CO₂ absorption flux and durability of the tri-epoxy cross-linked SiO₂ aerogel membranes with different tri-epoxy concentrations (7.5%, 15%, 30% and 45%) were also investigated in this study. This work indicates that the tri-epoxy cross-linked SiO₂ aerogel membrane with a tri-epoxy cross-linker concentration of 15% has the highest CO₂ absorption flux of approximately 1.4 mmol/m²s and shows durability over continuous CO₂ absorption for at least 4000 min. The CO₂ absorption flux of approximately 1.4 mmol/m²s using the tri-epoxy cross-linked SiO₂ aerogel membranes is higher than that using TEOS-derived (1.1 mmol/m²s) and MTMS-derived (1.2 mmol/m²s) SiO₂ aerogel membranes. Consequently, the mechanically robust tri-epoxy cross-linked SiO₂ aerogel membranes have potential applications in membrane contactor systems for CO₂ capture.

2. Experimental

For the preparation of silica aerogel membranes via the sol-gel method, a mixture of TEOS (99.7%, SHOWA Company, 0.01 mol), ethanol (EtOH, 99.5%, ECHO Chemical Company, 0.03 mol), and 0.14 wt% hydrochloric acid (HCl, 35%, SHOWA Company) was stirred for 90 min. EtOH (0.05 mol) and 0.15 wt% ammonium hydroxide (NH₄OH, 28–30%, Fisher Scientific Company) were added to the above solution and stirred for 30 min. Subsequently, the Al₂O₃ membrane supports (Kinik Company, Taiwan) with a



Scheme 2. Scheme of the synthesis of the FAS-modified tri-epoxy cross-linked SiO₂ aerogels.

diameter, thickness and pore size of approximately 47 mm, 2.4 mm and 1 μm, respectively, were immersed in the sol solution for several hours. After gelation, the as-prepared SiO₂ aerogel membrane was aged in EtOH at room temperature for 2 days, where the ethanol was refreshed each day. Then, the SiO₂ aerogel membrane was immersed in (3-aminopropyl)trimethoxysilane (APTMS)/EtOH solution (25 wt%) for 1 day. THF was added to the resulting solution to remove the residual APTMS and EtOH over 2 days, where the THF was refreshed each day (Scheme 2(a)). Afterward, the APTMS-modified SiO₂ aerogel membranes were immersed in a solution of tri-epoxy cross-linker (*N,N'*-diglycidyl-4-glycidyloxyaniline) at 50 °C for 24 h (Scheme 2(b)). The resulting membranes were further immersed in THF for 24 h to remove the residual tri-epoxy cross-linker. The tri-epoxy cross-linked SiO₂ aerogel membrane was then transferred into *n*-hexane to remove the THF before the subsequent surface modification. The

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